

# 2-Benzothiazolamine

<b>Other names:</b>	2(3H)-Benzothiazolimine 2-Aminobenzthiazole 2-Benzothiazolylamine 2-Iminobenzothiazoline 2-aminobenzo[d]thiazole 2-aminobenzothiazole Benzothiazol-2-ylamine Benzothiazole, 2-amino- NSC 4670 USAF EK-3941 USAF XR-27 o-Aminobenzothiazole
<b>Inchi:</b>	InChI=1S/C7H6N2S/c8-7-9-5-3-1-2-4-6(5)10-7/h1-4H,(H2,8,9)
<b>InchiKey:</b>	UHGULLIUJBCTEF-UHFFFAOYSA-N
<b>Formula:</b>	C7H6N2S
<b>SMILES:</b>	<chem>Nc1nc2ccccc2s1</chem>
<b>Mol. weight [g/mol]:</b>	150.20
<b>CAS:</b>	136-95-8

## Physical Properties

Property code	Value	Unit	Source
log10ws	-2.43		Crippen Method
logp	1.878		Crippen Method
mvol	106.880	ml/mol	McGowan Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
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cps	153.21	J/mol×K	298.15	Standard enthalpies of formation of 2-aminobenzothiazoles in the crystalline phase by rotating-bomb combustion calorimetry
hvapt	102.10	kJ/mol	298.15	Thermodynamic study of 2-aminothiazole and 2-aminobenzothiazole: Experimental and computational approaches

## Sources

Crippen Method:	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
Crippen Method:	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
Standard enthalpies of formation of 2-aminobenzothiazoles in the crystalline phase by rotating-bomb combustion calorimetry:	<a href="https://www.doi.org/10.1016/j.jct.2014.01.018">https://www.doi.org/10.1016/j.jct.2014.01.018</a>
Thermodynamic study of 2-aminothiazole and 2-aminobenzothiazole: Experimental and computational approaches:	<a href="https://www.doi.org/10.1016/j.jct.2014.04.001">https://www.doi.org/10.1016/j.jct.2014.04.001</a>
NIST Webbook:	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C136958&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C136958&amp;Units=SI</a>

## Legend

<b>cps:</b>	Solid phase heat capacity
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume

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