

# Indane, 1-phenyl

<b>Inchi:</b>	InChI=1S/C15H14/c1-2-6-12(7-3-1)15-11-10-13-8-4-5-9-14(13)15/h1-9,15H,10-11H2
<b>InchiKey:</b>	VNAFWALXWOAPCK-UHFFFAOYSA-N
<b>Formula:</b>	C15H14
<b>SMILES:</b>	<chem>c1ccc(C2CCc3ccccc32)cc1</chem>
<b>Mol. weight [g/mol]:</b>	194.27

## Physical Properties

Property code	Value	Unit	Source
gf	351.36	kJ/mol	Joback Method
hf	181.46	kJ/mol	Joback Method
hfus	20.43	kJ/mol	Joback Method
hvap	54.11	kJ/mol	Joback Method
log10ws	-4.23		Crippen Method
logp	3.765		Crippen Method
mcvol	163.830	ml/mol	McGowan Method
pc	2826.33	kPa	Joback Method
rinpola	1618.00		NIST Webbook
tb	607.68	K	Joback Method
tc	861.33	K	Joback Method
tf	342.11	K	Joback Method
vc	0.617	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	406.94	J/molxK	607.68	Joback Method
cpg	425.64	J/molxK	649.95	Joback Method
cpg	442.78	J/molxK	692.23	Joback Method
cpg	458.49	J/molxK	734.50	Joback Method
cpg	472.90	J/molxK	776.78	Joback Method
cpg	486.15	J/molxK	819.05	Joback Method
cpg	498.37	J/molxK	861.33	Joback Method
dvisc	0.0018621	Paxs	342.11	Joback Method
dvisc	0.0012365	Paxs	386.37	Joback Method

dvisc	0.0008932	Paxs	430.63	Joback Method
dvisc	0.0006855	Paxs	474.89	Joback Method
dvisc	0.0005504	Paxs	519.16	Joback Method
dvisc	0.0004575	Paxs	563.42	Joback Method
dvisc	0.0003906	Paxs	607.68	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R73004&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R73004&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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