

# Benzene, 1-fluoro-2,4-dinitro-

<b>Other names:</b>	Dinitrofluorobenzene DFB DNFB Fluoro-2,4-dinitrobenzene Fluorodinitrobenzene FDNB 1-Fluoro-2,4-dinitrobenzene 2,4-Dinitro-1-fluorobenzene 2,4-Dinitrobenzene fluoride 2,4-Dinitrofluorobenzene 2,4-Dinitrophenyl fluoride Benzene, 2,4-dinitro-1-fluoro- 1,2,4-Fluorodinitrobenzene 2,4-DNFB NSC 33519
<b>Inchi:</b>	InChI=1S/C6H3FN2O4/c7-5-2-1-4(8(10)11)3-6(5)9(12)13/h1-3H
<b>InchiKey:</b>	LOTKRQAVGJMPNV-UHFFFAOYSA-N
<b>Formula:</b>	C6H3FN2O4
<b>SMILES:</b>	O=[N+]([O-])c1ccc(F)c([N+](=O)[O-])c1
<b>Mol. weight [g/mol]:</b>	186.10
<b>CAS:</b>	70-34-8

## Physical Properties

Property code	Value	Unit	Source
chs	-2800.10	kJ/mol	NIST Webbook
gf	-30.92	kJ/mol	Joback Method
hf	-171.21	kJ/mol	Joback Method
hfus	30.36	kJ/mol	Joback Method
hvap	64.91	kJ/mol	Joback Method
log10ws	-3.10		Crippen Method
logp	1.642		Crippen Method
mcvol	108.250	ml/mol	McGowan Method
pc	4362.63	kPa	Joback Method
tb	676.27	K	Joback Method
tc	943.68	K	Joback Method
tf	496.65	K	Joback Method
vc	0.446	m3/kmol	Joback Method

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## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	260.31	J/mol×K	676.27	Joback Method
cpg	268.52	J/mol×K	720.84	Joback Method
cpg	275.94	J/mol×K	765.41	Joback Method
cpg	282.62	J/mol×K	809.97	Joback Method
cpg	288.61	J/mol×K	854.54	Joback Method
cpg	293.93	J/mol×K	899.11	Joback Method
cpg	298.65	J/mol×K	943.68	Joback Method

## Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	451.20	K	3.30	NIST Webbook

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C70348&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C70348&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>chs:</b>	Standard solid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions

<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>w<sub>s</sub>:</b>	Log10 of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>mc<sub>vol</sub>:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>t<sub>brp</sub>:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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