

# (3-Iodophenyl) methanol, n-butyl ether

<b>Inchi:</b>	InChI=1S/C11H15IO/c1-2-3-7-13-9-10-5-4-6-11(12)8-10/h4-6,8H,2-3,7,9H2,1H3
<b>InchiKey:</b>	HRJJHUAJCDKJJS-UHFFFAOYSA-N
<b>Formula:</b>	C11H15IO
<b>SMILES:</b>	CCCCOCc1cccc(I)c1
<b>Mol. weight [g/mol]:</b>	290.14

## Physical Properties

Property code	Value	Unit	Source
gf	97.64	kJ/mol	Joback Method
hf	-100.66	kJ/mol	Joback Method
hfus	23.49	kJ/mol	Joback Method
hvap	54.80	kJ/mol	Joback Method
log10ws	-4.26		Crippen Method
logp	3.608		Crippen Method
mcvol	173.780	ml/mol	McGowan Method
pc	2507.52	kPa	Joback Method
rinpol	1657.00		NIST Webbook
rinpol	1657.00		NIST Webbook
tb	598.30	K	Joback Method
tc	828.60	K	Joback Method
tf	332.96	K	Joback Method
vc	0.649	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	381.01	J/molxK	598.30	Joback Method
cpg	395.52	J/molxK	636.68	Joback Method
cpg	409.12	J/molxK	675.07	Joback Method
cpg	421.86	J/molxK	713.45	Joback Method
cpg	433.75	J/molxK	751.83	Joback Method
cpg	444.85	J/molxK	790.21	Joback Method
cpg	455.18	J/molxK	828.60	Joback Method
dvisc	0.0019953	Paxs	332.96	Joback Method

dvisc	0.0010564	Paxs	377.18	Joback Method
dvisc	0.0006392	Paxs	421.41	Joback Method
dvisc	0.0004255	Paxs	465.63	Joback Method
dvisc	0.0003039	Paxs	509.85	Joback Method
dvisc	0.0002291	Paxs	554.08	Joback Method
dvisc	0.0001800	Paxs	598.30	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U374574&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U374574&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/10-991-0/3-Iodophenyl-methanol-n-butyl-ether.pdf>

Generated by Cheméo on 2024-04-23 07:40:25.272128134 +0000 UTC m=+16147274.192705444.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.