

# Benzonitrile, 3-hydroxy-

<b>Other names:</b>	Benzonitrile, m-hydroxy- m-Cyanophenol m-Hydroxybenzonitrile 3-Cyanophenol 3-Hydroxybenzonitrile 3-Hydroxybenzoic acid nitrile
<b>Inchi:</b>	InChI=1S/C7H5NO/c8-5-6-2-1-3-7(9)4-6/h1-4,9H
<b>InchiKey:</b>	SGHBRHKBCLLVCI-UHFFFAOYSA-N
<b>Formula:</b>	C7H5NO
<b>SMILES:</b>	N#Cc1cccc(O)c1
<b>Mol. weight [g/mol]:</b>	119.12
<b>CAS:</b>	873-62-1

## Physical Properties

Property code	Value	Unit	Source
gf	99.03	kJ/mol	Joback Method
hf	36.29	kJ/mol	Joback Method
hfus	15.22	kJ/mol	Joback Method
hvap	56.94	kJ/mol	Joback Method
ie	9.39 ± 0.05	eV	NIST Webbook
log10ws	-1.37		Crippen Method
logp	1.264		Crippen Method
mcvol	92.980	ml/mol	McGowan Method
pc	4856.20	kPa	Joback Method
tb	568.94	K	Joback Method
tc	817.38	K	Joback Method
tf	371.78	K	Joback Method
vc	0.311	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	199.90	J/mol×K	568.94	Joback Method
cpg	207.76	J/mol×K	610.35	Joback Method

cpg	214.93	J/mol×K	651.75	Joback Method
cpg	221.50	J/mol×K	693.16	Joback Method
cpg	227.57	J/mol×K	734.57	Joback Method
cpg	233.22	J/mol×K	775.98	Joback Method
cpg	238.57	J/mol×K	817.38	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C873621&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C873621&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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