

# Benzoic acid, 4-[(4-chlorophenyl)sulfonyl]-

<b>Other names:</b>	4-Chloro-4'-carboxydiphenyl sulfone
<b>Inchi:</b>	InChI=1S/C13H9ClO4S/c14-10-3-7-12(8-4-10)19(17,18)11-5-1-9(2-6-11)13(15)16/h1-8H
<b>InchiKey:</b>	MIVYLPYNSKNBPW-UHFFFAOYSA-N
<b>Formula:</b>	C13H9ClO4S
<b>SMILES:</b>	O=C(O)c1ccc(S(=O)(=O)c2ccc(Cl)cc2)cc1
<b>Mol. weight [g/mol]:</b>	296.73
<b>CAS:</b>	37940-65-1

## Physical Properties

Property code	Value	Unit	Source
gf	-482.07	kJ/mol	Joback Method
hf	-595.43	kJ/mol	Joback Method
hfus	37.99	kJ/mol	Joback Method
hvap	96.85	kJ/mol	Joback Method
log10ws	-3.58		Crippen Method
logp	2.871		Crippen Method
mcvol	194.280	ml/mol	McGowan Method
pc	4103.88	kPa	Joback Method
tb	791.42	K	Joback Method
tc	1017.05	K	Joback Method
tf	493.38	K	Joback Method
vc	0.748	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	491.77	J/mol×K	791.42	Joback Method
cpg	501.22	J/mol×K	829.02	Joback Method
cpg	509.67	J/mol×K	866.63	Joback Method
cpg	517.16	J/mol×K	904.23	Joback Method
cpg	523.71	J/mol×K	941.84	Joback Method
cpg	529.36	J/mol×K	979.44	Joback Method
cpg	534.13	J/mol×K	1017.05	Joback Method

# Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C37940651&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C37940651&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvpap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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