

Methanediol, dinitrate

Inchi:

InChI=1S/CH2N2O6/c4-2(5)8-1-9-3(6)7/h1H2

InchiKey:

MKQRTTJKPCAWRP-UHFFFAOYSA-N

Formula:

CH2N2O6

SMILES:

O=[N+]([O-])OCO[N+](=O)[O-]

Mol. weight [g/mol]:

138.04

CAS:

38483-28-2

Physical Properties

Property code	Value	Unit	Source
chl	-575.70	kJ/mol	NIST Webbook
gf	-181.36	kJ/mol	Joback Method
hf	-349.93	kJ/mol	Joback Method
hfl	93.30	kJ/mol	NIST Webbook
hfus	23.44	kJ/mol	Joback Method
hvap	55.82	kJ/mol	Joback Method
log10ws	-1.06		Crippen Method
logp	-0.640		Crippen Method
mcvol	71.530	ml/mol	McGowan Method
pc	5704.58	kPa	Joback Method
tb	570.80	K	Joback Method
tc	816.09	K	Joback Method
tf	432.71	K	Joback Method
vc	0.291	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	160.02	J/mol×K	570.80	Joback Method
cpg	165.72	J/mol×K	611.68	Joback Method
cpg	171.10	J/mol×K	652.56	Joback Method
cpg	176.15	J/mol×K	693.45	Joback Method
cpg	180.84	J/mol×K	734.33	Joback Method
cpg	185.12	J/mol×K	775.21	Joback Method
cpg	188.99	J/mol×K	816.09	Joback Method

Sources

NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C38483282&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307I
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772

Legend

chl:	Standard liquid enthalpy of combustion
cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfl:	Liquid phase enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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