

# DCPA

**Other names:**

1,4-Benzenedicarboxylic acid, 2,3,5,6-tetrachloro-, dimethyl ester  
Terephthalic acid, tetrachloro-, dimethyl ester  
Chlorthal-methyl  
Dacthal  
Dimethyl tetrachloroterephthalate  
Dimethyl 2,3,5,6-tetrachloroterephthalate  
DAC 4  
DAC-893  
Tetrachloroterephthalic acid dimethyl ester  
Chlorothal-methyl  
Dacthalor  
Dimethyl 2,3,5,6-tetrachloro-1,4-benzenedicarboxylate  
Fatal  
2,3,5,6-Tetrachloroterephthalic acid, dimethyl ester  
2,3,5,6-Tetrachlorphthalsaure-dimethylester  
2,3,5,6-Tetrachloro-1,4-benzenedicarboxylic acid dimethyl ester  
2,3,5,6-Tetrachloro-1,4-benzenedicarboxylic acid dimethyl ester  
Chlorthal-dimethyl  
Dimethyl ester of tetrachloroterephthalic acid  
Terechloroterephthalic acid dimethyl ester  
Terephthalic acid, 2,3,5,6-tetrachloro-,dimethyl ester  
Vegetable turf and ornamental weeder  
Chlorthal dimethyl ester  
NSC 155745  
TCTP  
Tetral

**Inchi:** InChI=1S/C10H6Cl4O4/c1-17-9(15)3-5(11)7(13)4(10(16)18-2)8(14)6(3)12/h1-2H3  
**InchiKey:** NPOJQCVWMSKXDN-UHFFFAOYSA-N  
**Formula:** C10H6Cl4O4  
**SMILES:** COC(=O)c1c(Cl)c(Cl)c(C(=O)OC)c(Cl)c1Cl  
**Mol. weight [g/mol]:** 331.96  
**CAS:** 1861-32-1

## Physical Properties

Property code	Value	Unit	Source
gf	-417.98	kJ/mol	Joback Method

hf	-623.11		kJ/mol	Joback Method
h <sub>fus</sub>	36.11		kJ/mol	Joback Method
h <sub>vap</sub>	79.29		kJ/mol	Joback Method
ie	9.57		eV	NIST Webbook
log <sub>10</sub> w <sub>s</sub>	-4.70			Crippen Method
log <sub>p</sub>	3.873			Crippen Method
m <sub>cvol</sub>	191.840		ml/mol	McGowan Method
pc	2555.92		kPa	Joback Method
tb	782.08		K	Joback Method
tc	1016.00		K	Joback Method
tf	431.87 ± 0.20		K	NIST Webbook
tf	431.30 ± 0.20		K	NIST Webbook
vc	0.732		m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
c <sub>pg</sub>	415.76	J/mol×K	782.08	Joback Method
c <sub>pg</sub>	423.59	J/mol×K	821.07	Joback Method
c <sub>pg</sub>	430.69	J/mol×K	860.05	Joback Method
c <sub>pg</sub>	437.02	J/mol×K	899.04	Joback Method
c <sub>pg</sub>	442.55	J/mol×K	938.03	Joback Method
c <sub>pg</sub>	447.25	J/mol×K	977.02	Joback Method
c <sub>pg</sub>	451.11	J/mol×K	1016.00	Joback Method
d <sub>visc</sub>	0.0003493	Paxs	593.25	Joback Method
d <sub>visc</sub>	0.0004626	Paxs	555.48	Joback Method
d <sub>visc</sub>	0.0002728	Paxs	631.01	Joback Method
d <sub>visc</sub>	0.0002191	Paxs	668.78	Joback Method
d <sub>visc</sub>	0.0001801	Paxs	706.55	Joback Method
d <sub>visc</sub>	0.0001510	Paxs	744.31	Joback Method
d <sub>visc</sub>	0.0001288	Paxs	782.08	Joback Method
h <sub>fust</sub>	30.23	kJ/mol	431.70	NIST Webbook
h <sub>fust</sub>	30.23	kJ/mol	431.70	NIST Webbook
h <sub>subt</sub>	104.90 ± 1.40	kJ/mol	390.50	NIST Webbook

## Sources

McGowan Method:

<http://link.springer.com/article/10.1007/BF02311772>

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C1861321&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C1861321&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hfust:</b>	Enthalpy of fusion at a given temperature
<b>hsub:</b>	Enthalpy of sublimation at a given temperature
<b>hvac:</b>	Enthalpy of vaporization at standard conditions
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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