

# 3-(4-Hydroxyphenyl)propionitrile

<b>Other names:</b>	«beta»-(4-Hydroxyphenyl)propionitrile Benzenepropanenitrile, 4-hydroxy- 3-(p-Hydroxyphenyl)propionitrile
<b>Inchi:</b>	InChI=1S/C9H9NO/c10-7-1-2-8-3-5-9(11)6-4-8/h3-6,11H,1-2H2
<b>InchiKey:</b>	KDMJGLYRWRHKJS-UHFFFAOYSA-N
<b>Formula:</b>	C9H9NO
<b>SMILES:</b>	N#CCCc1ccc(O)cc1
<b>Mol. weight [g/mol]:</b>	147.17
<b>CAS:</b>	17362-17-3

## Physical Properties

Property code	Value	Unit	Source
gf	115.87	kJ/mol	Joback Method
hf	-4.99	kJ/mol	Joback Method
hfus	20.40	kJ/mol	Joback Method
hvap	61.40	kJ/mol	Joback Method
log10ws	-2.11		Crippen Method
logp	1.848		Crippen Method
mcvol	121.160	ml/mol	McGowan Method
pc	3777.68	kPa	Joback Method
rinsol	1569.80		NIST Webbook
tb	614.70	K	Joback Method
tc	852.69	K	Joback Method
tf	394.32	K	Joback Method
vc	0.423	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	287.94	J/molxK	614.70	Joback Method
cpg	297.96	J/molxK	654.37	Joback Method
cpg	307.23	J/molxK	694.03	Joback Method
cpg	315.86	J/molxK	733.70	Joback Method
cpg	323.93	J/molxK	773.36	Joback Method

cpg	331.55	J/mol×K	813.03	Joback Method
cpg	338.79	J/mol×K	852.69	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C17362173&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C17362173&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvac:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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