

2,4-diazaspiro[5.5]undecane-1,3,5-trione

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|-----------------------------|---|
| Other names: | cyclohexyl-5-spirobarbituric acid |
| Inchi: | InChI=1S/C9H12N2O3/c12-6-9(4-2-1-3-5-9)7(13)11-8(14)10-6/h1-5H2,(H2,10,11,12,13,14) |
| InchiKey: | NBLIVIWUPTWDMD-UHFFFAOYSA-N |
| Formula: | C9H12N2O3 |
| SMILES: | O=C1NC(=O)C2(CCCCC2)C(=O)N1 |
| Mol. weight [g/mol]: | 196.21 |

Physical Properties

| Property code | Value | Unit | Source |
|---------------|---------|----------------------|---|
| gf | -104.23 | kJ/mol | Joback Method |
| hf | -416.19 | kJ/mol | Joback Method |
| hfus | 15.18 | kJ/mol | Joback Method |
| hvap | 61.73 | kJ/mol | Joback Method |
| log10ws | -3.06 | | Estimated Solubility Method |
| log10ws | -3.06 | | Aqueous and cosolvent solubility data for drug-like organic compounds |
| log10ws | -3.06 | | Aqueous Solubility Prediction Method |
| logp | 0.303 | | Crippen Method |
| mcvol | 140.620 | ml/mol | McGowan Method |
| pc | 4665.71 | kPa | Joback Method |
| tb | 745.62 | K | Joback Method |
| tc | 1041.00 | K | Joback Method |
| tf | 652.33 | K | Joback Method |
| vc | 0.507 | m ³ /kmol | Joback Method |

Temperature Dependent Properties

| Property code | Value | Unit | Temperature [K] | Source |
|---------------|--------|---------|-----------------|---------------|
| cpg | 426.50 | J/mol×K | 745.62 | Joback Method |
| cpg | 445.57 | J/mol×K | 794.85 | Joback Method |
| cpg | 463.43 | J/mol×K | 844.08 | Joback Method |
| cpg | 480.12 | J/mol×K | 893.31 | Joback Method |

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|-----|--------|---------|---------|---------------|
| cpg | 495.70 | J/mol×K | 942.54 | Joback Method |
| cpg | 510.20 | J/mol×K | 991.77 | Joback Method |
| cpg | 523.67 | J/mol×K | 1041.00 | Joback Method |

Sources

| | |
|---|---|
| Joback Method: | https://en.wikipedia.org/wiki/Joback_method |
| Aqueous Solubility Prediction Method: | http://onschallenge.wikispaces.com/file/view/AqueousDataset002.xlsx/351826032/AqueousDataset002.xlsx |
| Estimated Solubility Method: | http://pubs.acs.org/doi/suppl/10.1021/ci034243x/suppl_file/ci034243xsi20040112_053635.txt |
| Aqueous and cosolvent solubility data for drug-like organic compounds: McGowan Method: | https://www.ncbi.nlm.nih.gov/pmc/articles/PMC2751500/ |
| Crippen Method: | http://link.springer.com/article/10.1007/BF02311772 |
| | http://pubs.acs.org/doi/abs/10.1021/ci9903071 |

Legend

| | |
|-----------------|---|
| cpg: | Ideal gas heat capacity |
| gf: | Standard Gibbs free energy of formation |
| hf: | Enthalpy of formation at standard conditions |
| hfus: | Enthalpy of fusion at standard conditions |
| hvap: | Enthalpy of vaporization at standard conditions |
| log10ws: | Log10 of Water solubility in mol/l |
| logp: | Octanol/Water partition coefficient |
| mcvol: | McGowan's characteristic volume |
| pc: | Critical Pressure |
| tb: | Normal Boiling Point Temperature |
| tc: | Critical Temperature |
| tf: | Normal melting (fusion) point |
| vc: | Critical Volume |

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