

# Fumaric acid, butyl 3,4-dimethoxyphenyl ester

<b>Inchi:</b>	InChI=1S/C16H20O6/c1-4-5-10-21-15(17)8-9-16(18)22-12-6-7-13(19-2)14(11-12)20-3/h6
<b>InchiKey:</b>	PQYHMLFEDDSNHV-CMDGGGOBGSA-N
<b>Formula:</b>	C16H20O6
<b>SMILES:</b>	CCCCOC(=O)C=CC(=O)Oc1ccc(OC)c(OC)c1
<b>Mol. weight [g/mol]:</b>	308.33

## Physical Properties

Property code	Value	Unit	Source
gf	-420.63	kJ/mol	Joback Method
hf	-796.80	kJ/mol	Joback Method
hfus	38.61	kJ/mol	Joback Method
hvap	77.90	kJ/mol	Joback Method
log10ws	-3.25		Crippen Method
logp	2.509		Crippen Method
mcvol	234.860	ml/mol	McGowan Method
pc	1821.61	kPa	Joback Method
rinsol	2365.00		NIST Webbook
tb	803.70	K	Joback Method
tc	1010.69	K	Joback Method
tf	505.24	K	Joback Method
vc	0.887	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	677.41	J/molxK	803.70	Joback Method
cpg	735.66	J/molxK	976.19	Joback Method
cpg	726.11	J/molxK	941.70	Joback Method
cpg	715.51	J/molxK	907.20	Joback Method
cpg	703.85	J/molxK	872.70	Joback Method
cpg	691.14	J/molxK	838.20	Joback Method
cpg	744.15	J/molxK	1010.69	Joback Method
dvisc	0.0000474	Paxs	803.70	Joback Method
dvisc	0.0000592	Paxs	753.96	Joback Method

dvisc	0.0000762	Paxs	704.21	Joback Method
dvisc	0.0001021	Paxs	654.47	Joback Method
dvisc	0.0001434	Paxs	604.73	Joback Method
dvisc	0.0002141	Paxs	554.98	Joback Method
dvisc	0.0003460	Paxs	505.24	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U348166&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U348166&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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