

anthranilamide

Inchi: InChI=1S/C7H8N2O/c8-6-4-2-1-3-5(6)7(9)10/h1-4H,8H2,(H2,9,10)
InchiKey: PXBFMLJZNCDSMP-UHFFFAOYSA-N
Formula: C7H8N2O
SMILES: NC(=O)c1ccccc1N
Mol. weight [g/mol]: 136.15

Physical Properties

Property code	Value	Unit	Source
gf	114.82	kJ/mol	Joback Method
hf	-7.75	kJ/mol	Joback Method
hfus	21.90	kJ/mol	Solubility Measurement and Phase Equilibrium Modeling of 2-Aminobenzamide in 15 Pure Solvents
hfus	20.90	kJ/mol	Experimental and computational thermodynamic study of ortho- meta- and para-aminobenzamide
hvap	62.14	kJ/mol	Joback Method
log10ws	-1.28		Crippen Method
logp	0.368		Crippen Method
mcvol	107.260	ml/mol	McGowan Method
pc	5123.98	kPa	Joback Method
tb	590.15	K	Joback Method
tc	837.19	K	Joback Method
tf	424.04	K	Joback Method
tt	384.40	K	Experimental and computational thermodynamic study of ortho- meta- and para-aminobenzamide
tt	383.65	K	Solubility Measurement and Phase Equilibrium Modeling of 2-Aminobenzamide in 15 Pure Solvents
vc	0.384	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	248.85	J/molxK	590.15	Joback Method
cpg	259.00	J/molxK	631.32	Joback Method
cpg	268.39	J/molxK	672.50	Joback Method
cpg	277.04	J/molxK	713.67	Joback Method
cpg	284.99	J/molxK	754.84	Joback Method
cpg	292.28	J/molxK	796.02	Joback Method
cpg	298.94	J/molxK	837.19	Joback Method
psub	1.50e-04	kPa	339.17	Experimental and computational thermodynamic study of ortho-meta- and para-aminobenzamide
psub	1.21e-04	kPa	337.17	Experimental and computational thermodynamic study of ortho-meta- and para-aminobenzamide
psub	1.52e-04	kPa	339.17	Experimental and computational thermodynamic study of ortho-meta- and para-aminobenzamide
psub	1.87e-04	kPa	341.10	Experimental and computational thermodynamic study of ortho-meta- and para-aminobenzamide
psub	2.36e-04	kPa	343.16	Experimental and computational thermodynamic study of ortho-meta- and para-aminobenzamide
psub	2.98e-04	kPa	345.16	Experimental and computational thermodynamic study of ortho-meta- and para-aminobenzamide
psub	3.59e-04	kPa	347.11	Experimental and computational thermodynamic study of ortho-meta- and para-aminobenzamide

psub	4.41e-04	kPa	349.16	Experimental and computational thermodynamic study of ortho-meta- and para-aminobenzamide
psub	5.45e-04	kPa	351.17	Experimental and computational thermodynamic study of ortho-meta- and para-aminobenzamide
psub	6.67e-04	kPa	353.12	Experimental and computational thermodynamic study of ortho-meta- and para-aminobenzamide
psub	8.20e-04	kPa	355.16	Experimental and computational thermodynamic study of ortho-meta- and para-aminobenzamide
psub	9.93e-04	kPa	357.16	Experimental and computational thermodynamic study of ortho-meta- and para-aminobenzamide
psub	9.60e-05	kPa	335.11	Experimental and computational thermodynamic study of ortho-meta- and para-aminobenzamide
psub	1.21e-04	kPa	337.17	Experimental and computational thermodynamic study of ortho-meta- and para-aminobenzamide
psub	9.50e-05	kPa	335.11	Experimental and computational thermodynamic study of ortho-meta- and para-aminobenzamide
psub	1.90e-04	kPa	341.10	Experimental and computational thermodynamic study of ortho-meta- and para-aminobenzamide
psub	2.37e-04	kPa	343.16	Experimental and computational thermodynamic study of ortho-meta- and para-aminobenzamide

psub	2.90e-04	kPa	345.16	Experimental and computational thermodynamic study of ortho-meta- and para-aminobenzamide
psub	3.59e-04	kPa	347.11	Experimental and computational thermodynamic study of ortho-meta- and para-aminobenzamide
psub	4.45e-04	kPa	349.16	Experimental and computational thermodynamic study of ortho-meta- and para-aminobenzamide
psub	5.39e-04	kPa	351.17	Experimental and computational thermodynamic study of ortho-meta- and para-aminobenzamide
psub	6.67e-04	kPa	353.12	Experimental and computational thermodynamic study of ortho-meta- and para-aminobenzamide
psub	8.26e-04	kPa	355.16	Experimental and computational thermodynamic study of ortho-meta- and para-aminobenzamide
psub	9.94e-04	kPa	357.16	Experimental and computational thermodynamic study of ortho-meta- and para-aminobenzamide
psub	9.50e-05	kPa	335.11	Experimental and computational thermodynamic study of ortho-meta- and para-aminobenzamide
psub	1.20e-04	kPa	337.17	Experimental and computational thermodynamic study of ortho-meta- and para-aminobenzamide
psub	1.49e-04	kPa	339.17	Experimental and computational thermodynamic study of ortho-meta- and para-aminobenzamide

psub	1.88e-04	kPa	341.10	Experimental and computational thermodynamic study of ortho-meta- and para-aminobenzamide
psub	2.32e-04	kPa	343.16	Experimental and computational thermodynamic study of ortho-meta- and para-aminobenzamide
psub	2.85e-04	kPa	345.16	Experimental and computational thermodynamic study of ortho-meta- and para-aminobenzamide
psub	3.57e-04	kPa	347.11	Experimental and computational thermodynamic study of ortho-meta- and para-aminobenzamide
psub	4.35e-04	kPa	349.16	Experimental and computational thermodynamic study of ortho-meta- and para-aminobenzamide
psub	5.30e-04	kPa	351.17	Experimental and computational thermodynamic study of ortho-meta- and para-aminobenzamide
psub	6.63e-04	kPa	353.12	Experimental and computational thermodynamic study of ortho-meta- and para-aminobenzamide
psub	8.06e-04	kPa	355.16	Experimental and computational thermodynamic study of ortho-meta- and para-aminobenzamide
psub	9.93e-04	kPa	357.16	Experimental and computational thermodynamic study of ortho-meta- and para-aminobenzamide

Sources

Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci9903071
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Experimental and computational thermodynamic study of ortho- meta- Solubility Measurement and Phase Equilibrium Modeling of Joback Method in 15 Pure Solvents:	https://www.doi.org/10.1016/j.jct.2012.12.006
McGowan Method:	https://www.doi.org/10.1021/acs.jced.9b00350
	https://en.wikipedia.org/wiki/Joback_method
	http://link.springer.com/article/10.1007/BF02311772

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
psub:	Sublimation pressure
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
tt:	Triple Point Temperature
vc:	Critical Volume

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