

# Ethanol, 2-methoxy-, carbonate (2:1)

<b>Other names:</b>	Bis(2-methoxyethyl) carbonate Ethanol, 2-methoxy-, carbonate
<b>Inchi:</b>	InChI=1S/C7H14O5/c1-9-3-5-11-7(8)12-6-4-10-2/h3-6H2,1-2H3
<b>InchiKey:</b>	STLGQBDVSTWGIP-UHFFFAOYSA-N
<b>Formula:</b>	C7H14O5
<b>SMILES:</b>	COCCOC(=O)OCCOC
<b>Mol. weight [g/mol]:</b>	178.18
<b>CAS:</b>	626-84-6

## Physical Properties

Property code	Value	Unit	Source
gf	-540.86	kJ/mol	Joback Method
hf	-829.27	kJ/mol	Joback Method
hfus	20.24	kJ/mol	Joback Method
hvap	47.56	kJ/mol	Joback Method
log10ws	0.15		Crippen Method
logp	0.432		Crippen Method
mvol	134.540	ml/mol	McGowan Method
pc	2784.72	kPa	Joback Method
tb	503.11	K	Joback Method
tc	603.50 ± 0.70	K	NIST Webbook
tf	307.50	K	Joback Method
vc	0.505	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	307.98	J/molxK	503.11	Joback Method
cpg	318.82	J/molxK	532.27	Joback Method
cpg	329.42	J/molxK	561.42	Joback Method
cpg	339.76	J/molxK	590.58	Joback Method
cpg	349.80	J/molxK	619.73	Joback Method
cpg	359.52	J/molxK	648.89	Joback Method
cpg	368.91	J/molxK	678.04	Joback Method

dvisc	0.0013531	Paxs	307.50	Joback Method
dvisc	0.0007955	Paxs	340.10	Joback Method
dvisc	0.0005132	Paxs	372.70	Joback Method
dvisc	0.0003553	Paxs	405.31	Joback Method
dvisc	0.0002598	Paxs	437.91	Joback Method
dvisc	0.0001984	Paxs	470.51	Joback Method
dvisc	0.0001569	Paxs	503.11	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C626846&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C626846&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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