

2,3-Dimethyltridecane

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|-----------------------------|---|
| Inchi: | InChI=1S/C15H32/c1-5-6-7-8-9-10-11-12-13-15(4)14(2)3/h14-15H,5-13H2,1-4H3 |
| InchiKey: | SWUJSLXRUPXTQB-UHFFFAOYSA-N |
| Formula: | C15H32 |
| SMILES: | CCCCCCCCCCC(C)C(C)C |
| Mol. weight [g/mol]: | 212.41 |
| CAS: | 18435-20-6 |

Physical Properties

| Property code | Value | Unit | Source |
|---------------|---------|---------|----------------|
| gf | 70.54 | kJ/mol | Joback Method |
| hf | -363.49 | kJ/mol | Joback Method |
| hfus | 27.56 | kJ/mol | Joback Method |
| hvap | 48.21 | kJ/mol | Joback Method |
| log10ws | -5.62 | | Crippen Method |
| logp | 5.809 | | Crippen Method |
| mcvol | 222.210 | ml/mol | McGowan Method |
| pc | 1432.63 | kPa | Joback Method |
| tb | 541.72 | K | Joback Method |
| tc | 706.30 | K | Joback Method |
| tf | 228.81 | K | Joback Method |
| vc | 0.864 | m3/kmol | Joback Method |

Temperature Dependent Properties

| Property code | Value | Unit | Temperature [K] | Source |
|---------------|-----------|---------|-----------------|---------------|
| cpg | 558.41 | J/molxK | 541.72 | Joback Method |
| cpg | 577.82 | J/molxK | 569.15 | Joback Method |
| cpg | 596.45 | J/molxK | 596.58 | Joback Method |
| cpg | 614.33 | J/molxK | 624.01 | Joback Method |
| cpg | 631.47 | J/molxK | 651.44 | Joback Method |
| cpg | 647.90 | J/molxK | 678.87 | Joback Method |
| cpg | 663.64 | J/molxK | 706.30 | Joback Method |
| dvisc | 0.0138942 | Paxs | 228.81 | Joback Method |
| dvisc | 0.0032300 | Paxs | 280.96 | Joback Method |

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|-------|-----------|--------|--------|---------------|
| dvisc | 0.0011857 | Paxs | 333.11 | Joback Method |
| dvisc | 0.0005709 | Paxs | 385.26 | Joback Method |
| dvisc | 0.0003272 | Paxs | 437.42 | Joback Method |
| dvisc | 0.0002112 | Paxs | 489.57 | Joback Method |
| dvisc | 0.0001483 | Paxs | 541.72 | Joback Method |
| hvapt | 56.30 | kJ/mol | 468.00 | NIST Webbook |

Sources

| | |
|------------------------|---|
| Crippen Method: | http://pubs.acs.org/doi/abs/10.1021/ci9903071 |
| Crippen Method: | https://www.chemeo.com/doc/models/crippen_log10ws |
| Joback Method: | https://en.wikipedia.org/wiki/Joback_method |
| McGowan Method: | http://link.springer.com/article/10.1007/BF02311772 |
| NIST Webbook: | http://webbook.nist.gov/cgi/cbook.cgi?ID=C18435206&Units=SI |

Legend

| | |
|-----------------|---|
| cpg: | Ideal gas heat capacity |
| dvisc: | Dynamic viscosity |
| gf: | Standard Gibbs free energy of formation |
| hf: | Enthalpy of formation at standard conditions |
| hfus: | Enthalpy of fusion at standard conditions |
| hvap: | Enthalpy of vaporization at standard conditions |
| hvapt: | Enthalpy of vaporization at a given temperature |
| log10ws: | Log10 of Water solubility in mol/l |
| logp: | Octanol/Water partition coefficient |
| mcvol: | McGowan's characteristic volume |
| pc: | Critical Pressure |
| tb: | Normal Boiling Point Temperature |
| tc: | Critical Temperature |
| tf: | Normal melting (fusion) point |
| vc: | Critical Volume |

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