

# 1,2,4,5,7,8-hexaoxacyclononane, 3,3,6,6-tetramethyl, 9,9-diethyl

Inchi:	InChI=1S/C11H22O6/c1-7-11(8-2)16-14-9(3,4)12-13-10(5,6)15-17-11/h7-8H2,1-6H3
InchiKey:	DEAVANREEHNQGL-UHFFFAOYSA-N
Formula:	C11H22O6
SMILES:	CCC1(CC)OOC(C)(C)OOC(C)(C)OO1
Mol. weight [g/mol]:	250.29

## Physical Properties

Property code	Value	Unit	Source
gf	-518.72	kJ/mol	Joback Method
hf	-1021.49	kJ/mol	Joback Method
hfus	40.90	kJ/mol	Joback Method
hvap	64.01	kJ/mol	Joback Method
log10ws	-3.64		Crippen Method
logp	2.833		Crippen Method
mcvol	190.210	ml/mol	McGowan Method
pc	2553.34	kPa	Joback Method
rinpol	1269.50		NIST Webbook
rinpol	1270.90		NIST Webbook
rinpol	1275.70		NIST Webbook
rinpol	1270.40		NIST Webbook
rinpol	1268.80		NIST Webbook
rinpol	1273.10		NIST Webbook
rinpol	1257.10		NIST Webbook
tb	636.52	K	Joback Method
tc	867.83	K	Joback Method
tf	433.19	K	Joback Method
vc	0.678	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	551.86	J/molxK	636.52	Joback Method
cpg	570.88	J/molxK	675.07	Joback Method
cpg	589.16	J/molxK	713.62	Joback Method

cpg	606.99	J/mol×K	752.18	Joback Method
cpg	624.64	J/mol×K	790.73	Joback Method
cpg	642.40	J/mol×K	829.28	Joback Method
cpg	660.55	J/mol×K	867.83	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R419831&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R419831&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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