

Thiocyanic acid, 1,1-dimethyl-3-oxobutyl ester

Inchi:	InChI=1S/C7H11NOS/c1-6(9)4-7(2,3)10-5-8/h4H2,1-3H3
InchiKey:	KZOYOWHFCLDEAN-UHFFFAOYSA-N
Formula:	C7H11NOS
SMILES:	CC(=O)CC(C)(C)SC#N
Mol. weight [g/mol]:	157.23
CAS:	26094-11-1

Physical Properties

Property code	Value	Unit	Source
gf	48.28	kJ/mol	Joback Method
hf	-102.39	kJ/mol	Joback Method
hfus	13.71	kJ/mol	Joback Method
hvap	53.92	kJ/mol	Joback Method
log10ws	-2.39		Crippen Method
logp	1.958		Crippen Method
mcvol	128.790	ml/mol	McGowan Method
pc	3032.27	kPa	Joback Method
tb	581.06	K	Joback Method
tc	809.82	K	Joback Method
tf	320.39	K	Joback Method
vc	0.502	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	291.98	J/molxK	581.06	Joback Method
cpg	302.59	J/molxK	619.19	Joback Method
cpg	312.47	J/molxK	657.31	Joback Method
cpg	321.66	J/molxK	695.44	Joback Method
cpg	330.18	J/molxK	733.57	Joback Method
cpg	338.07	J/molxK	771.69	Joback Method
cpg	345.36	J/molxK	809.82	Joback Method

Sources

Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci9903071
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C26094111&Units=SI

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mccvol:	McGowan's characteristic volume
pc:	Critical Pressure
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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