

# Cyclohexanone, 3,3,5-trimethyl-

<b>Other names:</b>	(dl) 3,5,5-trimethylcyclohexanone 3,3,5-Trimethylcyclohexan-1-one 3,3,5-Trimethylcyclohexanone 3,3,5-trimethyl-cyclohexanone Dihydroisophorone
<b>Inchi:</b>	InChI=1S/C9H16O/c1-7-4-8(10)6-9(2,3)5-7/h7H,4-6H2,1-3H3
<b>InchiKey:</b>	POSWICCRDBKBMH-UHFFFAOYSA-N
<b>Formula:</b>	C9H16O
<b>SMILES:</b>	CC1CC(=O)CC(C)(C)C1
<b>Mol. weight [g/mol]:</b>	140.22
<b>CAS:</b>	873-94-9

## Physical Properties

Property code	Value	Unit	Source
gf	-86.44	kJ/mol	Joback Method
hf	-317.57	kJ/mol	Joback Method
hfus	5.18	kJ/mol	Joback Method
hvap	38.84	kJ/mol	Joback Method
log10ws	-2.28		Crippen Method
logp	2.402		Crippen Method
mvol	128.380	ml/mol	McGowan Method
pc	2986.06	kPa	Joback Method
rinpol	1285.00		NIST Webbook
rinpol	1285.00		NIST Webbook
tb	488.26	K	Joback Method
tc	714.57	K	Joback Method
tf	286.45	K	Joback Method
vc	0.476	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	292.44	J/mol×K	488.26	Joback Method
cpg	310.90	J/mol×K	525.98	Joback Method

cpg	328.31	J/mol×K	563.70	Joback Method
cpg	344.77	J/mol×K	601.42	Joback Method
cpg	360.36	J/mol×K	639.14	Joback Method
cpg	375.16	J/mol×K	676.85	Joback Method
cpg	389.26	J/mol×K	714.57	Joback Method
hvapt	39.30	kJ/mol	443.00	NIST Webbook

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.38745e+01
Coeff. B	-3.70503e+03
Coeff. C	-6.96280e+01
Temperature range (K), min.	305.72
Temperature range (K), max.	502.31

## Sources

The Yaws Handbook of Vapor Pressure:  
Crippen Method:

<https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure>  
<http://pubs.acs.org/doi/abs/10.1021/ci990307i>

Crippen Method:

[https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)

Joback Method:

[https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

McGowan Method:

<http://link.springer.com/article/10.1007/BF02311772>

NIST Webbook:

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C873949&Units=SI>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature

<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>rinpola:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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