

Benzene, 1,2-dimethyl-4-(1-methylethyl)-

Other names:	1,2-Dimethyl-4-isopropylbenzene 4-isopropyl-1,2-dimethyl-benzene Benzene, 1,2-dimethyl-4-isopropyl
Inchi:	InChI=1S/C11H16/c1-8(2)11-6-5-9(3)10(4)7-11/h5-8H,1-4H3
InchiKey:	MGMSKQZIAGFMRU-UHFFFAOYSA-N
Formula:	C11H16
SMILES:	<chem>Cc1ccc(C(C)C)cc1C</chem>
Mol. weight [g/mol]:	148.24
CAS:	4132-77-8

Physical Properties

Property code	Value	Unit	Source
gf	132.45	kJ/mol	Joback Method
hf	-62.06	kJ/mol	Joback Method
hfus	13.99	kJ/mol	Joback Method
hvap	43.29	kJ/mol	Joback Method
log10ws	-3.61		Crippen Method
logp	3.427		Crippen Method
mcvol	142.090	ml/mol	McGowan Method
pc	2589.85	kPa	Joback Method
ripol	1130.00		NIST Webbook
ripol	1130.00		NIST Webbook
ripol	1120.00		NIST Webbook
ripol	1140.00		NIST Webbook
ripol	1140.00		NIST Webbook
ripol	1139.00		NIST Webbook
ripol	1117.50		NIST Webbook
ripol	1130.00		NIST Webbook
ripol	1130.00		NIST Webbook
ripol	1130.00		NIST Webbook
ripol	1372.00		NIST Webbook
ripol	1372.00		NIST Webbook
ripol	1399.00		NIST Webbook
ripol	1450.00		NIST Webbook
ripol	1435.00		NIST Webbook
ripol	1423.00		NIST Webbook
ripol	1412.00		NIST Webbook

ripol	1411.00		NIST Webbook
ripol	1372.50		NIST Webbook
tb	473.00 ± 5.00	K	NIST Webbook
tc	696.41	K	Joback Method
tf	250.19	K	Joback Method
vc	0.537	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	301.65	J/mol×K	487.28	Joback Method
cpg	317.34	J/mol×K	522.13	Joback Method
cpg	332.23	J/mol×K	556.99	Joback Method
cpg	346.36	J/mol×K	591.84	Joback Method
cpg	359.74	J/mol×K	626.70	Joback Method
cpg	372.40	J/mol×K	661.55	Joback Method
cpg	384.36	J/mol×K	696.41	Joback Method
dvisc	0.0024629	Paxs	250.19	Joback Method
dvisc	0.0012026	Paxs	289.70	Joback Method
dvisc	0.0006975	Paxs	329.22	Joback Method
dvisc	0.0004546	Paxs	368.74	Joback Method
dvisc	0.0003219	Paxs	408.25	Joback Method
dvisc	0.0002423	Paxs	447.76	Joback Method
dvisc	0.0001909	Paxs	487.28	Joback Method

Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.42585e+01
Coeff. B	-3.87977e+03
Coeff. C	-7.24920e+01
Temperature range (K), min.	350.20
Temperature range (K), max.	506.13

Sources

McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C4132778&Units=SI
The Yaws Handbook of Vapor Pressure:	https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307I
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method

Legend

cp_g:	Ideal gas heat capacity
d_{visc}:	Dynamic viscosity
g_f:	Standard Gibbs free energy of formation
h_f:	Enthalpy of formation at standard conditions
h_{fus}:	Enthalpy of fusion at standard conditions
h_{vap}:	Enthalpy of vaporization at standard conditions
log₁₀ws:	Log ₁₀ of Water solubility in mol/l
log_p:	Octanol/Water partition coefficient
mc_{vol}:	McGowan's characteristic volume
pc:	Critical Pressure
pv_{ap}:	Vapor pressure
ri_{npol}:	Non-polar retention indices
ri_{pol}:	Polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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