

# N-acetyl-N'-methyl-L-alpha-tyrosinamide

<b>Inchi:</b>	InChI=1S/C12H16N2O3/c1-8(15)14-11(12(17)13-2)7-9-3-5-10(16)6-4-9/h3-6,11,16H,7H2
<b>InchiKey:</b>	KHRMBHFDICMQHO-UHFFFAOYSA-N
<b>Formula:</b>	C12H16N2O3
<b>SMILES:</b>	CNC(=O)C(Cc1ccc(O)cc1)NC(C)=O
<b>Mol. weight [g/mol]:</b>	236.27

## Physical Properties

Property code	Value	Unit	Source
gf	-73.55	kJ/mol	Joback Method
hf	-355.29	kJ/mol	Joback Method
hfus	36.53	kJ/mol	Joback Method
hvap	83.57	kJ/mol	Joback Method
log10ws	-1.54		Crippen Method
logp	0.185		Crippen Method
mcvol	185.150	ml/mol	McGowan Method
pc	3341.24	kPa	Joback Method
tb	788.90	K	Joback Method
tc	1014.84	K	Joback Method
tf	199.00	K	Studies of heterogeneous interactions between N-acetyl-N'-methyl-L-a-amino acid amides and urea molecules in water at 298.15 K
vc	0.641	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	536.60	J/molxK	788.90	Joback Method
cpg	548.36	J/molxK	826.56	Joback Method
cpg	559.40	J/molxK	864.21	Joback Method
cpg	569.81	J/molxK	901.87	Joback Method
cpg	579.69	J/molxK	939.53	Joback Method
cpg	589.13	J/molxK	977.18	Joback Method
cpg	598.22	J/molxK	1014.84	Joback Method

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## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.cheméo.com/doc/models/crippen_log10ws">https://www.cheméo.com/doc/models/crippen_log10ws</a>
<b>Studies of heterogeneous interactions between N-acetyl-N'-methyl-L-<math>\alpha</math>-amino acids and urea molecules in water at 298.15 K:</b>	<a href="https://www.doi.org/10.1016/j.tca.2009.12.017">https://www.doi.org/10.1016/j.tca.2009.12.017</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvp:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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