Diethyl carbonate

Other names:	C2H5OC(O)OC2H5
	CARBONIC ACID,DIETHYL ESTER
	Carbonic acid, diethyl ester
	DEC
	DIATOL
	Diaethylcarbonat
	Diethyl ester of carbonic acid
	Diethylester kyseliny uhlicite
	Diethylkarbonat
	ETHYL CARBONATE
	Ethoxyformic anhydride
	Ethyl carbonate ((EtO)2CO)
	Eufin
	NCI-C60899
	NSC 8849
	UN 2366
Inchi:	InChI=1S/C5H10O3/c1-3-7-5(6)8-4-2/h3-4H2,1-2H3
InchiKey:	OIFBSDVPJOWBCH-UHFFFAOYSA-N
Formula:	C5H10O3
SMILES:	CCOC(=O)OCC
Mol. weight [g/mol]:	118.13
CAS:	105-58-8

Physical Properties

Property code	Value	Unit	Source
chl	-2672.50	kJ/mol	NIST Webbook
chl	-2715.17 ± 0.69	kJ/mol	NIST Webbook
gf	-347.70	kJ/mol	Joback Method
hf	-637.90 ± 0.80	kJ/mol	NIST Webbook
hf	-642.40 ± 2.20	kJ/mol	NIST Webbook
hfl	-681.50 ± 0.80	kJ/mol	NIST Webbook
hfl	-724.17	kJ/mol	NIST Webbook
hfus	12.68	kJ/mol	Joback Method
hvap	81.80	kJ/mol	NIST Webbook
hvap	43.60	kJ/mol	NIST Webbook
hvap	39.10	kJ/mol	NIST Webbook
hvap	43.60 ± 0.15	kJ/mol	NIST Webbook

hvap	43.60	kJ/mol	NIST Webbook
hvap	43.60 ± 0.20	kJ/mol	NIST Webbook
hvap	44.40 ± 0.20	kJ/mol	NIST Webbook
log10ws	-0.84		Crippen Method
logp	1.179		Crippen Method
mcvol	94.620	ml/mol	McGowan Method
nfpaf	%!d(float64=3)		KDB
nfpas	%!d(float64=1)		KDB
pc	3589.94	kPa	Joback Method
rinpol	769.00		NIST Webbook
rinpol	769.00		NIST Webbook
rinpol	765.00		NIST Webbook
rinpol	784.00		NIST Webbook
rinpol	768.00		NIST Webbook
rinpol	765.00		NIST Webbook
rinpol	765.00		NIST Webbook
rinpol	768.00		NIST Webbook
rinpol	747.00		NIST Webbook
rinpol	761.00		NIST Webbook
rinpol	747.00		NIST Webbook
rinpol	747.00		NIST Webbook
rinpol	785.00		NIST Webbook
ripol	1104.00		NIST Webbook
ripol	1083.00		NIST Webbook
ripol	1083.00		NIST Webbook
ripol	1100.00		NIST Webbook
ripol	1101.00		NIST Webbook
ripol	1116.00		NIST Webbook
ripol	1099.00		NIST Webbook
ripol	1103.00		NIST Webbook
ripol	1105.00		NIST Webbook
ripol	1105.00		NIST Webbook
ripol	1102.00		NIST Webbook
ripol	1099.00		NIST Webbook
ripol	1100.00		NIST Webbook
tb	399.60	K	Vapor-Liquid Equilibrium and Mixing Properties of Methanol + Diethyl Carbonate and Vinyl Acetate + Diethyl Carbonate Systems
tb	399.65	K	KDB
tb	399.32	K	Isobaric Phase Equilibria of Diethyl Carbonate with Five Alcohols at 101.3 kPa

tb	399.32	K	VLE of the binary systems (dimethyl carbonate with 2-propanol or 2-butanol) and (diethyl carbonate with methylcyclohexane) at 101.3 kPa
tc	578.10	K	Measurement of critical properties for binary and ternary mixtures containing potential gasoline additive diethyl carbonate (DEC)
tf	230.15	K	KDB
tf	196.40	K	Vapor Pressures and Thermophysical Properties of Dimethyl Carbonate, Diethyl Carbonate, and Dipropyl Carbonate
tf	195.86	K	Efficient determination of crystallisation and melting points at low cooling and heating rates with novel computer controlled equipment
vc	0.357	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	231.33	J/mol×K	591.12	Joback Method
cpg	224.02	J/mol×K	561.35	Joback Method
cpg	216.48	J/mol×K	531.58	Joback Method
cpg	208.73	J/mol×K	501.81	Joback Method
cpg	200.78	J/mol×K	472.05	Joback Method
cpg	192.64	J/mol×K	442.28	Joback Method
cpg	184.32	J/mol×K	412.51	Joback Method
cpl	240.30	J/mol×K	348.15	Excess Molar Enthalpies, Molar Heat Capacities, Densities, Viscosities, and Refractive Indices of Dimethyl Sulfoxide + Esters of Carbonic Acid at 308.15 K and Atmospheric Pressure

cpl	222.30	J/mol×K	303.15	Excess Molar Enthalpies, Molar Heat Capacities, Densities, Viscosities, and Refractive Indices of Dimethyl Sulfoxide + Esters of Carbonic Acid at 308.15 K and Atmospheric Pressure	
cpl	223.40	J/mol×K	308.15	Excess Molar Enthalpies, Molar Heat Capacities, Densities, Viscosities, and Refractive Indices of Dimethyl Sulfoxide + Esters of Carbonic Acid at 308.15 K and Atmospheric Pressure	
cpl	225.30	J/mol×K	313.15	Excess Molar Enthalpies, Molar Heat Capacities, Densities, Viscosities, and Refractive Indices of Dimethyl Sulfoxide + Esters of Carbonic Acid at 308.15 K and Atmospheric Pressure	
cpl	226.60	J/mol×K	318.15	Excess Molar Enthalpies, Molar Heat Capacities, Densities, Viscosities, and Refractive Indices of Dimethyl Sulfoxide + Esters of Carbonic Acid at 308.15 K and Atmospheric Pressure	

cpl	228.40	J/mol×K	323.15	Excess Molar Enthalpies, Molar Heat Capacities, Densities, Viscosities, and Refractive Indices of Dimethyl Sulfoxide + Esters of Carbonic Acid at 308.15 K and Atmospheric Pressure
cpl	230.80	J/mol×K	328.15	Excess Molar Enthalpies, Molar Heat Capacities, Densities, Viscosities, and Refractive Indices of Dimethyl Sulfoxide + Esters of Carbonic Acid at 308.15 K and Atmospheric Pressure
cpl	232.00	J/mol×K	333.15	Excess Molar Enthalpies, Molar Heat Capacities, Densities, Viscosities, and Refractive Indices of Dimethyl Sulfoxide + Esters of Carbonic Acid at 308.15 K and Atmospheric Pressure
cpl	234.60	J/mol×K	338.15	Excess Molar Enthalpies, Molar Heat Capacities, Densities, Viscosities, and Refractive Indices of Dimethyl Sulfoxide + Esters of Carbonic Acid at 308.15 K and Atmospheric Pressure

cpl	237.60	J/mol×K	343.15	Excess Molar Enthalpies, Molar Heat Capacities, Densities, Viscosities, and Refractive Indices of Dimethyl Sulfoxide + Esters of Carbonic Acid at 308.15 K and Atmospheric Pressure
cpl	219.90	J/mol×K	293.15	Excess Molar Enthalpies, Molar Heat Capacities, Densities, Viscosities, and Refractive Indices of Dimethyl Sulfoxide + Esters of Carbonic Acid at 308.15 K and Atmospheric Pressure
cpl	242.50	J/mol×K	353.15	Excess Molar Enthalpies, Molar Heat Capacities, Densities, Viscosities, and Refractive Indices of Dimethyl Sulfoxide + Esters of Carbonic Acid at 308.15 K and Atmospheric Pressure
cpl	245.50	J/mol×K	358.15	Excess Molar Enthalpies, Molar Heat Capacities, Densities, Viscosities, and Refractive Indices of Dimethyl Sulfoxide + Esters of Carbonic Acid at 308.15 K and Atmospheric Pressure

cpl	248.90	J/mol×K	363.15	Excess Molar Enthalpies, Molar Heat Capacities, Densities, Viscosities, and Refractive Indices of Dimethyl Sulfoxide + Esters of Carbonic Acid at 308.15 K and Atmospheric Pressure
cpl	251.30	J/mol×K	368.15	Excess Molar Enthalpies, Molar Heat Capacities, Densities, Viscosities, and Refractive Indices of Dimethyl Sulfoxide + Esters of Carbonic Acid at 308.15 K and Atmospheric Pressure
cpl	254.40	J/mol×K	373.15	Excess Molar Enthalpies, Molar Heat Capacities, Densities, Viscosities, and Refractive Indices of Dimethyl Sulfoxide + Esters of Carbonic Acid at 308.15 K and Atmospheric Pressure
cpl	210.20	J/mol×K	288.15	Densities, Viscosities, Refractive Indices, and Heat Capacities of Poly(ethylene glycol-ran-propylene glycol) + Esters of Carbonic Acid at (293.15 and 313.15) K and at Atmospheric Pressure

cpl	211.49	J/mol×K	293.15	Densities, Viscosities, Refractive Indices, and Heat Capacities of Poly(ethylene glycol-ran-propylene glycol) + Esters of Carbonic Acid at (293.15 and 313.15) K and at Atmospheric Pressure
cpl	212.46	J/mol×K	298.15	Densities, Viscosities, Refractive Indices, and Heat Capacities of Poly(ethylene glycol-ran-propylene glycol) + Esters of Carbonic Acid at (293.15 and 313.15) K and at Atmospheric Pressure
cpl	213.83	J/mol×K	303.15	Densities, Viscosities, Refractive Indices, and Heat Capacities of Poly(ethylene glycol-ran-propylene glycol) + Esters of Carbonic Acid at (293.15 and 313.15) K and at Atmospheric Pressure
cpl	214.89	J/mol×K	308.15	Densities, Viscosities, Refractive Indices, and Heat Capacities of Poly(ethylene glycol-ran-propylene glycol) + Esters of Carbonic Acid at (293.15 and 313.15) K and at Atmospheric Pressure
cpl	215.80	J/mol×K	313.15	Densities, Viscosities, Refractive Indices, and Heat Capacities of Poly(ethylene glycol-ran-propylene glycol) + Esters of Carbonic Acid at (293.15 and 313.15) K and at Atmospheric Pressure

cpl	218.55	J/mol×K	318.15	Densities, Viscosities, Refractive Indices, and Heat Capacities of Poly(ethylene glycol-ran-propylene glycol) + Esters of Carbonic Acid at (293.15 and 313.15) K and at Atmospheric Pressure
cpl	220.32	J/mol×K	323.15	Densities, Viscosities, Refractive Indices, and Heat Capacities of Poly(ethylene glycol-ran-propylene glycol) + Esters of Carbonic Acid at (293.15 and 313.15) K and at Atmospheric Pressure
cpl	210.90	J/mol×K	294.20	NIST Webbook
cpl	210.90	J/mol×K	294.70	NIST Webbook
cpl	182.00	J/mol×K	298.00	NIST Webbook
cpl	221.20	J/mol×K	298.15	Excess Molar Enthalpies, Molar Heat Capacities, Densities, Viscosities, and Refractive Indices of Dimethyl Sulfoxide + Esters of Carbonic Acid at 308.15 K and Atmospheric Pressure
dvisc	0.0006950	Pa×s	303.15	Densities and Viscosities of Diethyl Carbonate + Toluene, + Methanol, and + 2-Propanol from (293.15 to 363.15) K
dvisc	0.0006100	Pa×s	313.15	Densities and Viscosities of Diethyl Carbonate + Toluene, + Methanol, and + 2-Propanol from (293.15 to 363.15) K

dvisc	0.0005410	Pa×s	323.15	Densities and Viscosities of Diethyl Carbonate + Toluene, + Methanol, and + 2-Propanol from (293.15 to 363.15) K	
dvisc	0.0004830	Pa×s	333.15	Densities and Viscosities of Diethyl Carbonate + Toluene, + Methanol, and + 2-Propanol from (293.15 to 363.15) K	
dvisc	0.0004350	Pa×s	343.15	Densities and Viscosities of Diethyl Carbonate + Toluene, + Methanol, and + 2-Propanol from (293.15 to 363.15) K	
dvisc	0.0003930	Pa×s	353.15	Densities and Viscosities of Diethyl Carbonate + Toluene, + Methanol, and + 2-Propanol from (293.15 to 363.15) K	
dvisc	0.0003580	Pa×s	363.15	Densities and Viscosities of Diethyl Carbonate + Toluene, + Methanol, and + 2-Propanol from (293.15 to 363.15) K	
dvisc	0.0008000	Pa×s	293.15	Density and Viscosity of Binary Mixtures of Diethyl Carbonate with Alcohols at (293.15 to 363.15) K and Predictive Results by UNIFAC-VISCO Group Contribution Method	

dvisc	0.0007460	Pa×s	298.15	Density and Viscosity of Binary Mixtures of Diethyl Carbonate with Alcohols at (293.15 to 363.15) K and Predictive Results by UNIFAC-VISCO Group Contribution Method
dvisc	0.0006950	Pa×s	303.15	Density and Viscosity of Binary Mixtures of Diethyl Carbonate with Alcohols at (293.15 to 363.15) K and Predictive Results by UNIFAC-VISCO Group Contribution Method
dvisc	0.0006100	Pa×s	313.15	Density and Viscosity of Binary Mixtures of Diethyl Carbonate with Alcohols at (293.15 to 363.15) K and Predictive Results by UNIFAC-VISCO Group Contribution Method
dvisc	0.0005410	Pa×s	323.15	Density and Viscosity of Binary Mixtures of Diethyl Carbonate with Alcohols at (293.15 to 363.15) K and Predictive Results by UNIFAC-VISCO Group Contribution Method

dvisc	0.0004830	Pa×s	333.15	Density and Viscosity of Binary Mixtures of Diethyl Carbonate with Alcohols at (293.15 to 363.15) K and Predictive Results by UNIFAC-VISCO Group Contribution Method
dvisc	0.0004350	Pa×s	343.15	Density and Viscosity of Binary Mixtures of Diethyl Carbonate with Alcohols at (293.15 to 363.15) K and Predictive Results by UNIFAC-VISCO Group Contribution Method
dvisc	0.0003930	Pa×s	353.15	Density and Viscosity of Binary Mixtures of Diethyl Carbonate with Alcohols at (293.15 to 363.15) K and Predictive Results by UNIFAC-VISCO Group Contribution Method
dvisc	0.0003580	Pa×s	363.15	Density and Viscosity of Binary Mixtures of Diethyl Carbonate with Alcohols at (293.15 to 363.15) K and Predictive Results by UNIFAC-VISCO Group Contribution Method
dvisc	0.0008640	Pa×s	288.15	Densities and Viscosities of Four Binary Diethyl Carbonate + 1-Alcohol Systems from (288.15 to 313.15) K

dvisc	0.0008030	Pa×s	293.15	Densities and Viscosities of Four Binary Diethyl Carbonate + 1-Alcohol Systems from (288.15 to 313.15) K	
dvisc	0.0007570	Pa×s	298.15	Densities and Viscosities of Four Binary Diethyl Carbonate + 1-Alcohol Systems from (288.15 to 313.15) K	
dvisc	0.0007000	Pa×s	303.15	Densities and Viscosities of Four Binary Diethyl Carbonate + 1-Alcohol Systems from (288.15 to 313.15) K	
dvisc	0.0006160	Pa×s	313.15	Densities and Viscosities of Four Binary Diethyl Carbonate + 1-Alcohol Systems from (288.15 to 313.15) K	
dvisc	0.0007460	Pa×s	298.15	Densities and Viscosities of Diethyl Carbonate + Toluene, + Methanol, and + 2-Propanol from (293.15 to 363.15) K	
dvisc	0.0008000	Pa×s	293.15	Densities and Viscosities of Diethyl Carbonate + Toluene, + Methanol, and + 2-Propanol from (293.15 to 363.15) K	
dvisc	0.0008010	Pa×s	293.15	Densities and Viscosities of Diethyl Carbonate + Toluene, + Methanol, and + 2-Propanol from (293.15 to 363.15) K	

dvisc	0.0006180	Pa×s	313.15	Density, Viscosity, and Speed of Sound of Dialkyl Carbonates with Cyclopentane and Methyl Cyclohexane at Several Temperatures	
dvisc	0.0006580	Pa×s	308.15	Density, Viscosity, and Speed of Sound of Dialkyl Carbonates with Cyclopentane and Methyl Cyclohexane at Several Temperatures	
dvisc	0.0007020	Pa×s	303.15	Density, Viscosity, and Speed of Sound of Dialkyl Carbonates with Cyclopentane and Methyl Cyclohexane at Several Temperatures	
dvisc	0.0008090	Pa×s	293.15	Density, Viscosity, and Speed of Sound of Dialkyl Carbonates with Cyclopentane and Methyl Cyclohexane at Several Temperatures	
dvisc	0.0007500	Pa×s	298.15	Density, Viscosity, and Speed of Sound of Dialkyl Carbonates with Cyclopentane and Methyl Cyclohexane at Several Temperatures	
hfust	9.24	kJ/mol	198.20	NIST Webbook	
hvapt	44.30	kJ/mol	331.00	NIST Webbook	
hvapt	39.10	kJ/mol	338.00	NIST Webbook	
hvapt	40.90	kJ/mol	354.00	NIST Webbook	
hvapt	39.70	kJ/mol	377.50	NIST Webbook	
hvapt	42.30	kJ/mol	371.00	NIST Webbook	
pvap	1.45	kPa	298.10	Vapour pressure and enthalpy of vaporization of aliphatic dialkyl carbonates	

pvap	2.45	kPa	306.80	Vapour pressure and enthalpy of vaporization of aliphatic dialkyl carbonates	
pvap	2.69	kPa	307.80	Vapour pressure and enthalpy of vaporization of aliphatic dialkyl carbonates	
pvap	2.64	kPa	308.20	Vapour pressure and enthalpy of vaporization of aliphatic dialkyl carbonates	
pvap	2.69	kPa	308.20	Vapour pressure and enthalpy of vaporization of aliphatic dialkyl carbonates	
pvap	2.67	kPa	308.20	Vapour pressure and enthalpy of vaporization of aliphatic dialkyl carbonates	
pvap	3.07	kPa	310.90	Vapour pressure and enthalpy of vaporization of aliphatic dialkyl carbonates	
pvap	3.65	kPa	313.30	Vapour pressure and enthalpy of vaporization of aliphatic dialkyl carbonates	
pvap	3.96	kPa	315.00	Vapour pressure and enthalpy of vaporization of aliphatic dialkyl carbonates	
pvap	3.95	kPa	315.20	Vapour pressure and enthalpy of vaporization of aliphatic dialkyl carbonates	
pvap	0.40	kPa	277.96	Low pressure methane solubility in lithium-ion batteries based solvents and electrolytes as a function of temperature. Measurement and prediction	

pvap	0.52	kPa	282.06	Low pressure methane solubility in lithium-ion batteries based solvents and electrolytes as a function of temperature. Measurement and prediction	
pvap	0.67	kPa	286.15	Low pressure methane solubility in lithium-ion batteries based solvents and electrolytes as a function of temperature. Measurement and prediction	
pvap	0.80	kPa	288.15	Low pressure methane solubility in lithium-ion batteries based solvents and electrolytes as a function of temperature. Measurement and prediction	
pvap	1.03	kPa	293.15	Low pressure methane solubility in lithium-ion batteries based solvents and electrolytes as a function of temperature. Measurement and prediction	
pvap	1.41	kPa	298.15	Low pressure methane solubility in lithium-ion batteries based solvents and electrolytes as a function of temperature. Measurement and prediction	

pvap	1.84	kPa	303.15	Low pressure methane solubility in lithium-ion batteries based solvents and electrolytes as a function of temperature. Measurement and prediction	
pvap	2.35	kPa	306.64	Low pressure methane solubility in lithium-ion batteries based solvents and electrolytes as a function of temperature. Measurement and prediction	
pvap	3.01	kPa	310.74	Low pressure methane solubility in lithium-ion batteries based solvents and electrolytes as a function of temperature. Measurement and prediction	
pvap	4.58	kPa	318.15	Low pressure methane solubility in lithium-ion batteries based solvents and electrolytes as a function of temperature. Measurement and prediction	
pvap	6.17	kPa	323.15	Low pressure methane solubility in lithium-ion batteries based solvents and electrolytes as a function of temperature. Measurement and prediction	

pvap	7.16	kPa	328.15	Low pressure methane solubility in lithium-ion batteries based solvents and electrolytes as a function of temperature. Measurement and prediction	
pvap	9.09	kPa	333.15	Low pressure methane solubility in lithium-ion batteries based solvents and electrolytes as a function of temperature. Measurement and prediction	
pvap	11.56	kPa	338.15	Low pressure methane solubility in lithium-ion batteries based solvents and electrolytes as a function of temperature. Measurement and prediction	
pvap	13.78	kPa	343.15	Low pressure methane solubility in lithium-ion batteries based solvents and electrolytes as a function of temperature. Measurement and prediction	
pvap	16.09	kPa	348.15	Low pressure methane solubility in lithium-ion batteries based solvents and electrolytes as a function of temperature. Measurement and prediction	

pvap	20.83	kPa	353.15	Low pressure methane solubility in lithium-ion batteries based solvents and electrolytes as a function of temperature. Measurement and prediction
pvap	45.10	kPa	373.00	Isothermal (vapour + liquid) equilibrium (VLE) for binary mixtures containing diethyl carbonate, phenyl acetate, diphenyl carbonate, or ethyl acetate
pvap	143.00	kPa	412.50	Isothermal (vapour + liquid) equilibrium (VLE) for binary mixtures containing diethyl carbonate, phenyl acetate, diphenyl carbonate, or ethyl acetate
pvap	361.00	kPa	452.20	Isothermal (vapour + liquid) equilibrium (VLE) for binary mixtures containing diethyl carbonate, phenyl acetate, diphenyl carbonate, or ethyl acetate
pvap	1.49	kPa	298.00	Vapour pressure and enthalpy of vaporization of aliphatic dialkyl carbonates
pvap	2.61	kPa	308.15	Measurement and Correlation of Isothermal Binary Vapor Liquid Equilibrium for Diethyl Carbonate + Isooctane/n-Heptane/Toluene Systems

pvap	3.43	kPa	313.15	Measurement and Correlation of Isothermal Binary Vapor Liquid Equilibrium for Diethyl Carbonate + Isooctane/n-Heptane/Toluene Systems
pvap	4.46	kPa	318.15	Measurement and Correlation of Isothermal Binary Vapor Liquid Equilibrium for Diethyl Carbonate + Isooctane/n-Heptane/Toluene Systems
pvap	5.75	kPa	323.15	Measurement and Correlation of Isothermal Binary Vapor Liquid Equilibrium for Diethyl Carbonate + Isooctane/n-Heptane/Toluene Systems
pvap	1.99	kPa	303.15	Measurement and Correlation of Isothermal Binary Vapor Liquid Equilibrium for Diethyl Carbonate + Isooctane/n-Heptane/Toluene Systems
pvap	2.65	kPa	308.15	Measurement and Correlation of Isothermal Binary Vapor Liquid Equilibrium for Diethyl Carbonate + Isooctane/n-Heptane/Toluene Systems
pvap	3.45	kPa	313.15	Measurement and Correlation of Isothermal Binary Vapor Liquid Equilibrium for Diethyl Carbonate + Isooctane/n-Heptane/Toluene Systems

pvap	4.51	kPa	318.15	Measurement and Correlation of Isothermal Binary Vapor Liquid Equilibrium for Diethyl Carbonate + Isooctane/n-Heptane/Toluene Systems
pvap	5.83	kPa	323.15	Measurement and Correlation of Isothermal Binary Vapor Liquid Equilibrium for Diethyl Carbonate + Isooctane/n-Heptane/Toluene Systems
pvap	101.30	kPa	399.60	Vapor-Liquid Equilibrium and Mixing Properties of Methanol + Diethyl Carbonate and Vinyl Acetate + Diethyl Carbonate Systems
pvap	22.56	kPa	354.68	Phase Equilibria Involved in the Extractive Distillation of Cyclohexane + Cyclohexene Using Diethyl Carbonate as an Entrainer
pvap	25.03	kPa	357.44	Phase Equilibria Involved in the Extractive Distillation of Cyclohexane + Cyclohexene Using Diethyl Carbonate as an Entrainer
pvap	27.54	kPa	359.95	Phase Equilibria Involved in the Extractive Distillation of Cyclohexane + Cyclohexene Using Diethyl Carbonate as an Entrainer

pvap	29.99	kPa	362.26	Phase Equilibria Involved in the Extractive Distillation of Cyclohexane + Cyclohexene Using Diethyl Carbonate as an Entrainer
pvap	32.51	kPa	364.46	Phase Equilibria Involved in the Extractive Distillation of Cyclohexane + Cyclohexene Using Diethyl Carbonate as an Entrainer
pvap	35.05	kPa	366.57	Phase Equilibria Involved in the Extractive Distillation of Cyclohexane + Cyclohexene Using Diethyl Carbonate as an Entrainer
pvap	37.52	kPa	368.45	Phase Equilibria Involved in the Extractive Distillation of Cyclohexane + Cyclohexene Using Diethyl Carbonate as an Entrainer
pvap	40.03	kPa	370.28	Phase Equilibria Involved in the Extractive Distillation of Cyclohexane + Cyclohexene Using Diethyl Carbonate as an Entrainer
pvap	42.49	kPa	372.04	Phase Equilibria Involved in the Extractive Distillation of Cyclohexane + Cyclohexene Using Diethyl Carbonate as an Entrainer
pvap	45.05	kPa	373.73	Phase Equilibria Involved in the Extractive Distillation of Cyclohexane + Cyclohexene Using Diethyl Carbonate as an Entrainer

pvap	47.47	kPa	375.26	Phase Equilibria Involved in the Extractive Distillation of Cyclohexane + Cyclohexene Using Diethyl Carbonate as an Entrainer
pvap	50.04	kPa	376.81	Phase Equilibria Involved in the Extractive Distillation of Cyclohexane + Cyclohexene Using Diethyl Carbonate as an Entrainer
pvap	52.46	kPa	378.22	Phase Equilibria Involved in the Extractive Distillation of Cyclohexane + Cyclohexene Using Diethyl Carbonate as an Entrainer
pvap	55.03	kPa	379.67	Phase Equilibria Involved in the Extractive Distillation of Cyclohexane + Cyclohexene Using Diethyl Carbonate as an Entrainer
pvap	57.56	kPa	381.03	Phase Equilibria Involved in the Extractive Distillation of Cyclohexane + Cyclohexene Using Diethyl Carbonate as an Entrainer
pvap	59.95	kPa	382.32	Phase Equilibria Involved in the Extractive Distillation of Cyclohexane + Cyclohexene Using Diethyl Carbonate as an Entrainer
pvap	62.52	kPa	383.60	Phase Equilibria Involved in the Extractive Distillation of Cyclohexane + Cyclohexene Using Diethyl Carbonate as an Entrainer

pvap	64.97	kPa	384.81	Phase Equilibria Involved in the Extractive Distillation of Cyclohexane + Cyclohexene Using Diethyl Carbonate as an Entrainer	
pvap	67.44	kPa	385.98	Phase Equilibria Involved in the Extractive Distillation of Cyclohexane + Cyclohexene Using Diethyl Carbonate as an Entrainer	
pvap	70.02	kPa	387.13	Phase Equilibria Involved in the Extractive Distillation of Cyclohexane + Cyclohexene Using Diethyl Carbonate as an Entrainer	
pvap	72.51	kPa	388.28	Phase Equilibria Involved in the Extractive Distillation of Cyclohexane + Cyclohexene Using Diethyl Carbonate as an Entrainer	
pvap	74.96	kPa	389.32	Phase Equilibria Involved in the Extractive Distillation of Cyclohexane + Cyclohexene Using Diethyl Carbonate as an Entrainer	
pvap	77.43	kPa	390.35	Phase Equilibria Involved in the Extractive Distillation of Cyclohexane + Cyclohexene Using Diethyl Carbonate as an Entrainer	
pvap	79.97	kPa	391.39	Phase Equilibria Involved in the Extractive Distillation of Cyclohexane + Cyclohexene Using Diethyl Carbonate as an Entrainer	

pvap	82.52	kPa	392.45	Phase Equilibria Involved in the Extractive Distillation of Cyclohexane + Cyclohexene Using Diethyl Carbonate as an Entrainer
pvap	84.84	kPa	393.34	Phase Equilibria Involved in the Extractive Distillation of Cyclohexane + Cyclohexene Using Diethyl Carbonate as an Entrainer
pvap	87.52	kPa	394.36	Phase Equilibria Involved in the Extractive Distillation of Cyclohexane + Cyclohexene Using Diethyl Carbonate as an Entrainer
pvap	89.81	kPa	395.23	Phase Equilibria Involved in the Extractive Distillation of Cyclohexane + Cyclohexene Using Diethyl Carbonate as an Entrainer
pvap	92.53	kPa	396.21	Phase Equilibria Involved in the Extractive Distillation of Cyclohexane + Cyclohexene Using Diethyl Carbonate as an Entrainer
pvap	94.89	kPa	397.03	Phase Equilibria Involved in the Extractive Distillation of Cyclohexane + Cyclohexene Using Diethyl Carbonate as an Entrainer
pvap	97.54	kPa	397.96	Phase Equilibria Involved in the Extractive Distillation of Cyclohexane + Cyclohexene Using Diethyl Carbonate as an Entrainer

pvap	100.03	kPa	398.81	Phase Equilibria Involved in the Extractive Distillation of Cyclohexane + Cyclohexene Using Diethyl Carbonate as an Entrainer
pvap	101.38	kPa	399.27	Phase Equilibria Involved in the Extractive Distillation of Cyclohexane + Cyclohexene Using Diethyl Carbonate as an Entrainer
pvap	13.80	kPa	343.77	Density, Viscosity, and Vapor Pressure for Binary Mixtures of Tricyclo [5.2.1.02.6] Decane and Diethyl Carbonate
pvap	19.30	kPa	351.67	Density, Viscosity, and Vapor Pressure for Binary Mixtures of Tricyclo [5.2.1.02.6] Decane and Diethyl Carbonate
pvap	22.70	kPa	355.63	Density, Viscosity, and Vapor Pressure for Binary Mixtures of Tricyclo [5.2.1.02.6] Decane and Diethyl Carbonate
pvap	26.00	kPa	358.29	Density, Viscosity, and Vapor Pressure for Binary Mixtures of Tricyclo [5.2.1.02.6] Decane and Diethyl Carbonate

pvap	31.30	kPa	363.84	Density, Viscosity, and Vapor Pressure for Binary Mixtures of Tricyclo [5.2.1.02.6] Decane and Diethyl Carbonate
pvap	35.20	kPa	366.98	Density, Viscosity, and Vapor Pressure for Binary Mixtures of Tricyclo [5.2.1.02.6] Decane and Diethyl Carbonate
pvap	39.10	kPa	369.75	Density, Viscosity, and Vapor Pressure for Binary Mixtures of Tricyclo [5.2.1.02.6] Decane and Diethyl Carbonate
pvap	42.80	kPa	372.28	Density, Viscosity, and Vapor Pressure for Binary Mixtures of Tricyclo [5.2.1.02.6] Decane and Diethyl Carbonate
pvap	46.50	kPa	374.61	Density, Viscosity, and Vapor Pressure for Binary Mixtures of Tricyclo [5.2.1.02.6] Decane and Diethyl Carbonate
pvap	50.50	kPa	377.05	Density, Viscosity, and Vapor Pressure for Binary Mixtures of Tricyclo [5.2.1.02.6] Decane and Diethyl Carbonate

pvap	55.00	kPa	379.59	Density, Viscosity, and Vapor Pressure for Binary Mixtures of Tricyclo [5.2.1.02.6] Decane and Diethyl Carbonate
pvap	58.80	kPa	381.49	Density, Viscosity, and Vapor Pressure for Binary Mixtures of Tricyclo [5.2.1.02.6] Decane and Diethyl Carbonate
pvap	63.10	kPa	383.37	Density, Viscosity, and Vapor Pressure for Binary Mixtures of Tricyclo [5.2.1.02.6] Decane and Diethyl Carbonate
pvap	63.50	kPa	383.57	Density, Viscosity, and Vapor Pressure for Binary Mixtures of Tricyclo [5.2.1.02.6] Decane and Diethyl Carbonate
pvap	68.50	kPa	386.18	Density, Viscosity, and Vapor Pressure for Binary Mixtures of Tricyclo [5.2.1.02.6] Decane and Diethyl Carbonate
pvap	72.80	kPa	387.61	Density, Viscosity, and Vapor Pressure for Binary Mixtures of Tricyclo [5.2.1.02.6] Decane and Diethyl Carbonate

pvap	78.50	kPa	390.35	Density, Viscosity, and Vapor Pressure for Binary Mixtures of Tricyclo [5.2.1.02.6] Decane and Diethyl Carbonate
pvap	84.00	kPa	392.39	Density, Viscosity, and Vapor Pressure for Binary Mixtures of Tricyclo [5.2.1.02.6] Decane and Diethyl Carbonate
pvap	88.20	kPa	393.93	Density, Viscosity, and Vapor Pressure for Binary Mixtures of Tricyclo [5.2.1.02.6] Decane and Diethyl Carbonate
pvap	93.40	kPa	395.72	Density, Viscosity, and Vapor Pressure for Binary Mixtures of Tricyclo [5.2.1.02.6] Decane and Diethyl Carbonate
pvap	101.50	kPa	398.28	Density, Viscosity, and Vapor Pressure for Binary Mixtures of Tricyclo [5.2.1.02.6] Decane and Diethyl Carbonate
pvap	1.45	kPa	296.80	Vapour pressure and enthalpy of vaporization of aliphatic dialkyl carbonates
pvap	1.06	kPa	293.30	Vapour pressure and enthalpy of vaporization of aliphatic dialkyl carbonates

pvap	1.08	kPa	293.20	Vapour pressure and enthalpy of vaporization of aliphatic dialkyl carbonates
pvap	1.04	kPa	293.10	Vapour pressure and enthalpy of vaporization of aliphatic dialkyl carbonates
pvap	1.03	kPa	293.10	Vapour pressure and enthalpy of vaporization of aliphatic dialkyl carbonates
pvap	1.04	kPa	293.10	Vapour pressure and enthalpy of vaporization of aliphatic dialkyl carbonates
pvap	1.99	kPa	303.30	Vapour pressure and enthalpy of vaporization of aliphatic dialkyl carbonates
pvap	1.05	kPa	293.00	Vapour pressure and enthalpy of vaporization of aliphatic dialkyl carbonates
pvap	1.02	kPa	291.80	Vapour pressure and enthalpy of vaporization of aliphatic dialkyl carbonates
pvap	0.78	kPa	288.40	Vapour pressure and enthalpy of vaporization of aliphatic dialkyl carbonates
pvap	0.83	kPa	288.30	Vapour pressure and enthalpy of vaporization of aliphatic dialkyl carbonates
pvap	0.79	kPa	288.20	Vapour pressure and enthalpy of vaporization of aliphatic dialkyl carbonates
pvap	0.78	kPa	286.80	Vapour pressure and enthalpy of vaporization of aliphatic dialkyl carbonates
pvap	0.63	kPa	285.00	Vapour pressure and enthalpy of vaporization of aliphatic dialkyl carbonates

pvap	0.56	kPa	283.30	Vapour pressure and enthalpy of vaporization of aliphatic dialkyl carbonates	
pvap	0.58	kPa	283.10	Vapour pressure and enthalpy of vaporization of aliphatic dialkyl carbonates	
pvap	0.56	kPa	282.40	Vapour pressure and enthalpy of vaporization of aliphatic dialkyl carbonates	
pvap	0.41	kPa	278.20	Vapour pressure and enthalpy of vaporization of aliphatic dialkyl carbonates	
pvap	0.41	kPa	278.10	Vapour pressure and enthalpy of vaporization of aliphatic dialkyl carbonates	
pvap	0.38	kPa	277.10	Vapour pressure and enthalpy of vaporization of aliphatic dialkyl carbonates	
pvap	0.36	kPa	276.50	Vapour pressure and enthalpy of vaporization of aliphatic dialkyl carbonates	
pvap	0.35	kPa	276.20	Vapour pressure and enthalpy of vaporization of aliphatic dialkyl carbonates	
pvap	0.33	kPa	275.00	Vapour pressure and enthalpy of vaporization of aliphatic dialkyl carbonates	
pvap	0.31	kPa	274.00	Vapour pressure and enthalpy of vaporization of aliphatic dialkyl carbonates	
pvap	0.30	kPa	274.00	Vapour pressure and enthalpy of vaporization of aliphatic dialkyl carbonates	
pvap	0.29	kPa	273.30	Vapour pressure and enthalpy of vaporization of aliphatic dialkyl carbonates	

pvap	0.29	kPa	273.10	Vapour pressure and enthalpy of vaporization of aliphatic dialkyl carbonates	
pvap	30.23	kPa	363.15	Isothermal vapor liquid and vapor liquid liquid equilibrium for the ternary system ethanol + water + diethyl carbonate and constituent binary systems at different temperatures	
pvap	20.74	kPa	353.15	Isothermal vapor liquid and vapor liquid liquid equilibrium for the ternary system ethanol + water + diethyl carbonate and constituent binary systems at different temperatures	
pvap	2.06	kPa	303.10	Vapour pressure and enthalpy of vaporization of aliphatic dialkyl carbonates	
pvap	1.52	kPa	298.10	Vapour pressure and enthalpy of vaporization of aliphatic dialkyl carbonates	
pvap	2.01	kPa	303.00	Vapour pressure and enthalpy of vaporization of aliphatic dialkyl carbonates	
pvap	1.91	kPa	301.80	Vapour pressure and enthalpy of vaporization of aliphatic dialkyl carbonates	
pvap	1.54	kPa	298.20	Vapour pressure and enthalpy of vaporization of aliphatic dialkyl carbonates	

pvap	1.95	kPa	303.15	Measurement and Correlation of Isothermal Binary Vapor Liquid Equilibrium for Diethyl Carbonate + Isooctane/n-Heptane/Toluene Systems	
pvap	2.02	kPa	303.10	Vapour pressure and enthalpy of vaporization of aliphatic dialkyl carbonates	
rfi	1.38450		293.15	Densities and derived thermodynamic properties of binary mixtures of diethylcarbonate, acetophenone, and 1-hexanol at T = (293.15 to 323.15) K for the liquid region and at ambient pressure	
rfi	1.38300		293.15	Vapor liquid equilibria of carbon dioxide with isopropyl acetate, diethyl carbonate and ethyl butyrate at elevated pressures	
rfi	1.38450		293.15	Densities, Viscosities, Speeds of Sound, and Refractive Indices for Binary Mixtures of Diethylcarbonate, Acetophenone, and 1-Hexanol at (293.15, 303.15, 313.15, and 323.15) K for the Liquid Region and at Ambient Pressure	

rfi	1.38300		298.15	Viscosities, Ultrasonic Velocities at (288.15 and 298.15) K, and Refractive Indices at (298.15) K of Binary Mixtures of 2,4,6-Trimethyl-1,3,5-trioxane with Dimethyl Carbonate, Diethyl Carbonate, and Propylene Carbonate
rfi	1.38280		298.15	Properties of ionic liquid HMIMPF6 with carbonates, ketones and alkyl acetates
rfi	1.38210		298.15	Isobaric Vapor-Liquid Equilibria of Binary Mixtures of Diethyl Carbonate with Methyl Acetate, n-Propyl Acetate, or Amyl Acetate at 100.17 kPa
rhol	957.98	kg/m3	308.15	Permittivity and density of binary systems of {dimethyl or diethyl carbonate} + n-dodecane from T=(288.15 to 328.15) K
rhol	935.08	kg/m3	328.15	Excess Molar Volume and Viscosity Deviation of [C2mim][NTf2]/[C4mim][NTf2] + DMC/DEC
rhol	940.86	kg/m3	323.15	Excess Molar Volume and Viscosity Deviation of [C2mim][NTf2]/[C4mim][NTf2] + DMC/DEC
rhol	946.60	kg/m3	318.15	Excess Molar Volume and Viscosity Deviation of [C2mim][NTf2]/[C4mim][NTf2] + DMC/DEC

rhol	952.32	kg/m3	313.15	Excess Molar Volume and Viscosity Deviation of [C2mim][NTf2]/[C4mim][NTf2] + DMC/DEC
rhol	958.00	kg/m3	308.15	Excess Molar Volume and Viscosity Deviation of [C2mim][NTf2]/[C4mim][NTf2] + DMC/DEC
rhol	963.66	kg/m3	303.15	Excess Molar Volume and Viscosity Deviation of [C2mim][NTf2]/[C4mim][NTf2] + DMC/DEC
rhol	969.30	kg/m3	298.15	Excess Molar Volume and Viscosity Deviation of [C2mim][NTf2]/[C4mim][NTf2] + DMC/DEC
rhol	974.91	kg/m3	293.15	Excess Molar Volume and Viscosity Deviation of [C2mim][NTf2]/[C4mim][NTf2] + DMC/DEC
rhol	980.49	kg/m3	288.15	Excess Molar Volume and Viscosity Deviation of [C2mim][NTf2]/[C4mim][NTf2] + DMC/DEC
rhol	969.08	kg/m3	298.15	Evaluation of Diethyl Carbonate and Methyl Isobutyl Ketone as Entrainers for the Separation of 1-Hexene and n-Hexane
rhol	967.83	kg/m3	298.15	Isobaric Vapor Liquid Equilibrium for the Binary Systems of Diethyl Carbonate with Xylene Isomers and Ethylbenzene at 101.33 kPa

rhol	969.00	kg/m3	298.15	Isobaric Vapor Liquid Equilibria for Binary Systems of Diethyl Carbonate + Propylene Carbonate, Diethyl Carbonate + Propylene Glycol, and Ethanol + Propylene Carbonate at 101.3 kPa	
rhol	969.32	kg/m3	298.15	Influence of alkyl group and temperature on thermophysical properties of carboxylic acid and their binary mixtures	
rhol	935.20	kg/m3	328.15	Density and viscosity of four binary mixtures of [C2mmim][NTf2]/[C4mmim][NTf2] + dimethyl carbonate/diethyl carbonate	
rhol	940.90	kg/m3	323.15	Density and viscosity of four binary mixtures of [C2mmim][NTf2]/[C4mmim][NTf2] + dimethyl carbonate/diethyl carbonate	
rhol	946.70	kg/m3	318.15	Density and viscosity of four binary mixtures of [C2mmim][NTf2]/[C4mmim][NTf2] + dimethyl carbonate/diethyl carbonate	
rhol	952.40	kg/m3	313.15	Density and viscosity of four binary mixtures of [C2mmim][NTf2]/[C4mmim][NTf2] + dimethyl carbonate/diethyl carbonate	
rhol	958.00	kg/m3	308.15	Density and viscosity of four binary mixtures of [C2mmim][NTf2]/[C4mmim][NTf2] + dimethyl carbonate/diethyl carbonate	

rhol	963.70	kg/m3	303.15	Density and viscosity of four binary mixtures of [C2mmim][NTf2]/[C4mmim][NTf2] + dimethyl carbonate/diethyl carbonate
rhol	969.30	kg/m3	298.15	Density and viscosity of four binary mixtures of [C2mmim][NTf2]/[C4mmim][NTf2] + dimethyl carbonate/diethyl carbonate
rhol	974.90	kg/m3	293.15	Density and viscosity of four binary mixtures of [C2mmim][NTf2]/[C4mmim][NTf2] + dimethyl carbonate/diethyl carbonate
rhol	940.70	kg/m3	323.15	Thermophysical and optical studies of molecular interactions in binary mixtures of diethyl carbonate with aromatic compounds at temperatures from 298.15 to 323.15 K
rhol	946.40	kg/m3	318.15	Thermophysical and optical studies of molecular interactions in binary mixtures of diethyl carbonate with aromatic compounds at temperatures from 298.15 to 323.15 K
rhol	952.10	kg/m3	313.15	Thermophysical and optical studies of molecular interactions in binary mixtures of diethyl carbonate with aromatic compounds at temperatures from 298.15 to 323.15 K

rhol	957.80	kg/m3	308.15	Thermophysical and optical studies of molecular interactions in binary mixtures of diethyl carbonate with aromatic compounds at temperatures from 298.15 to 323.15 K	
rhol	963.50	kg/m3	303.15	Thermophysical and optical studies of molecular interactions in binary mixtures of diethyl carbonate with aromatic compounds at temperatures from 298.15 to 323.15 K	
rhol	969.10	kg/m3	298.15	Thermophysical and optical studies of molecular interactions in binary mixtures of diethyl carbonate with aromatic compounds at temperatures from 298.15 to 323.15 K	
rhol	970.00	kg/m3	298.15	Low pressure carbon dioxide solubility in lithium-ion batteries based electrolytes as a function of temperature. Measurement and prediction	
rhol	935.12	kg/m3	328.15	Permittivity and density of binary systems of {dimethyl or diethyl carbonate} + n-dodecane from T=(288.15 to 328.15) K	

rhol	969.10	kg/m3	298.15	Viscosities of Dimethyl Carbonate or Diethyl Carbonate with Alkanes at Four Temperatures. New UNIFAC-VISCO Parameters	
rhol	952.15	kg/m3	313.15	Densities and Surface Tensions of Trimethylbenzene + Dimethyl Carbonate or + Diethyl Carbonate at 298.15 K and 313.15 K	
rhol	947.67	kg/m3	318.15	Permittivity and density of binary systems of {dimethyl or diethyl carbonate} + n-dodecane from T=(288.15 to 328.15) K	
rhol	969.12	kg/m3	298.15	Densities and Surface Tensions of Trimethylbenzene + Dimethyl Carbonate or + Diethyl Carbonate at 298.15 K and 313.15 K	
rhol	957.90	kg/m3	308.15	Excess Enthalpies, Densities, Viscosities, and Refractive Indices of Binary Mixtures Involving Some Poly(glycols) + Diethyl Carbonate at 308.15 K	
rhol	952.43	kg/m3	313.15	Densities, Viscosities, and Refractive Indices of New Mixtures of Poly(ethylene glycols) + Dialkyl Carbonates at 313.15 K	

rhol	969.10	kg/m3	298.15	Dynamic Viscosities of Diethyl Carbonate with Linear and Secondary Alcohols at Several Temperatures	
rhol	969.18	kg/m3	298.15	Solid-liquid equilibria for selected binary systems containing diphenyl carbonate	
rhol	980.29	kg/m3	288.15	Permittivity and density of binary systems of {dimethyl or diethyl carbonate} + n-dodecane from T=(288.15 to 328.15) K	
rhol	974.57	kg/m3	293.15	Permittivity and density of binary systems of {dimethyl or diethyl carbonate} + n-dodecane from T=(288.15 to 328.15) K	
rhol	969.07	kg/m3	298.15	Permittivity and density of binary systems of {dimethyl or diethyl carbonate} + n-dodecane from T=(288.15 to 328.15) K	
rhol	963.60	kg/m3	303.15	Permittivity and density of binary systems of {dimethyl or diethyl carbonate} + n-dodecane from T=(288.15 to 328.15) K	
rhol	969.23	kg/m3	298.15	Excess Properties of Binary Mixtures of Esters of Carbonic Acid + Three Aryl Alcohols at 308.15 K	

speedsl	1178.00	m/s	298.15	Vapor liquid equilibria for systems of diethyl carbonate and ketones and determination of group interaction parameters for the UNIFAC and ASOG methods
srf	0.02	N/m	338.17	Surface tension of diethyl carbonate, 1,2-dimethoxyethane and diethyl adipate
srf	0.02	N/m	348.17	Surface tension of diethyl carbonate, 1,2-dimethoxyethane and diethyl adipate
srf	0.02	N/m	353.19	Surface tension of diethyl carbonate, 1,2-dimethoxyethane and diethyl adipate
srf	0.02	N/m	358.18	Surface tension of diethyl carbonate, 1,2-dimethoxyethane and diethyl adipate
srf	0.02	N/m	363.16	Surface tension of diethyl carbonate, 1,2-dimethoxyethane and diethyl adipate
srf	0.02	N/m	368.18	Surface tension of diethyl carbonate, 1,2-dimethoxyethane and diethyl adipate
srf	0.02	N/m	373.17	Surface tension of diethyl carbonate, 1,2-dimethoxyethane and diethyl adipate
srf	0.03	N/m	298.15	Surface tension, density, and speed of sound for the ternary mixture {diethyl carbonate + p-xylene + decane}

srf	0.02	N/m	308.15	Surface Tension of Dialkyl Carbonates + (Alkanes or 1,4-Dimethylbenzene) and 1,4-Dimethylbenzene + Alkanes Binary Mixtures at T = 308.15 K
srf	0.02	N/m	333.18	Surface tension of diethyl carbonate, 1,2-dimethoxyethane and diethyl adipate
srf	0.02	N/m	328.12	Surface tension of diethyl carbonate, 1,2-dimethoxyethane and diethyl adipate
srf	0.02	N/m	323.16	Surface tension of diethyl carbonate, 1,2-dimethoxyethane and diethyl adipate
srf	0.02	N/m	318.16	Surface tension of diethyl carbonate, 1,2-dimethoxyethane and diethyl adipate
srf	0.02	N/m	313.19	Surface tension of diethyl carbonate, 1,2-dimethoxyethane and diethyl adipate
srf	0.03	N/m	308.11	Surface tension of diethyl carbonate, 1,2-dimethoxyethane and diethyl adipate
srf	0.03	N/m	303.17	Surface tension of diethyl carbonate, 1,2-dimethoxyethane and diethyl adipate
srf	0.02	N/m	343.17	Surface tension of diethyl carbonate, 1,2-dimethoxyethane and diethyl adipate

srf	0.03	N/m	298.10	Surface tension of diethyl carbonate, 1,2-dimethoxyethane and diethyl adipate
srf	0.03	N/m	293.16	Surface tension of diethyl carbonate, 1,2-dimethoxyethane and diethyl adipate
srf	0.03	N/m	288.18	Surface tension of diethyl carbonate, 1,2-dimethoxyethane and diethyl adipate
srf	0.03	N/m	278.18	Surface tension of diethyl carbonate, 1,2-dimethoxyethane and diethyl adipate
srf	0.03	N/m	283.18	Surface tension of diethyl carbonate, 1,2-dimethoxyethane and diethyl adipate
srf	0.03	N/m	273.18	Surface tension of diethyl carbonate, 1,2-dimethoxyethane and diethyl adipate

Correlations

Information	Value
Property code	pvap
Equation	ln(Pvp) = A + B/T + C*ln(T) + D*T^2
Coeff. A	1.76598e+02
Coeff. B	-1.16692e+04
Coeff. C	-2.43764e+01
Coeff. D	2.02849e-05
Temperature range (K), min.	230.15
Temperature range (K), max.	576.00

Datasets

Mass density, kg/m3

Temperature, K - Liquid	Pressure, kPa - Liquid	Mass density, kg/m3 - Liquid
288.15	100.00	980.3
288.15	5000.00	984.5
288.15	10000.00	988.6
288.15	20000.00	996.3
288.15	30000.00	1003.6
288.15	40000.00	1010.5
298.15	100.00	968.9
298.15	5000.00	973.8
298.15	10000.00	977.8
298.15	20000.00	986.0
298.15	30000.00	993.6
298.15	40000.00	1000.8
308.15	100.00	957.7
308.15	5000.00	962.4
308.15	10000.00	967.1
308.15	20000.00	975.7
308.15	30000.00	983.8
308.15	40000.00	991.3

Reference *https://www.doi.org/10.1016/j.jct.2012.11.011*

Temperature, K	Pressure, kPa	Mass density, kg/m3
298.15	100.00	969.3

Reference *https://www.doi.org/10.1016/j.jct.2014.06.004*

Temperature, K	Pressure, kPa	Mass density, kg/m3
298.15	100.00	969.4

Reference *https://www.doi.org/10.1021/acs.jced.9b00430*

Sources

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Densities, Viscosities, and Refractive Indices of New Mixtures of Poly(ethylene glycols) + Dialkyl Carbonates at 313.15 K:	https://www.doi.org/10.1021/je034159d
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Densities and Viscosities of Four Binary Diethyl Carbonate + 1-Alcohol Systems from (288.15 to 313.15) K:	https://www.doi.org/10.1021/je1013476
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Joback Method:	https://en.wikipedia.org/wiki/Joback_method
Dynamic Viscosities of Diethyl Carbonate with Linear and Secondary Alcohols at Several Temperatures:	https://www.doi.org/10.1021/je0341413
Isobaric Vapor Liquid Equilibria for Binary Systems of Diethyl Carbonate + Propylene Carbonate, Diethyl Carbonate + Propylene Glycol, and Ethanol + Propylene Carbonate at 101.3 kPa:	https://www.doi.org/10.1021/acs.jced.5b00064
Volumetric properties of dialkyl carbonate + n-alkane mixtures at high pressures: Experimental measurement and Nitta–Chao model prediction:	https://www.doi.org/10.1016/j.jct.2012.11.011
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Phase Equilibria Involved in the Extractive Distillation of Cyclohexane + Cyclohexene Using Diethyl Carbonate as an Entrainer:	https://www.doi.org/10.1021/je200784y
Low pressure methane solubility in lithium-ion batteries based solvents and electrolytes as a function of temperature. Measurement and prediction:	https://www.doi.org/10.1016/j.jct.2014.07.004
Excess Molar Volume and Viscosity Deviation of [C2mim][NTf2]/[C4mim][NTf2] + DMC/DEC:	https://www.doi.org/10.1021/acs.jced.8b00591
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Legend

chl:	Standard liquid enthalpy of combustion
cpg:	Ideal gas heat capacity
cpl:	Liquid phase heat capacity
dvisc:	Dynamic viscosity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfl:	Liquid phase enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hfust:	Enthalpy of fusion at a given temperature
hvap:	Enthalpy of vaporization at standard conditions
hvapt:	Enthalpy of vaporization at a given temperature
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
nfpaf:	NFPA Fire Rating
nfpas:	NFPA Safety Rating
pc:	Critical Pressure
pvap:	Vapor pressure
rfi:	Refractive Index
rhol:	Liquid Density
rinpol:	Non-polar retention indices
ripol:	Polar retention indices
speedsl:	Speed of sound in fluid
srf:	Surface Tension
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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