Multiflorenol (7-multiflorenenol) acetate

Inchi: InChI=1S/C32H52O2/c1-21(33)34-26-13-14-30(7)22-12-15-32(9)25-20-27(2,3)16-17-29(
InchiKey: BBIBQROZEQEFRD-UCYDDCCISA-N
Formula: C32H52O2
SMILES: CC(=O)OC1CCC2(C)C3CCC4(C)C5CC(C)(C)CCC5(C)CCC4(C)C3=CCC2C1(C)C
Mol. weight [g/mol]: 468.75

Physical Properties

Property code	Value	Unit	Source
gf	152.53	kJ/mol	Joback Method
hf	-591.68	kJ/mol	Joback Method
hfus	25.80	kJ/mol	Joback Method
hvap	89.25	kJ/mol	Joback Method
log10ws	-9.34		Crippen Method
logp	8.740		Crippen Method
mcvol	410.580	ml/mol	McGowan Method
pc	922.18	kPa	Joback Method
rinpol	3414.00		NIST Webbook
tb	1053.67	K	Joback Method
tc	1307.40	K	Joback Method
tf	723.10	K	Joback Method
vc	1.550	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1711.32	J/mol×K	1053.67	Joback Method
cpg	1781.21	J/mol×K	1095.96	Joback Method
cpg	1858.58	J/mol×K	1138.25	Joback Method
cpg	1944.46	J/mol×K	1180.54	Joback Method
cpg	2039.84	J/mol×K	1222.82	Joback Method
cpg	2145.73	J/mol×K	1265.11	Joback Method
cpg	2263.13	J/mol×K	1307.40	Joback Method

Sources

NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=R111369&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinpol:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume