

Hexafluoroisopropyl methacrylate

Other names:	2-Propenoic acid, 2-methyl-, 2,2,2-trifluoro-1-(trifluoromethyl)ethyl ester 1,1,1,3,3,3-Hexafluoroisopropyl methacrylate 2,2,2-trifluoro-1-(trifluoromethyl)ethyl methacrylate
Inchi:	InChI=1S/C7H6F6O2/c1-3(2)4(14)15-5(6(8,9)10)7(11,12)13/h5H,1H2,2H3
InchiKey:	FMQPBWHSNCRVQJ-UHFFFAOYSA-N
Formula:	C7H6F6O2
SMILES:	C=C(C)C(=O)OC(C(F)(F)F)C(F)(F)F
Mol. weight [g/mol]:	236.11
CAS:	3063-94-3

Physical Properties

Property code	Value	Unit	Source
gf	-1312.19	kJ/mol	Joback Method
hf	-1516.41	kJ/mol	Joback Method
hfus	14.21	kJ/mol	Joback Method
hvap	31.86	kJ/mol	Joback Method
log10ws	-2.91		Crippen Method
logp	2.599		Crippen Method
mcvol	123.250	ml/mol	McGowan Method
pc	2448.32	kPa	Joback Method
tb	372.00	K	NIST Webbook
tc	576.69	K	Joback Method
tf	218.47	K	Joback Method
vc	0.513	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	277.19	J/mol×K	421.13	Joback Method
cpg	287.45	J/mol×K	447.06	Joback Method
cpg	297.14	J/mol×K	472.98	Joback Method
cpg	306.28	J/mol×K	498.91	Joback Method
cpg	314.89	J/mol×K	524.84	Joback Method
cpg	323.00	J/mol×K	550.76	Joback Method

Sources

McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C3063943&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Crippen Method:	https://www.cheméo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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