

# Diethylmalonic acid, monochloride, 2,6-dimethoxyphenyl ester

Inchi:	InChI=1S/C15H19ClO5/c1-5-15(6-2,13(16)17)14(18)21-12-10(19-3)8-7-9-11(12)20-4/h7-
InchiKey:	SPKLDHKFKUDGKF-UHFFFAOYSA-N
Formula:	C15H19ClO5
SMILES:	CCC(CC)(C(=O)Cl)C(=O)Oc1c(OC)cccc1OC
Mol. weight [g/mol]:	314.76

## Physical Properties

Property code	Value	Unit	Source
gf	-413.36	kJ/mol	Joback Method
hf	-785.65	kJ/mol	Joback Method
hfus	31.41	kJ/mol	Joback Method
hvap	76.39	kJ/mol	Joback Method
log10ws	-3.80		Crippen Method
logp	3.181		Crippen Method
mcvol	231.440	ml/mol	McGowan Method
pc	1890.36	kPa	Joback Method
rinsol	2084.00		NIST Webbook
tb	788.44	K	Joback Method
tc	1003.86	K	Joback Method
tf	509.16	K	Joback Method
vc	0.872	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	647.61	J/molxK	788.44	Joback Method
cpg	661.06	J/molxK	824.34	Joback Method
cpg	673.45	J/molxK	860.25	Joback Method
cpg	684.80	J/molxK	896.15	Joback Method
cpg	695.12	J/molxK	932.05	Joback Method
cpg	704.41	J/molxK	967.96	Joback Method
cpg	712.71	J/molxK	1003.86	Joback Method
dvisc	0.0004457	Paxs	509.16	Joback Method
dvisc	0.0002755	Paxs	555.71	Joback Method

dvisc	0.0001834	Paxs	602.25	Joback Method
dvisc	0.0001295	Paxs	648.80	Joback Method
dvisc	0.0000958	Paxs	695.35	Joback Method
dvisc	0.0000735	Paxs	741.89	Joback Method
dvisc	0.0000583	Paxs	788.44	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U369819&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U369819&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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