# potassium chloride

Inchi: InChl=1S/CIH.K/h1H;/q;+1/p-1

InchiKey: WCUXLLCKKVVCTQ-UHFFFAOYSA-M

Formula: CIK

**SMILES:** [CI-].[K+] **Mol. weight [g/mol]:** 74.55

**CAS**: 7447-40-7

## **Physical Properties**

Property code	Value	Unit	Source
ea	$0.58 \pm 0.01$	eV	NIST Webbook
ea	0.63	eV	NIST Webbook
ea	1.27	eV	NIST Webbook
ie	8.40 ± 0.10	eV	NIST Webbook
ie	10.50	eV	NIST Webbook
ie	8.30	eV	NIST Webbook
ie	8.40 ± 0.10	eV	NIST Webbook
ie	10.10	eV	NIST Webbook
ie	$8.00 \pm 0.30$	eV	NIST Webbook
ie	8.70	eV	NIST Webbook
tt	1045.15	К	Fusion characterization of biomass ash

## **Temperature Dependent Properties**

Property code	Value	Unit	Temperature [K]	Source
rhos	1793.60	kg/m3	1023.00	Density of Crystalline Alkali Chlorides and Their Eutectic Mixtures Near the Melting Point
rhos	1805.80	kg/m3	973.00	Density of Crystalline Alkali Chlorides and Their Eutectic Mixtures Near the Melting Point

rhos	1803.60	kg/m3	983.00	Density of Crystalline Alkali Chlorides and Their Eutectic Mixtures Near the Melting Point	
rhos	1800.60	kg/m3	993.00	Density of Crystalline Alkali Chlorides and Their Eutectic Mixtures Near the Melting Point	
rhos	1797.10	kg/m3	1003.00	Density of Crystalline Alkali Chlorides and Their Eutectic Mixtures Near the Melting Point	
rhos	1794.50	kg/m3	1013.00	Density of Crystalline Alkali Chlorides and Their Eutectic Mixtures Near the Melting Point	
rhos	1793.60	kg/m3	1033.00	Density of Crystalline Alkali Chlorides and Their Eutectic Mixtures Near the Melting Point	
rhos	1796.60	kg/m3	1043.00	Density of Crystalline Alkali Chlorides and Their Eutectic Mixtures Near the Melting Point	
srf	0.09	N/m	1230.00	Surface tension of light rare earth fluoride molten salts electrolytesystem	

## **Correlations**

Information Value

Property code	pvap	
Equation	In(Pvp) = A + B/(T + C)	
Coeff. A	1.39492e+01	
Coeff. B	-1.33788e+04	
Coeff. C	-2.55170e+02	
Temperature range (K), min.	1044.00	
Temperature range (K), max.	3470.00	

#### Sources

https://www.doi.org/10.1021/je401009p Determination and Modeling of the Solubility of Na2siO3\*9H2O in the NaCl Rolubility of Na2siO3\*9H2O in the NaCl Rolubility of Na2siO3\*9H2O in the NaCl Rolubility of Na2siom Chloride and Potassium Chloride in Water + Ethanol Rolubility for K 12824 (12824 (12824) (12824) https://www.doi.org/10.1021/je400911m and Their Mixture Solutions:

Stable (solid + liquid) phase equilibrium https://www.doi.org/10.1016/j.jct.2015.06.011 for the ternary systems (K2SO4 + Refearings) https://www.doi.org/10.1016/j.jct.2010.0 jអ្នក់ខេត្ត for the ternary systems (K2SO4 + Refearings) https://www.doi.org/10.1016/j.jct.2010.0 jអ្នក់ខេត្ត for the ternary systems (the ternary systems for the ternary systems (the https://www.doi.org/10.1016/j.jct.2010.01.017 Beirgetive inclines for the Ternary
Beirgetive inclines for the Ternary
Systems we inclines for the Ternary
Billy lines by sociating in the Lines in the Systems of the Sys https://www.doi.org/10.1021/je400959n https://www.doi.org/10.1016/j.fluid.2019.04.018 https://www.doi.org/10.1016/j.jct.2018.12.011 quaternary systems:
Liternacy waternary syste MGHaftes (เดาะเก๊ะ คลูเน่งเน้าหลดร์ย่งe different amino acids in water: Phase Equilibrium for the Ternary https://www.doi.org/10.1016/j.fluid.2007.04.004 https://www.doi.org/10.1021/acs.jced.5b01111 Systems (KCI + K2SO4 + H2O) and Rail 2 Moter Rives & Familia by a 128 e 15 and Swaten School 2 by 18 to 18 and 18 https://www.doi.org/10.1021/acs.jced.6b00981 https://www.doi.org/10.1016/j.jct.2015.07.029 enthalpies of dilution and dissociation https://www.doi.org/10.1016/j.jct.2015.12.026 **Myskolodisheeinnicaeftiqieatrue**bus salt isopressed tailonniane and illustration to the same state of the https://www.doi.org/10.1016/j.jct.2013.12.017 Michalphierusiare reflicients of 3-methyl-1-butanol + n-heptane and practions in the property of the property https://www.doi.org/10.1016/j.jct.2018.10.023 Poliskianichavillersolotion: https://www.doi.org/10.1021/acs.jced.5b00579 Quaternary System K+//H2PO42, SO42
Stuply One point in https://www.doi.org/10.1021/acs.jced.sb00579
Quaternary System K+//H2PO42, SO42
Stuply One point in https://www.doi.org/10.1016/j.fluid.2006.10.018
ionic liquid 1. The solubility of alkali
Solubilitiem on imidate interior in https://www.doi.org/10.1016/j.jct.2011.03.002
liquids in aqueous salt solutions at
Eggility rium Diagrams of Water + NaCl

Solution 1. 2 Month 1.2 Proposed 1.4 the state of the sta Estation of the light in the light in the light in the light in water and aqueous to the light in the light i https://www.doi.org/10.1016/j.jct.2011.06.024 https://www.doi.org/10.1021/acs.jced.7b00503 Interpretation of billionact revision for Ayabas I concentrated the Itel Itel 23 serial to the content and the Itel Itel 23 serial to the content and the content in the co https://www.doi.org/10.1016/j.tca.2008.10.023 https://www.doi.org/10.1016/j.jct.2006.11.014 https://www.doi.org/10.1016/j.jct.2008.12.011 https://www.doi.org/10.1021/acs.jced.7b00433 https://www.doi.org/10.1021/je9007102 https://www.doi.org/10.1021/acs.jced.7b00840

and p-lodoaniline in Pure Solvents:

```
Liquid Equilibria in the Quaternary
        Shete คพรา chan dbook-on-vapor-pressure standard by the same for the 
    Programme:
Solubility and distribution of bicycle
                                                                                                                                                                                                                                                                                                                                                      https://www.doi.org/10.1016/j.jct.2017.08.012
 Solubility and distribution of bicycle derivatives of 1,3-selenazine in Enthalaises of all alias metal agetates; study on the Phase Equilibrium of Na+, K+//Cl-, CO32-H2O and Single-Salt of Pensis um & Exaras Molan Malumes of the Business 
                                                                                                                                                                                                                                                                                                                                                     https://www.doi.org/10.1016/j.tca.2004.11.004
                                                                                                                                                                                                                                                                                                                                                   https://www.doi.org/10.1021/acs.jced.9b00367
                                                                                                                                                                                                                                                                                                                                                     https://www.doi.org/10.1021/acs.jced.5b00366
                                                                                                                                                                                                                                                                                                                                                     https://www.doi.org/10.1007/s10765-015-2009-x
                                                                                                                                                                                                                                                                                                                                                      https://www.doi.org/10.1021/je100949x
 Surface Tension Measurements of Energy of Measurements of Interpolation of Energy of Measurements of Interpolation of Energy of Measurements of Interpolation of Int
     influence of salt concentration and
                                                                                                                                                                                                                                                                                                                                                     https://www.doi.org/10.1021/acs.jced.7b00800
EMPRICE OF THE CONTROL OF THE CONTRO
                                                                                                                                                                                                                                                                                                                                                      https://www.doi.org/10.1016/j.jct.2010.11.003
                                                                                                                                                                                                                                                                                                                                                      https://www.doi.org/10.1016/j.fluid.2011.01.026
                                                                                                                                                                                                                                                                                                                                                   https://www.doi.org/10.1021/je050048y
                                                                                                                                                                                                                                                                                                                                                     https://www.doi.org/10.1021/acs.jced.6b00960
                                                                                                                                                                                                                                                                                                                                                      https://www.doi.org/10.1021/acs.jced.8b00081
                                                                                                                                                                                                                                                                                                                                                     https://www.doi.org/10.1021/je900849b
                                                                                                                                                                                                                                                                                                                                                      https://www.doi.org/10.1021/acs.jced.7b00413
 30បារាមហេតុខេត្តបានប្រភេទប្រាប
CHRTHUS CHRTHBIAL 298.15 K:
REAL REAL POOLES (IPC) and resume +
Matery strate (អង្គម៉ាស្រីប ប្រើប្រាប់ Leetrical
ម្រឹងនេះ អាចមានប្រាប់ (Leetrical
ម្រឹងនេះ អាចមានប្រាប់ (Leetrical
ម្រឹងនេះ អាចមានប្រាប់ (Leetrical
ប្រាប់ Leetrical (Leetrical Leetrical Lee
                                                                                                                                                                                                                                                                                                                                                     https://www.doi.org/10.1021/je900351t
                                                                                                                                                                                                                                                                                                                                                     https://www.doi.org/10.1016/j.jct.2013.07.024
                                                                                                                                                                                                                                                                                                                                                      https://www.doi.org/10.1021/acs.jced.5b00771
    Hydrophilic Alcohols with Three
Sirrelies อาสเตอ Situsion Soefficients of https://www.doi.org/10.1021/je049582g
    Aminon Acids in Action as Solutions:
Liquid-liquid equilibria for water +
                                                                                                                                                                                                                                                                                                                                                     https://www.doi.org/10.1016/j.fluid.2015.03.050
     1-propanol (or 1-butanol) + potassium
Activitie of มีใส่เรื่องสิทธิการาชาย
                                                                                                                                                                                                                                                                                                                                                      https://www.doi.org/10.1021/je300701m
  Gosfeigiants total managed Insee-Basic
ฟลาคัด Activitie กลายเรื่องเราะ เล่า
Solutions, with and without KCI, at
ชีวธราชอง ynamics Phase Equilibria of
                                                                                                                                                                                                                                                                                                                                                     https://www.doi.org/10.1021/je700466s
                                                                                                                                                                                                                                                                                                                                                     https://www.doi.org/10.1021/je500700d
    the Aqueous Ternary Systems LiCl + Isophing (1) Measing mesus of Osmotic and Activity Coefficients of
                                                                                                                                                                                                                                                                                                                                                      https://www.doi.org/10.1021/acs.jced.8b00400
  and Activity Coefficients of Name and Activity Coefficients of Name and Nam
https://www.doi.org/10.1021/acs.jced.7b00978
https://www.doi.org/10.1021/acs.jced.7b00978
https://www.doi.org/10.1021/je020173z
https://www.doi.org/10.1021/je020173z
https://www.doi.org/10.1016/j.jct.2013.08.018
https://www.doi.org/10.1016/j.jct.2013.08.018
https://www.doi.org/10.1016/j.jct.2013.08.018
https://www.doi.org/10.1016/j.jct.2013.08.018
https://www.doi.org/10.1016/j.ftca.2004.12.008
https://www.doi.org/10.1016/j.fluid.2015.11.018
https://www.doi.org/10.1016/j.fluid.2015.11.018
https://www.doi.org/10.1016/j.fluid.2015.11.018
https://www.doi.org/10.1021/acs.jced.7b00218
https://www.doi.org/10.1021/je5009944
https://www.doi.org/10.1016/j.fluid.2015.11.018
https://www.doi.org/10.1021/je5009944
https://www.doi.org/10.1021/je5009944
https://www.doi.org/10.1016/j.fluid.2015.11.018
https://www.doi.org/10.1021/je5009944
                                                                                                                                                                                                                                                                                                                                                      https://www.doi.org/10.1016/j.tca.2004.12.008
                                                                                                                                                                                                                                                                                                                                                      https://www.doi.org/10.1016/j.fluid.2015.11.018
                                                                                                                                                                                                                                                                                                                                                      https://www.doi.org/10.1021/acs.jced.7b00218
     MgCl2 PbCl2 ZnCl2 H2O at 373 K:
```

Measurement and Prediction of Solid + https://www.doi.org/10.1021/acs.jced.9b00271

Viscosities of L-Histidine/L-Glutamic Acid/L-Tryptophan/Glycylglycine+2 M Resistivity viscosity relationship in liquid liquid critical mixtures with -proline, L-valine, L-leucine + aqueous Religions in the NaCl - KCl system: The modynamic Study of (KCl + Sating-out effect of alkali fields
chlorides (lithium, sodium, and
MutasdiDiffiveirz-Spaffigientrume
Aqueous schlab high-sinassunaa
Makaliness schlab high-sinassunaa
Makaliness schlab high-sinassunaa
Makalinessa satiand
Aganasi Eagl Dagesics of Maria:

Dagesics satian MTRETANDICES+ salts systems at your parties and Viscosity Self-effection of lumbers solifore selive in the selicities from (0 to 70) deg C:
Extractive from ( Saltation controls solutions at ive Dichloromethane, 1,2-Dichloroethane, and 1,2-Dichlorobenzene:

https://www.doi.org/10.1007/s10765-011-0996-9 Rhase Est With richnofos the Torns and https://www.doi.org/10.1021/je501175q
System KCI-CuCl2-H2O at 298.15 K by
Hegtieal Gandove to the Gold CI-CsCI https://www.doi.org/10.1021/acs.jced.5b00682 https://www.doi.org/10.1016/j.fluid.2006.10.027 https://www.doi.org/10.1016/j.jct.2008.07.019 https://www.doi.org/10.1016/j.jct.2013.03.015 Thermodynamic Study of (KCI + Ntps://www.doi.org/10.1021/acs.jced.8b00009 N,N-Dimethylformamide + Water)

Sizetra is Search is Search in Molal Melange and Partial Molal Melange and Partial Molal Gold in the State of the Search is Search in Molal Gold in the ternary system water Data in Molal Solubility data (I) in the preparation Bhoses is agrees of the Fig. 15 for the ternary system water Data in Molal Salt in Molal Search is Molal Search in https://www.doi.org/10.1021/acs.jced.8b00009 Density of Aqueous Alkali Halide Salt Shamble specifications for water + Someonic for the specification of the spe https://www.doi.org/10.1021/je200808q https://www.doi.org/10.1021/je060492g Aftainable entirement sate soft target:
Pressures over Saturated Aqueous sanatories with the sacutations with the sacutation of sacutations of Potassium Chloride and believe in sacutations with the sacutation of solubilities of Potassium Chloride and believe in sacutations with the sacutation of sacutations of Potassium Chloride and believe in sacutations with the sacutation of sacutations of Potassium Chloride and believe in sacutations with the sacutation of sacutations with the sacutation of sacutations with the sacutation of sacutation of sacutations with the sacutation of sacutations with the sacutation of sacutations with the sacutation of sacutation of sacutation of sacutations with the sacutation of https://www.doi.org/10.1021/je800963g https://www.doi.org/10.1021/acs.jced.5b00941 Constants in Aqueous Solutions of Kertwisel Qersivet Maters from (5 to 90)

May Cativity Coefficients and Osmotic https://www.doi.org/10.1021/je101012n

https://www.doi.org/10.1021/je201012n

https://www.doi.org/10.1021/je2010544

Coefficients in Dilute Aqueous Sodium

https://www.doi.org/10.1016/j.ict.2014.03.001 https://www.doi.org/10.1016/j.jct.2014.03.001 https://www.doi.org/10.1021/acs.jced.8b00510 https://www.doi.org/10.1021/acs.jced.8b00590

**Enthalpies of Dilution of** https://www.doi.org/10.1021/je9004504 N-Glycylglycine in Aqueous Sodium N-Glycylglycine in Aqueous Sodium Candius (Inches) เลียง (Inches) (Inches) เลียง (Inches) https://www.doi.org/10.1007/s10765-006-0096-4 https://www.doi.org/10.1021/je060430q https://www.doi.org/10.1021/je900199j https://www.doi.org/10.1021/je700185m https://www.doi.org/10.1021/je050242k Solweilities5Denaities0and Refractive https://www.doi.org/10.1021/je200443t https://www.doi.org/10.1016/j.jct.2017.04.015 https://www.doi.org/10.1021/acs.jced.5b00992 https://www.doi.org/10.1021/acs.jced.5b00351 https://www.doi.org/10.1021/acs.jced.8b00087 https://www.doi.org/10.1016/j.jct.2006.06.014 https://www.doi.org/10.1021/je900260g Button and house the process of Aqueous and Electrolyte solubilities of Aqueous and Happing and Continuous and Electrolyte solubilities and Solution and Electrolyte solubilities and Solution and Electrolyte solubilities and Solution the Aqueous and Solution a The halbidy a invited self-association of Learnigh of the Lagrange of the Lagr Enthalpitypairwise self-association of https://www.doi.org/10.1016/j.jct.2015.11.028 https://www.doi.org/10.1016/j.fluid.2017.02.015 https://www.doi.org/10.1021/acs.jced.9b00490 https://www.doi.org/10.1016/j.jct.2014.05.019 https://www.doi.org/10.1016/j.tca.2007.03.007 Solubilities for Six Ternary Systems:
NaCl + NH4Cl + H2O, KCl + NH4Cl +
M20s warenent and qozoelakirin for l +
Epopolities of all imples all shorides in
Mittis Galante mill imples all shorides in
Mittis Galante million in million https://www.doi.org/10.1016/j.fluid.2004.07.019 https://www.doi.org/10.1016/j.tca.2016.03.033 https://www.doi.org/10.1021/acs.jced.8b01248 Equilibrium for the Water-Ethylene ଧାରୁ ପ୍ରେମ୍ପ ଅନ୍ୟୁକ୍ତ Glycol Diacetate http://webbook.nist.gov/cgi/cbook.cgi?ID=C7447407&Units=SI System: Solid-Liquid Equilibria in the Ternary https://www.doi.org/10.1021/acs.jced.5b00094 Systems NaCl SrCl2 H2O and KCl SrCl2 Partial modar volume and partial molar https://www.doi.org/10.1016/j.tca.2013.08.002 https://www.doi.org/10.1016/j.tca.2010.12.026 https://www.doi.org/10.1016/j.tca.2015.11.024 https://www.doi.org/10.1016/j.jct.2013.02.015 to the thermodynamics of aqueous https://www.doi.org/10.1021/je800855p Soluthility: Data in the System KCI + (2CrO4 + H2O K2CrO4 + m2O: Density and refractive index https://www.doi.org/10.1016/j.fluid.2010.02.018 https://www.doi.org/10.1007/s10765-010-0725-9 https://www.doi.org/10.1021/je2013704 https://www.doi.org/10.1016/j.tca.2016.06.008 (283 and 472) K, Pressures up to 68.5 MPa,

and Molalities up to 6 mol\*kg-1:

Solubility of Sodium Oxalate in Concentrated Electrolyte Solutions: Dortmund Data Bank Vapor-Liquid Equilibrium Data: Solubilities of six lithium salts in five non-aqueous solvents and in a few of their binary mixtures:

https://www.doi.org/10.1021/acs.jced.7b00690 http://www.ddbst.com/en/EED/VLE/VLE%20Water%3BPotassium chloride.php https://www.doi.org/10.1016/j.fluid.2017.12.034

#### Legend

ea: Electron affinityie: Ionization energypvap: Vapor pressurerhos: Solid Densitysrf: Surface Tension

tt: Triple Point Temperature

Latest version available from:

https://www.chemeo.com/cid/10-665-2/potassium-chloride.pdf

Generated by Cheméo on 2025-12-05 07:56:05.018715482 +0000 UTC m=+4669562.548756136.

Cheméo (https://www.chemeo.com) is the biggest free database of chemical and physical data for the process industry.