

# Benzoic acid, 4-bromo-, methyl ester

<b>Other names:</b>	4-Bromobenzoic acid, methyl ester Benzoic acid, p-bromo-, methyl ester methyl 4-bromobenzoate methyl p-bromobenzoate p-Bromobenzoic acid, methyl ester
<b>Inchi:</b>	InChI=1S/C8H7BrO2/c1-11-8(10)6-2-4-7(9)5-3-6/h2-5H,1H3
<b>InchiKey:</b>	CZNGTXVOZOWWKM-UHFFFAOYSA-N
<b>Formula:</b>	C8H7BrO2
<b>SMILES:</b>	COC(=O)c1ccc(Br)cc1
<b>Mol. weight [g/mol]:</b>	215.04
<b>CAS:</b>	619-42-1

## Physical Properties

Property code	Value	Unit	Source
gf	-100.34	kJ/mol	Joback Method
hf	-201.86	kJ/mol	Joback Method
hfus	18.20	kJ/mol	Joback Method
hvap	51.93	kJ/mol	Joback Method
ie	9.47	eV	NIST Webbook
log10ws	-2.88		Crippen Method
logp	2.236		Crippen Method
mcvol	124.760	ml/mol	McGowan Method
pc	4135.61	kPa	Joback Method
rinpol	1336.00		NIST Webbook
rinpol	1336.00		NIST Webbook
rinpol	1336.00		NIST Webbook
tb	556.55	K	Joback Method
tc	792.86	K	Joback Method
tf	351.22	K	The influence of the halogen size in the volatility and melting of methyl p-halobenzoic esters and of their parent acids
vc	0.462	m3/kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	246.83	J/mol×K	556.55	Joback Method
cpg	257.12	J/mol×K	595.93	Joback Method
cpg	266.74	J/mol×K	635.32	Joback Method
cpg	275.70	J/mol×K	674.70	Joback Method
cpg	284.02	J/mol×K	714.09	Joback Method
cpg	291.72	J/mol×K	753.47	Joback Method
cpg	298.81	J/mol×K	792.86	Joback Method
dvisc	0.0015564	Paxs	350.82	Joback Method
dvisc	0.0010113	Paxs	385.11	Joback Method
dvisc	0.0007051	Paxs	419.40	Joback Method
dvisc	0.0005191	Paxs	453.69	Joback Method
dvisc	0.0003990	Paxs	487.97	Joback Method
dvisc	0.0003175	Paxs	522.26	Joback Method
dvisc	0.0002598	Paxs	556.55	Joback Method

## Sources

Crippen Method: [https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)

The influence of the halogen size in the volatility and melting of methyl halide Methyl esters and of their parent acids: <https://www.doi.org/10.1016/j.jct.2012.07.027>

Joback Method: [https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

McGowan Method: <http://link.springer.com/article/10.1007/BF02311772>

NIST Webbook: <http://webbook.nist.gov/cgi/cbook.cgi?ID=C619421&Units=SI>

Crippen Method: <http://pubs.acs.org/doi/abs/10.1021/ci990307l>

## Legend

cpg:	Ideal gas heat capacity
dvisc:	Dynamic viscosity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
ie:	Ionization energy

<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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