

Propane, 1,3-dimethoxy-

Other names:	Trimethylene glycol dimethyl ether 1,3-Dimethoxypropane CH3O(CH2)3OCH3
Inchi:	InChI=1S/C5H12O2/c1-6-4-3-5-7-2/h3-5H2,1-2H3
InchiKey:	UUAMLBIYJDPGFU-UHFFFAOYSA-N
Formula:	C5H12O2
SMILES:	COCCCO
Mol. weight [g/mol]:	104.15
CAS:	17081-21-9

Physical Properties

Property code	Value	Unit	Source
affp	897.20	kJ/mol	NIST Webbook
basg	858.60	kJ/mol	NIST Webbook
gf	-218.78	kJ/mol	Joback Method
hf	-410.97	kJ/mol	Joback Method
hfus	11.08	kJ/mol	Joback Method
hvap	31.54	kJ/mol	Joback Method
ie	9.30	eV	NIST Webbook
ie	9.68	eV	NIST Webbook
log10ws	-0.09		Crippen Method
logp	0.669		Crippen Method
mcvol	93.050	ml/mol	McGowan Method
pc	3333.53	kPa	Joback Method
rinpol	697.30		NIST Webbook
rinpol	700.00		NIST Webbook
tb	358.64	K	Joback Method
tc	524.09	K	Joback Method
tf	190.57	K	Joback Method
vc	0.351	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
---------------	-------	------	-----------------	--------

cpg	169.54	J/mol×K	358.64	Joback Method
cpg	210.37	J/mol×K	496.51	Joback Method
cpg	202.53	J/mol×K	468.94	Joback Method
cpg	194.52	J/mol×K	441.36	Joback Method
cpg	186.35	J/mol×K	413.79	Joback Method
cpg	178.02	J/mol×K	386.21	Joback Method
cpg	218.03	J/mol×K	524.09	Joback Method
dvisc	0.0002027	Pa×s	358.64	Joback Method
dvisc	0.0002582	Pa×s	330.63	Joback Method
dvisc	0.0003441	Pa×s	302.62	Joback Method
dvisc	0.0004861	Pa×s	274.61	Joback Method
dvisc	0.0007430	Pa×s	246.59	Joback Method
dvisc	0.0012659	Pa×s	218.58	Joback Method
dvisc	0.0025228	Pa×s	190.57	Joback Method

Sources

McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C17081219&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method

Legend

affp:	Proton affinity
basg:	Gas basicity
cpg:	Ideal gas heat capacity
dvisc:	Dynamic viscosity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
ie:	Ionization energy
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure

rinpol:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/10-880-3/Propane-1-3-dimethoxy.pdf>

Generated by Cheméo on 2025-12-21 04:51:53.754353916 +0000 UTC m=+6040911.284394570.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.