

[2-(dipropylamino)-2-oxoethyl] benzoate

Inchi:	InChI=1S/C15H21NO3/c1-3-10-16(11-4-2)14(17)12-19-15(18)13-8-6-5-7-9-13/h5-9H,3-4
InchiKey:	COTRONQEHWUCRV-UHFFFAOYSA-N
Formula:	C15H21NO3
SMILES:	CCCN(CCC)C(=O)COC(=O)c1ccccc1
Mol. weight [g/mol]:	263.34

Physical Properties

Property code	Value	Unit	Source
gf	-64.23	kJ/mol	Joback Method
hf	-406.25	kJ/mol	Joback Method
hfus	36.05	kJ/mol	Joback Method
hvap	69.20	kJ/mol	Joback Method
log10ws	-2.38		Aqueous Solubility Prediction Method
logp	2.492		Crippen Method
mcvol	217.440	ml/mol	McGowan Method
pc	2029.06	kPa	Joback Method
tb	711.88	K	Joback Method
tc	914.00	K	Joback Method
tf	439.79	K	Joback Method
vc	0.816	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	606.84	J/molxK	711.88	Joback Method
cpg	622.25	J/molxK	745.57	Joback Method
cpg	636.67	J/molxK	779.25	Joback Method
cpg	650.13	J/molxK	812.94	Joback Method
cpg	662.69	J/molxK	846.63	Joback Method
cpg	674.36	J/molxK	880.32	Joback Method
cpg	685.18	J/molxK	914.00	Joback Method

Sources

Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci9903071
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
Aqueous Solubility Prediction Method:	http://onschallenge.wikispaces.com/file/view/AqueousDataset002.xlsx/351826032/AqueousDa
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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