

# Benzene, 1,1',1'',1'''-(1,3-butadiene-1,2,3,4-tetrayl)tetrakis-

Other names:

1,3-Butadiene, 1,2,3,4-tetraphenyl-

1,2,3,4-Tetraphenyl-1,3-butadiene

(1-Benzylidene-2,3-diphenyl-2-propenyl)benzene

Inchi: InChI=1S/C28H22/c1-5-13-23(14-6-1)21-27(25-17-9-3-10-18-25)28(26-19-11-4-12-20-26

InchiKey: DAABVBOFAIYKNX-GPAWKIAZSA-N

Formula: C28H22

SMILES: C(=C(C(=Cc1ccccc1)c1ccccc1)c1ccccc1)c1ccccc1

Mol. weight [g/mol]: 358.47

CAS: 806-71-3

## Physical Properties

Property code	Value	Unit	Source
chs	-14521.30 ± 3.00	kJ/mol	NIST Webbook
gf	777.86	kJ/mol	Joback Method
hf	539.73	kJ/mol	Joback Method
hfus	42.22	kJ/mol	Joback Method
hvap	87.10	kJ/mol	Joback Method
log10ws	-8.33		Crippen Method
logp	7.468		Crippen Method
mcvol	301.740	ml/mol	McGowan Method
pc	1641.76	kPa	Joback Method
tb	954.84	K	Joback Method
tc	1237.13	K	Joback Method
tf	472.92	K	Joback Method
vc	1.133	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	911.19	J/mol×K	954.84	Joback Method
cpg	928.75	J/mol×K	1001.89	Joback Method
cpg	945.32	J/mol×K	1048.94	Joback Method
cpg	961.20	J/mol×K	1095.98	Joback Method
cpg	976.74	J/mol×K	1143.03	Joback Method

cpg	992.26	J/mol×K	1190.08	Joback Method
cpg	1008.08	J/mol×K	1237.13	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C806713&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C806713&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>chs:</b>	Standard solid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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