

5-Ethyl-5-pentylbarbituric acid

Inchi:	InChI=1S/C11H18N2O3/c1-3-5-6-7-11(4-2)8(14)12-10(16)13-9(11)15/h3-7H2,1-2H3,(H2
InchiKey:	XYGXSCVUMIDZRR-UHFFFAOYSA-N
Formula:	C11H18N2O3
SMILES:	CCCCC1(CC)C(=O)NC(=O)NC1=O
Mol. weight [g/mol]:	226.28

Physical Properties

Property code	Value	Unit	Source
gf	-131.65	kJ/mol	Joback Method
hf	-538.29	kJ/mol	Joback Method
hfus	27.49	kJ/mol	Joback Method
hvap	65.61	kJ/mol	Joback Method
log10ws	-2.34		Aqueous and cosolvent solubility data for drug-like organic compounds
logp	1.329		Crippen Method
mcvol	179.660	ml/mol	McGowan Method
pc	2862.74	kPa	Joback Method
tb	771.43	K	Joback Method
tc	1018.96	K	Joback Method
tf	659.73	K	Joback Method
vc	0.677	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	549.70	J/mol×K	771.43	Joback Method
cpg	567.97	J/mol×K	812.69	Joback Method
cpg	585.29	J/mol×K	853.94	Joback Method
cpg	601.66	J/mol×K	895.20	Joback Method
cpg	617.09	J/mol×K	936.45	Joback Method
cpg	631.61	J/mol×K	977.71	Joback Method
cpg	645.20	J/mol×K	1018.96	Joback Method

Sources

Aqueous and cosolvent solubility data for drug-like organic compounds: McGowan Method: <https://www.ncbi.nlm.nih.gov/pmc/articles/PMC2751500/>
<http://link.springer.com/article/10.1007/BF02311772>

Crippen Method: <http://pubs.acs.org/doi/abs/10.1021/ci990307l>

Joback Method: https://en.wikipedia.org/wiki/Joback_method

Legend

cpg: Ideal gas heat capacity
gf: Standard Gibbs free energy of formation
hf: Enthalpy of formation at standard conditions
hfus: Enthalpy of fusion at standard conditions
hvap: Enthalpy of vaporization at standard conditions
log10ws: Log10 of Water solubility in mol/l
logp: Octanol/Water partition coefficient
mcvol: McGowan's characteristic volume
pc: Critical Pressure
tb: Normal Boiling Point Temperature
tc: Critical Temperature
tf: Normal melting (fusion) point
vc: Critical Volume

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