

4-amino-N-(4-chlorophenyl)benzenesulfonamide

| | |
|-----------------------------|--|
| Inchi: | InChI=1S/C12H11ClN2O2S/c13-9-1-5-11(6-2-9)15-18(16,17)12-7-3-10(14)4-8-12/h1-8,1 |
| InchiKey: | RBJGYDGGKXTYSSL-UHFFFAOYSA-N |
| Formula: | C12H11ClN2O2S |
| SMILES: | <chem>Nc1ccc(S(=O)(=O)Nc2ccc(Cl)cc2)cc1</chem> |
| Mol. weight [g/mol]: | 282.75 |

Physical Properties

| Property code | Value | Unit | Source |
|---------------|---------------|----------------------|--|
| gf | -68.91 | kJ/mol | Joback Method |
| hf | -222.72 | kJ/mol | Joback Method |
| hfus | 23.80 ± 0.50 | kJ/mol | Impact of Sulfonamide Structure on Solubility and Transfer Processes in Biologically Relevant Solvents |
| hvap | 88.28 | kJ/mol | Joback Method |
| log10ws | -3.30 | | Crippen Method |
| logp | 2.723 | | Crippen Method |
| mcvol | 192.710 | ml/mol | McGowan Method |
| pc | 4162.33 | kPa | Joback Method |
| tb | 745.19 | K | Joback Method |
| tc | 991.17 | K | Joback Method |
| tf | 467.90 ± 0.20 | K | Impact of Sulfonamide Structure on Solubility and Transfer Processes in Biologically Relevant Solvents |
| vc | 0.731 | m ³ /kmol | Joback Method |

Temperature Dependent Properties

| Property code | Value | Unit | Temperature [K] | Source |
|---------------|--------|---------|-----------------|---------------|
| cpg | 489.22 | J/mol×K | 745.19 | Joback Method |
| cpg | 501.37 | J/mol×K | 786.19 | Joback Method |
| cpg | 512.29 | J/mol×K | 827.18 | Joback Method |
| cpg | 522.01 | J/mol×K | 868.18 | Joback Method |
| cpg | 530.58 | J/mol×K | 909.17 | Joback Method |

| | | | | |
|-----|--------|---------|--------|---------------|
| cpg | 538.03 | J/mol×K | 950.17 | Joback Method |
| cpg | 544.41 | J/mol×K | 991.17 | Joback Method |

Sources

Impact of Sulfonamide Structure on Solubility and Transfer Processes in Biologically Relevant Solvents:

<https://www.doi.org/10.1021/je500918t>

Joback Method:

https://en.wikipedia.org/wiki/Joback_method

McGowan Method:

<http://link.springer.com/article/10.1007/BF02311772>

Crippen Method:

<http://pubs.acs.org/doi/abs/10.1021/ci990307i>

Crippen Method:

https://www.chemeo.com/doc/models/crippen_log10ws

Legend

| | |
|-----------------|---|
| cpg: | Ideal gas heat capacity |
| gf: | Standard Gibbs free energy of formation |
| hf: | Enthalpy of formation at standard conditions |
| hfus: | Enthalpy of fusion at standard conditions |
| h vap: | Enthalpy of vaporization at standard conditions |
| log10ws: | Log10 of Water solubility in mol/l |
| logp: | Octanol/Water partition coefficient |
| mcvol: | McGowan's characteristic volume |
| pc: | Critical Pressure |
| tb: | Normal Boiling Point Temperature |
| tc: | Critical Temperature |
| tf: | Normal melting (fusion) point |
| vc: | Critical Volume |

Latest version available from:

<https://www.chemeo.com/cid/105-572-0/4-amino-N-4-chlorophenyl-benzenesulfonamide.pdf>

Generated by Cheméo on 2024-04-26 03:35:55.506115298 +0000 UTC m=+16391804.426692613.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.