

# 10-Nonadecanone

<b>Other names:</b>	Di-n-nonyl ketone Dinonyl ketone nonadecan-10-one
<b>Inchi:</b>	InChI=1S/C19H38O/c1-3-5-7-9-11-13-15-17-19(20)18-16-14-12-10-8-6-4-2/h3-18H2,1-2H
<b>InchiKey:</b>	YUPOCHDBBHTUBJ-UHFFFAOYSA-N
<b>Formula:</b>	C19H38O
<b>SMILES:</b>	CCCCCCCCC(=O)CCCCCCCCC
<b>Mol. weight [g/mol]:</b>	282.50
<b>CAS:</b>	504-57-4

## Physical Properties

Property code	Value	Unit	Source
gf	-19.82	kJ/mol	Joback Method
hf	-548.07	kJ/mol	Joback Method
hfus	46.56	kJ/mol	Joback Method
hvap	64.63	kJ/mol	Joback Method
log10ws	-7.06		Crippen Method
logp	6.837		Crippen Method
mcvol	280.140	ml/mol	McGowan Method
pc	1126.83	kPa	Joback Method
rinpola	2045.00		NIST Webbook
rinpola	2082.00		NIST Webbook
rinpola	2045.00		NIST Webbook
tb	687.99	K	Joback Method
tc	855.89	K	Joback Method
tf	330.00 ± 0.10	K	NIST Webbook
tf	329.15 ± 1.50	K	NIST Webbook
tf	330.00 ± 0.50	K	NIST Webbook
vc	1.105	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	904.83	J/mol×K	827.90	Joback Method

cpg	920.22	J/molxK	855.89	Joback Method
cpg	816.10	J/molxK	687.99	Joback Method
cpg	835.51	J/molxK	715.97	Joback Method
cpg	854.06	J/molxK	743.96	Joback Method
cpg	871.78	J/molxK	771.94	Joback Method
cpg	888.70	J/molxK	799.92	Joback Method
dvisc	0.0001094	Paxs	687.99	Joback Method
dvisc	0.0001477	Paxs	632.30	Joback Method
dvisc	0.0027375	Paxs	353.82	Joback Method
dvisc	0.0011112	Paxs	409.51	Joback Method
dvisc	0.0005597	Paxs	465.21	Joback Method
dvisc	0.0003265	Paxs	520.90	Joback Method
dvisc	0.0002113	Paxs	576.60	Joback Method
hfust	66.67	kJ/mol	330.00	NIST Webbook
hfust	66.67	kJ/mol	330.00	NIST Webbook

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.45877e+01
Coeff. B	-5.10168e+03
Coeff. C	-1.11522e+02
Temperature range (K), min.	468.28
Temperature range (K), max.	661.49

## Sources

**The Yaws Handbook of Vapor**

**Pressure:**  
Crippen Method:

<https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure>

<http://pubs.acs.org/doi/abs/10.1021/ci990307I>

**Crippen Method:**

[https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)

**Joback Method:**

[https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

**McGowan Method:**

<http://link.springer.com/article/10.1007/BF02311772>

**NIST Webbook:**

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C504574&Units=SI>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hfust:</b>	Enthalpy of fusion at a given temperature
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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