

# Methanone, bis[4-(dimethylamino)phenyl]-

<b>Other names:</b>	Benzophenone, 4,4'-bis(dimethylamino)- p,p'-bis(Dimethylamino)benzophenone Bis[p-(N,N-dimethylamino)phenyl] ketone Michler's ketone N,N,N',N'-Tetramethyl-4,4'-diaminobenzophenone 4,4'-Bis(dimethylamino)benzophenone 4,4'-di(Dimethylamino)benzophenone Michlers ketone p,p'-Tetramethyldiaminobenzophenone (4,4'-Tetramethyldiamino)benzophenone p,p'-Bis(N,N-dimethylamino)benzophenone p,p'-Michler's ketone Michler ketone NCI-C02006 Bis(4-(dimethylamino)phenyl)methanone 4,4'-Bis(N,N-dimethylamino)benzophenone NSC 9602
<b>Inchi:</b>	InChI=1S/C17H20N2O/c1-18(2)15-9-5-13(6-10-15)17(20)14-7-11-16(12-8-14)19(3)4/h5-
<b>InchiKey:</b>	VVBLNCFGVYUYGU-UHFFFAOYSA-N
<b>Formula:</b>	C17H20N2O
<b>SMILES:</b>	CN(C)c1ccc(C(=O)c2ccc(N(C)C)cc2)cc1
<b>Mol. weight [g/mol]:</b>	268.35
<b>CAS:</b>	90-94-8

## Physical Properties

Property code	Value	Unit	Source
gf	390.46	kJ/mol	Joback Method
hf	78.39	kJ/mol	Joback Method
hfus	34.73	kJ/mol	Joback Method
hvap	70.14	kJ/mol	Joback Method
log10ws	-3.25		Crippen Method
logp	3.050		Crippen Method
mcvol	224.400	ml/mol	McGowan Method
pc	2145.33	kPa	Joback Method
tb	730.43	K	Joback Method
tc	957.04	K	Joback Method
tf	474.10	K	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	622.97	J/mol×K	730.43	Joback Method
cpg	639.75	J/mol×K	768.20	Joback Method
cpg	655.28	J/mol×K	805.97	Joback Method
cpg	669.63	J/mol×K	843.74	Joback Method
cpg	682.89	J/mol×K	881.51	Joback Method
cpg	695.13	J/mol×K	919.28	Joback Method
cpg	706.46	J/mol×K	957.04	Joback Method

## Sources

**Crippen Method:**

<http://pubs.acs.org/doi/abs/10.1021/ci990307l>

**Crippen Method:**

[https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)

**Joback Method:**

[https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

**McGowan Method:**

<http://link.springer.com/article/10.1007/BF02311772>

**NIST Webbook:**

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C90948&Units=SI>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point

**vc:** Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/108-896-8/Methanone-bis-4-dimethylamino-phenyl.pdf>

Generated by Cheméo on 2024-07-21 16:31:03.829719251 +0000 UTC m=+269333.076824614.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.