

# 4-amino-N-(4-cyanophenyl)benzenesulfonamide

Inchi:	InChI=1S/C13H11N3O2S/c14-9-10-1-5-12(6-2-10)16-19(17,18)13-7-3-11(15)4-8-13/h1-8
InchiKey:	PLPNDNWDSGFETI-UHFFFAOYSA-N
Formula:	C13H11N3O2S
SMILES:	N#Cc1ccc(NS(=O)(=O)c2ccc(N)cc2)cc1
Mol. weight [g/mol]:	273.32

## Physical Properties

Property code	Value	Unit	Source
gf	84.62	kJ/mol	Joback Method
hf	-62.74	kJ/mol	Joback Method
hfus	30.90	kJ/mol	Thermodynamic aspects of solubility, solvation and partitioning processes of some sulfonamides
hfus	30.90 ± 0.50	kJ/mol	Impact of Sulfonamide Structure on Solubility and Transfer Processes in Biologically Relevant Solvents
hvap	96.60	kJ/mol	Joback Method
log10ws	-2.96		Crippen Method
logp	1.941		Crippen Method
mcvol	195.940	ml/mol	McGowan Method
pc	3745.38	kPa	Joback Method
tb	832.72	K	Joback Method
tc	1081.12	K	Joback Method
tf	451.60 ± 0.20	K	Impact of Sulfonamide Structure on Solubility and Transfer Processes in Biologically Relevant Solvents
tf	451.50	K	Thermodynamic aspects of solubility, solvation and partitioning processes of some sulfonamides
vc	0.763	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	575.13	J/molxK	1081.12	Joback Method
cpg	570.69	J/molxK	1039.72	Joback Method
cpg	565.21	J/molxK	998.32	Joback Method
cpg	558.64	J/molxK	956.92	Joback Method
cpg	550.93	J/molxK	915.52	Joback Method
cpg	542.06	J/molxK	874.12	Joback Method
cpg	531.98	J/molxK	832.72	Joback Method
psub	3.17e-05	kPa	434.75	Thermodynamic aspects of solubility, solvation and partitioning processes of some sulfonamides
psub	2.28e-05	kPa	431.75	Thermodynamic aspects of solubility, solvation and partitioning processes of some sulfonamides
psub	2.33e-05	kPa	432.35	Thermodynamic aspects of solubility, solvation and partitioning processes of some sulfonamides
psub	2.63e-05	kPa	433.05	Thermodynamic aspects of solubility, solvation and partitioning processes of some sulfonamides
psub	2.11e-05	kPa	431.35	Thermodynamic aspects of solubility, solvation and partitioning processes of some sulfonamides
psub	3.58e-05	kPa	435.75	Thermodynamic aspects of solubility, solvation and partitioning processes of some sulfonamides

psub	4.24e-05	kPa	437.15	Thermodynamic aspects of solubility, solvation and partitioning processes of some sulfonamides
psub	4.37e-05	kPa	437.55	Thermodynamic aspects of solubility, solvation and partitioning processes of some sulfonamides
psub	1.94e-05	kPa	429.65	Thermodynamic aspects of solubility, solvation and partitioning processes of some sulfonamides
psub	1.66e-05	kPa	428.35	Thermodynamic aspects of solubility, solvation and partitioning processes of some sulfonamides
psub	1.36e-05	kPa	426.55	Thermodynamic aspects of solubility, solvation and partitioning processes of some sulfonamides
psub	1.13e-05	kPa	424.75	Thermodynamic aspects of solubility, solvation and partitioning processes of some sulfonamides
psub	7.91e-06	kPa	421.35	Thermodynamic aspects of solubility, solvation and partitioning processes of some sulfonamides

psub	7.23e-06	kPa	420.95	Thermodynamic aspects of solubility, solvation and partitioning processes of some sulfonamides
psub	6.74e-06	kPa	420.15	Thermodynamic aspects of solubility, solvation and partitioning processes of some sulfonamides

## Sources

Crippen Method:	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
Crippen Method:	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
Thermodynamic aspects of solubility, solvation and partitioning processes of some sulfonamides	<a href="https://www.doi.org/10.1016/j.jct.2010.12.007">https://www.doi.org/10.1016/j.jct.2010.12.007</a>
Impact of Sulfonamide Structure on Solubility and Transfer Processes in Biologically Relevant Solvents:	<a href="https://www.doi.org/10.1021/je500918t">https://www.doi.org/10.1021/je500918t</a>
Joback Method:	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
McGowan Method:	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>psub:</b>	Sublimation pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.cheméo.com/cid/103-689-3/4-amino-N-4-cyanophenyl-benzenesulfonamide.pdf>

Generated by Cheméo on 2025-12-06 00:04:34.932881241 +0000 UTC m=+4727672.462921910.

Cheméo (<https://www.cheméo.com>) is the biggest free database of chemical and physical data for the process industry.