

# Nitrofen

<b>Other names:</b>	2',4'-Dichloro-4-nitrodiphenyl ether 2,4-Dichloro-4'-Nitrodiphenyl ether 2,4-Dichlorophenyl 4-nitrophenyl ether 2,4-Dichlorophenyl p-nitrophenyl ether 4'-Nitro-2,4-dichlorodiphenyl ether 4-(2,4-Dichlorophenoxy)nitrobenzene 4-Nitro-2',4'-dichlorophenyl ether Benzene, 2,4-dichloro-1-(4-nitrophenoxy)- Ether, 2,4-dichlorophenyl p-nitrophenyl FW 925 Mezotox NIP Niclofen Nitraphen Nitrochlor Nitrophen Nitrophenene Preparation 125 TOK TOK E TOK E 25 TOK E 40 Trizilin Trizilin 25
<b>Inchi:</b>	InChI=1S/C12H7Cl2NO3/c13-8-1-6-12(11(14)7-8)18-10-4-2-9(3-5-10)15(16)17/h1-7H
<b>InchiKey:</b>	XITQUSLLOS KDTB-UHFFFAOYSA-N
<b>Formula:</b>	C12H7Cl2NO3
<b>SMILES:</b>	O=[N+](O-)[c1ccc(Oc2ccc(Cl)cc2Cl)cc1
<b>Mol. weight [g/mol]:</b>	284.10
<b>CAS:</b>	1836-75-5

## Physical Properties

Property code	Value	Unit	Source
gf	152.78	kJ/mol	Joback Method
hf	-26.82	kJ/mol	Joback Method
hfus	34.69	kJ/mol	Joback Method

hvap	76.61			kJ/mol	Joback Method
log10ws	-5.46				Estimated Solubility Method
log10ws	-5.46				Aqueous Solubility Prediction Method
logp	4.694				Crippen Method
mcvol	180.190			ml/mol	McGowan Method
pc	2979.54			kPa	Joback Method
rinpol	2230.00				NIST Webbook
rinpol	2173.00				NIST Webbook
rinpol	2173.00				NIST Webbook
ripol	3353.00				NIST Webbook
ripol	3353.00				NIST Webbook
tb	791.38			K	Joback Method
tc	1065.45			K	Joback Method
tf	343.32			K	Aqueous Solubility Prediction Method
tf	342.72 ± 0.20			K	NIST Webbook
tf	343.00 ± 1.00			K	NIST Webbook
tf	344.70 ± 0.20			K	NIST Webbook
vc	0.690			m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	469.92	J/mol×K	974.09	Joback Method
cpg	476.06	J/mol×K	1019.77	Joback Method
cpg	435.12	J/mol×K	791.38	Joback Method
cpg	445.46	J/mol×K	837.06	Joback Method
cpg	454.67	J/mol×K	882.74	Joback Method
cpg	462.81	J/mol×K	928.41	Joback Method
cpg	481.28	J/mol×K	1065.45	Joback Method
hfust	22.96	kJ/mol	342.00	NIST Webbook
hfust	2.70	kJ/mol	343.00	NIST Webbook
hvapt	115.90	kJ/mol	643.00	NIST Webbook
hvapt	90.40	kJ/mol	365.50	NIST Webbook
sfust	7.87	J/mol×K	343.00	NIST Webbook
svapt	180.20	J/mol×K	643.00	NIST Webbook

# Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>Aqueous Solubility Prediction Method:</b>	<a href="http://onschallenge.wikispaces.com/file/view/AqueousDataset002.xlsx/351826032/AqueousDa">http://onschallenge.wikispaces.com/file/view/AqueousDataset002.xlsx/351826032/AqueousDa</a>
<b>Estimated Solubility Method:</b>	<a href="http://pubs.acs.org/doi/suppl/10.1021/ci034243x/suppl_file/ci034243xsi20040112_053635.txt">http://pubs.acs.org/doi/suppl/10.1021/ci034243x/suppl_file/ci034243xsi20040112_053635.txt</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C1836755&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C1836755&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hfust:</b>	Enthalpy of fusion at a given temperature
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>sfust:</b>	Entropy of fusion at a given temperature
<b>svapt:</b>	Entropy of vaporization at a given temperature
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.cheméo.com/cid/103-907-0/Nitrofen.pdf>

Generated by Cheméo on 2024-04-20 05:31:18.852943379 +0000 UTC m=+15880327.773520695.

Cheméo (<https://www.cheméo.com>) is the biggest free database of chemical and physical data for the process industry.