# 4-amino-N-(4-cyanophenyl)benzenesulfonamide

Inchi:	InChI=1S/C13H11N3O2S/c14-9-10-1-5-12(6-2-10)16-19(17,18)13-7-3-11(15)4-8-13/h1-8
InchiKey:	PLPNDNWDSGFETI-UHFFFAOYSA-N
Formula:	C13H11N3O2S
SMILES:	N#Cc1ccc(NS(=O)(=O)c2ccc(N)cc2)cc1
Mol. weight [g/mol]:	273.32

# **Physical Properties**

Property code	Value	Unit	Source
gf	84.62	kJ/mol	Joback Method
hf	-62.74	kJ/mol	Joback Method
hfus	30.90	kJ/mol	Thermodynamic aspects of solubility, solvation and partitioning processes of some sulfonamides
hfus	30.90 ± 0.50	kJ/mol	Impact of Sulfonamide Structure on Solubility and Transfer Processes in Biologically Relevant Solvents
hvap	96.60	kJ/mol	Joback Method
log10ws	-2.96		Crippen Method
logp	1.941		Crippen Method
mcvol	195.940	ml/mol	McGowan Method
рс	3745.38	kPa	Joback Method
tb	832.72	К	Joback Method
tc	1081.12	К	Joback Method
tf	451.50	К	Thermodynamic aspects of solubility, solvation and partitioning processes of some sulfonamides
tf	451.60 ± 0.20	К	Impact of Sulfonamide Structure on Solubility and Transfer Processes in Biologically Relevant Solvents
VC	0.763	m3/kmol	Joback Method

# **Temperature Dependent Properties**

Property code	Value	Unit	Temperature [K]	Source	
cpg	531.98	J/mol×K	832.72	Joback Method	
cpg	542.06	J/mol×K	874.12	Joback Method	
cpg	550.93	J/mol×K	915.52	Joback Method	
cpg	558.64	J/mol×K	956.92	Joback Method	
срд	565.21	J/mol×K	998.32	Joback Method	
срд	570.69	J/mol×K	1039.72	Joback Method	
срд	575.13	J/mol×K	1081.12	Joback Method	
psub	1.36e-05	kPa	426.55	Thermodynamic aspects of solubility, solvation and partitioning processes of some sulfonamides	
psub	7.23e-06	kPa	420.95	Thermodynamic aspects of solubility, solvation and partitioning processes of some sulfonamides	
psub	7.91e-06	kPa	421.35	Thermodynamic aspects of solubility, solvation and partitioning processes of some sulfonamides	
psub	1.13e-05	kPa	424.75	Thermodynamic aspects of solubility, solvation and partitioning processes of some sulfonamides	
psub	6.74e-06	kPa	420.15	Thermodynamic aspects of solubility, solvation and partitioning processes of some sulfonamides	
psub	1.66e-05	kPa	428.35	Thermodynamic aspects of solubility, solvation and partitioning processes of some sulfonamides	

psub	1.94e-05	kPa	429.65	Thermodynamic aspects of solubility, solvation and partitioning processes of some sulfonamides	
psub	2.11e-05	kPa	431.35	Thermodynamic aspects of solubility, solvation and partitioning processes of some sulfonamides	
psub	2.28e-05	kPa	431.75	Thermodynamic aspects of solubility, solvation and partitioning processes of some sulfonamides	
psub	2.33e-05	kPa	432.35	Thermodynamic aspects of solubility, solvation and partitioning processes of some sulfonamides	
psub	2.63e-05	kPa	433.05	Thermodynamic aspects of solubility, solvation and partitioning processes of some sulfonamides	
psub	3.17e-05	kPa	434.75	Thermodynamic aspects of solubility, solvation and partitioning processes of some sulfonamides	
psub	3.58e-05	kPa	435.75	Thermodynamic aspects of solubility, solvation and partitioning processes of some sulfonamides	

psub	4.24e-05	kPa	437.15	Thermodynamic aspects of solubility, solvation and partitioning processes of some sulfonamides	
psub	4.37e-05	kPa	437.55	Thermodynamic aspects of solubility, solvation and partitioning processes of some sulfonamides	

https://en.wikipedia.org/wiki/Joback\_method

http://pubs.acs.org/doi/abs/10.1021/ci990307I

https://www.doi.org/10.1016/j.jct.2010.12.007

https://www.doi.org/10.1021/je500918t

http://link.springer.com/article/10.1007/BF02311772

https://www.chemeo.com/doc/models/crippen\_log10ws

### Sources

Joback Method: McGowan Method: Crippen Method: Crippen Method: Thermodynamic aspects of solubility, solvation and partitioning processes of

solvation and partitioning processes of somecs of Solubility and Transfer Processes in Biologically Relevant Solvents:

# Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
psub:	Sublimation pressure
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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