

[2-[(2-amino-2-oxoethyl)amino]-2-oxoethyl] benzoate

Inchi:	InChI=1S/C11H12N2O4/c12-9(14)6-13-10(15)7-17-11(16)8-4-2-1-3-5-8/h1-5H,6-7H2,(H2)
InchiKey:	XYGUXAGVJDBSBP-UHFFFAOYSA-N
Formula:	C11H12N2O4
SMILES:	NC(=O)CNC(=O)COC(=O)c1ccccc1
Mol. weight [g/mol]:	236.23

Physical Properties

Property code	Value	Unit	Source
gf	-181.77	kJ/mol	Joback Method
hf	-416.54	kJ/mol	Joback Method
hfus	34.57	kJ/mol	Joback Method
hvap	82.08	kJ/mol	Joback Method
log10ws	-1.50		Aqueous Solubility Prediction Method
logp	-0.555		Crippen Method
mcvol	172.630	ml/mol	McGowan Method
pc	3456.14	kPa	Joback Method
tb	784.49	K	Joback Method
tc	1012.42	K	Joback Method
tf	424.65	K	Aqueous Solubility Prediction Method
vc	0.643	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	479.91	J/molxK	784.49	Joback Method
cpg	490.25	J/molxK	822.48	Joback Method
cpg	499.66	J/molxK	860.47	Joback Method
cpg	508.17	J/molxK	898.46	Joback Method
cpg	515.81	J/molxK	936.45	Joback Method
cpg	522.61	J/molxK	974.43	Joback Method
cpg	528.59	J/molxK	1012.42	Joback Method

Sources

Joback Method:	https://en.wikipedia.org/wiki/Joback_method
Aqueous Solubility Prediction Method:	http://onschallenge.wikispaces.com/file/view/AqueousDataset002.xlsx/351826032/AqueousDataset002.xlsx
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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