

# Isophthalic acid, isohexyl pent-4-enyl ester

<b>Inchi:</b>	InChI=1S/C19H26O4/c1-4-5-6-12-22-18(20)16-10-7-11-17(14-16)19(21)23-13-8-9-15(2)3
<b>InchiKey:</b>	MGNWETZBCZH XIQ-UHFFFAOYSA-N
<b>Formula:</b>	C19H26O4
<b>SMILES:</b>	C=CCCCOC(=O)c1cccc(C(=O)OCCCC(C)C)c1
<b>Mol. weight [g/mol]:</b>	318.41

## Physical Properties

Property code	Value	Unit	Source
gf	-170.56	kJ/mol	Joback Method
hf	-579.88	kJ/mol	Joback Method
hfus	39.39	kJ/mol	Joback Method
hvap	78.08	kJ/mol	Joback Method
log10ws	-5.34		Crippen Method
logp	4.403		Crippen Method
mvol	265.390	ml/mol	McGowan Method
pc	1497.67	kPa	Joback Method
rinpol	2303.00		NIST Webbook
rinpol	2303.00		NIST Webbook
tb	814.60	K	Joback Method
tc	1018.14	K	Joback Method
tf	470.39	K	Joback Method
vc	1.014	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	792.85	J/molxK	814.60	Joback Method
cpg	808.44	J/molxK	848.52	Joback Method
cpg	822.93	J/molxK	882.45	Joback Method
cpg	836.36	J/molxK	916.37	Joback Method
cpg	848.74	J/molxK	950.29	Joback Method
cpg	860.11	J/molxK	984.22	Joback Method
cpg	870.48	J/molxK	1018.14	Joback Method
dvisc	0.0007559	Paxs	470.39	Joback Method

dvisc	0.0003966	Paxs	527.76	Joback Method
dvisc	0.0002362	Paxs	585.13	Joback Method
dvisc	0.0001543	Paxs	642.50	Joback Method
dvisc	0.0001080	Paxs	699.86	Joback Method
dvisc	0.0000799	Paxs	757.23	Joback Method
dvisc	0.0000616	Paxs	814.60	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U356715&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U356715&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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