

Pentan-2-one, 4-methyl-4-ethoxyamino

Inchi:	InChI=1S/C8H17NO2/c1-5-11-9-8(3,4)6-7(2)10/h9H,5-6H2,1-4H3
InchiKey:	VKMICPIUEABHKH-UHFFFAOYSA-N
Formula:	C8H17NO2
SMILES:	CCONC(C)(C)CC(C)=O
Mol. weight [g/mol]:	159.23

Physical Properties

Property code	Value	Unit	Source
gf	-125.21	kJ/mol	Joback Method
hf	-408.53	kJ/mol	Joback Method
hfus	16.95	kJ/mol	Joback Method
hvap	47.70	kJ/mol	Joback Method
log10ws	-1.83		Crippen Method
logp	1.285		Crippen Method
mcvol	141.000	ml/mol	McGowan Method
pc	2746.90	kPa	Joback Method
rinpol	1007.00		NIST Webbook
tb	505.67	K	Joback Method
tc	694.20	K	Joback Method
tf	307.16	K	Joback Method
vc	0.531	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	329.96	J/molxK	505.67	Joback Method
cpg	343.63	J/molxK	537.09	Joback Method
cpg	356.62	J/molxK	568.51	Joback Method
cpg	368.97	J/molxK	599.94	Joback Method
cpg	380.70	J/molxK	631.36	Joback Method
cpg	391.81	J/molxK	662.78	Joback Method
cpg	402.34	J/molxK	694.20	Joback Method

Sources

Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=R315794&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307I

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvac:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mccol:	McGowan's characteristic volume
pc:	Critical Pressure
rinpol:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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