

# Phthalic acid, 3,3-dimethylbut-2-yl octyl ester

<b>Inchi:</b>	InChI=1S/C22H34O4/c1-6-7-8-9-10-13-16-25-20(23)18-14-11-12-15-19(18)21(24)26-17(
<b>InchiKey:</b>	TXFQFZFATOIAER-UHFFFAOYSA-N
<b>Formula:</b>	C22H34O4
<b>SMILES:</b>	CCCCCCCCOC(=O)c1cccc1C(=O)OC(C)C(C)(C)C
<b>Mol. weight [g/mol]:</b>	362.50

## Physical Properties

Property code	Value	Unit	Source
gf	-230.30	kJ/mol	Joback Method
hf	-775.98	kJ/mol	Joback Method
hfus	41.02	kJ/mol	Joback Method
hvap	84.13	kJ/mol	Joback Method
log10ws	-6.85		Crippen Method
logp	5.795		Crippen Method
mvol	311.960	ml/mol	McGowan Method
pc	1193.17	kPa	Joback Method
rinpol	2428.00		NIST Webbook
rinpol	2428.00		NIST Webbook
tb	883.33	K	Joback Method
tc	1091.23	K	Joback Method
tf	508.38	K	Joback Method
vc	1.190	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	998.32	J/molxK	883.33	Joback Method
cpg	1015.05	J/molxK	917.98	Joback Method
cpg	1030.54	J/molxK	952.63	Joback Method
cpg	1044.85	J/molxK	987.28	Joback Method
cpg	1058.01	J/molxK	1021.93	Joback Method
cpg	1070.07	J/molxK	1056.58	Joback Method
cpg	1081.10	J/molxK	1091.23	Joback Method
dvisc	0.0004944	Paxs	508.38	Joback Method

dvisc	0.0002388	Paxs	570.87	Joback Method
dvisc	0.0001332	Paxs	633.36	Joback Method
dvisc	0.0000825	Paxs	695.86	Joback Method
dvisc	0.0000553	Paxs	758.35	Joback Method
dvisc	0.0000394	Paxs	820.84	Joback Method
dvisc	0.0000294	Paxs	883.33	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U357007&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U357007&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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