

# 2-bromo-2-chloropropane

Inchi:	InChI=1S/C3H6BrCl/c1-3(2,4)5/h1-2H3
InchiKey:	YSQXISYCKHFJFJ-UHFFFAOYSA-N
Formula:	C3H6BrCl
SMILES:	CC(C)(Cl)Br
Mol. weight [g/mol]:	157.44
CAS:	2310-98-7

## Physical Properties

Property code	Value	Unit	Source
gf	-20.39	kJ/mol	Joback Method
hf	-103.41	kJ/mol	Joback Method
hfus	5.59	kJ/mol	Joback Method
hvap	31.80	kJ/mol	Joback Method
log10ws	-2.27		Crippen Method
logp	2.356		Crippen Method
mcvol	82.870	ml/mol	McGowan Method
pc	4723.61	kPa	Joback Method
tb	364.10 ± 0.50	K	NIST Webbook
tc	578.15	K	Joback Method
tf	215.71	K	Joback Method
vc	0.303	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	118.38	J/mol×K	368.40	Joback Method
cpg	125.76	J/mol×K	403.36	Joback Method
cpg	132.56	J/mol×K	438.32	Joback Method
cpg	138.83	J/mol×K	473.28	Joback Method
cpg	144.59	J/mol×K	508.24	Joback Method
cpg	149.89	J/mol×K	543.20	Joback Method
cpg	154.76	J/mol×K	578.15	Joback Method
dvisc	0.0058124	Paxs	215.71	Joback Method
dvisc	0.0030930	Paxs	241.16	Joback Method

dvisc	0.0018566	Paxs	266.61	Joback Method
dvisc	0.0012181	Paxs	292.06	Joback Method
dvisc	0.0008550	Paxs	317.50	Joback Method
dvisc	0.0006325	Paxs	342.95	Joback Method
dvisc	0.0004878	Paxs	368.40	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.cheméo.com/doc/models/crippen_log10ws">https://www.cheméo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C2310987&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C2310987&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.cheméo.com/cid/105-142-7/2-bromo-2-chloropropane.pdf>

Generated by Cheméo on 2024-04-24 16:11:50.808505395 +0000 UTC m=+16264359.729082759.

Cheméo (<https://www.cheméo.com>) is the biggest free database of chemical and physical data for the process industry.