

# 5-ethyl-5-methyl-1,3-diazinane-2,4,6-trione

<b>Other names:</b>	5-methyl-5-ethylbarbituric acid
<b>Inchi:</b>	InChI=1S/C7H10N2O3/c1-3-7(2)4(10)8-6(12)9-5(7)11/h3H2,1-2H3,(H2,8,9,10,11,12)
<b>InchiKey:</b>	RCFRTWDBBMGLGK-UHFFFAOYSA-N
<b>Formula:</b>	C7H10N2O3
<b>SMILES:</b>	CCC1(C)C(=O)NC(=O)NC1=O
<b>Mol. weight [g/mol]:</b>	170.17

## Physical Properties

Property code	Value	Unit	Source
gf	-165.33	kJ/mol	Joback Method
hf	-455.73	kJ/mol	Joback Method
hfus	17.13	kJ/mol	Joback Method
hvap	56.71	kJ/mol	Joback Method
log10ws	-1.16		Aqueous and cosolvent solubility data for drug-like organic compounds
log10ws	-1.23		Estimated Solubility Method
log10ws	-1.23		Aqueous Solubility Prediction Method
logp	-0.231		Crippen Method
mcvol	123.300	ml/mol	McGowan Method
pc	4534.68	kPa	Joback Method
tb	679.91	K	Joback Method
tc	951.71	K	Joback Method
tf	614.65	K	Joback Method
vc	0.454	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	336.63	J/mol×K	679.91	Joback Method
cpg	352.44	J/mol×K	725.21	Joback Method
cpg	367.54	J/mol×K	770.51	Joback Method
cpg	381.92	J/mol×K	815.81	Joback Method

cpg	395.57	J/mol×K	861.11	Joback Method
cpg	408.48	J/mol×K	906.41	Joback Method
cpg	420.62	J/mol×K	951.71	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>Aqueous Solubility Prediction Method:</b>	<a href="http://onschallenge.wikispaces.com/file/view/AqueousDataset002.xlsx/351826032/AqueousDa">http://onschallenge.wikispaces.com/file/view/AqueousDataset002.xlsx/351826032/AqueousDa</a>
<b>Estimated Solubility Method:</b>	<a href="http://pubs.acs.org/doi/suppl/10.1021/ci034243x/suppl_file/ci034243xsi20040112_053635.txt">http://pubs.acs.org/doi/suppl/10.1021/ci034243x/suppl_file/ci034243xsi20040112_053635.txt</a>
<b>Aqueous and cosolvent solubility data for drug-like organic compounds:</b>	<a href="https://www.ncbi.nlm.nih.gov/pmc/articles/PMC2751500/">https://www.ncbi.nlm.nih.gov/pmc/articles/PMC2751500/</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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