

# 3-Methyl-hex-2-enedioic acid dimethyl ester, E

<b>Inchi:</b>	InChI=1S/C7H10O4/c1-11-7(10)5-3-2-4-6(8)9/h2,4H,3,5H2,1H3,(H,8,9)/b4-2+
<b>InchiKey:</b>	HBZOSRCVWPAGOA-DUXPYHPUSA-N
<b>Formula:</b>	C7H10O4
<b>SMILES:</b>	COC(=O)CCC=CC(=O)O
<b>Mol. weight [g/mol]:</b>	158.15

## Physical Properties

Property code	Value	Unit	Source
gf	-411.38	kJ/mol	Joback Method
hf	-580.20	kJ/mol	Joback Method
hfus	22.56	kJ/mol	Joback Method
hvap	63.72	kJ/mol	Joback Method
log10ws	-0.57		Crippen Method
logp	0.580		Crippen Method
mcvol	120.070	ml/mol	McGowan Method
pc	3731.66	kPa	Joback Method
rinpol	1287.00		NIST Webbook
tb	586.06	K	Joback Method
tc	770.92	K	Joback Method
tf	346.48	K	Joback Method
vc	0.457	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	282.10	J/molxK	586.06	Joback Method
cpg	290.62	J/molxK	616.87	Joback Method
cpg	298.72	J/molxK	647.68	Joback Method
cpg	306.42	J/molxK	678.49	Joback Method
cpg	313.73	J/molxK	709.30	Joback Method
cpg	320.65	J/molxK	740.11	Joback Method
cpg	327.20	J/molxK	770.92	Joback Method
dvisc	0.0044440	Paxs	346.48	Joback Method
dvisc	0.0016709	Paxs	386.41	Joback Method

dvisc	0.0007546	Paxs	426.34	Joback Method
dvisc	0.0003905	Paxs	466.27	Joback Method
dvisc	0.0002242	Paxs	506.20	Joback Method
dvisc	0.0001396	Paxs	546.13	Joback Method
dvisc	0.0000927	Paxs	586.06	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R249320&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R249320&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/10-974-9/3-Methyl-hex-2-enedioic-acid-dimethyl-ester-E.pdf>

Generated by Cheméo on 2024-04-25 14:58:32.466770283 +0000 UTC m=+16346361.387347595.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.