

# Di-(2-o-chlorophenyl-ethylene)-ketone

<b>Inchi:</b>	InChI=1S/C17H12Cl2O/c18-16-7-3-1-5-13(16)9-11-15(20)12-10-14-6-2-4-8-17(14)19/h1-
<b>InchiKey:</b>	OZBIBYIDJNVAOQ-WGDLNXRISA-N
<b>Formula:</b>	C17H12Cl2O
<b>SMILES:</b>	O=C(C=Cc1ccccc1Cl)C=Cc1ccccc1Cl
<b>Mol. weight [g/mol]:</b>	303.18
<b>CAS:</b>	5332-98-9

## Physical Properties

Property code	Value	Unit	Source
gf	305.48	kJ/mol	Joback Method
hf	146.29	kJ/mol	Joback Method
hfus	37.49	kJ/mol	Joback Method
hvap	74.74	kJ/mol	Joback Method
log10ws	-5.84		Crippen Method
logp	5.289		Crippen Method
mcvol	220.320	ml/mol	McGowan Method
pc	2214.53	kPa	Joback Method
tb	788.73	K	Joback Method
tc	1045.01	K	Joback Method
tf	458.84	K	Joback Method
vc	0.836	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	543.01	J/molxK	788.73	Joback Method
cpg	555.80	J/molxK	831.44	Joback Method
cpg	567.58	J/molxK	874.16	Joback Method
cpg	578.47	J/molxK	916.87	Joback Method
cpg	588.61	J/molxK	959.59	Joback Method
cpg	598.14	J/molxK	1002.30	Joback Method
cpg	607.20	J/molxK	1045.01	Joback Method
dvisc	0.0007971	Paxs	458.84	Joback Method
dvisc	0.0004537	Paxs	513.82	Joback Method

dvisc	0.0002879	Paxs	568.80	Joback Method
dvisc	0.0001980	Paxs	623.79	Joback Method
dvisc	0.0001447	Paxs	678.77	Joback Method
dvisc	0.0001108	Paxs	733.75	Joback Method
dvisc	0.0000881	Paxs	788.73	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C5332989&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C5332989&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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