

# Cyclodecane, methyl-

<b>Other names:</b>	Methylcyclodecane
<b>Inchi:</b>	InChI=1S/C11H22/c1-11-9-7-5-3-2-4-6-8-10-11/h11H,2-10H2,1H3
<b>InchiKey:</b>	WFWNFBYXZQJMFM-UHFFFAOYSA-N
<b>Formula:</b>	C11H22
<b>SMILES:</b>	CC1CCCCCCCCC1
<b>Mol. weight [g/mol]:</b>	154.29
<b>CAS:</b>	13151-43-4

## Physical Properties

Property code	Value	Unit	Source
gf	17.79	kJ/mol	Joback Method
hf	-240.69	kJ/mol	Joback Method
hfus	7.68	kJ/mol	Joback Method
hvap	41.20	kJ/mol	Joback Method
log10ws	-4.08		Crippen Method
logp	4.147		Crippen Method
mcvol	154.990	ml/mol	McGowan Method
pc	2579.34	kPa	Joback Method
rinpol	1195.00		NIST Webbook
rinpol	1202.00		NIST Webbook
rinpol	1195.00		NIST Webbook
tb	487.71	K	Joback Method
tc	714.38	K	Joback Method
tf	207.03	K	Joback Method
vc	0.552	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	348.47	J/molxK	487.71	Joback Method
cpg	373.55	J/molxK	525.49	Joback Method
cpg	397.33	J/molxK	563.27	Joback Method
cpg	419.81	J/molxK	601.05	Joback Method
cpg	441.00	J/molxK	638.82	Joback Method

cpg	460.91	J/molxK	676.60	Joback Method
cpg	479.53	J/molxK	714.38	Joback Method
dvisc	0.0751750	Paxs	207.03	Joback Method
dvisc	0.0094962	Paxs	253.81	Joback Method
dvisc	0.0022840	Paxs	300.59	Joback Method
dvisc	0.0008064	Paxs	347.37	Joback Method
dvisc	0.0003645	Paxs	394.15	Joback Method
dvisc	0.0001950	Paxs	440.93	Joback Method
dvisc	0.0001176	Paxs	487.71	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C13151434&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C13151434&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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