

# Equilin

**Other names:**

1,3,5,7-Estratetraen-3-ol-17-one  
3-Hydroxyestra-1,3,5(10),7-tetraen-17-one  
3-hydroxyoestra-1,3,5(10),7-tetraen-17-one  
7-Dehydroestrone  
Dihydroequilenin

**Inchi:**

Estra-1,3,5(10),7-tetraen-17-one, 3-hydroxy-  
InChI=1S/C18H20O2/c1-18-9-8-14-13-5-3-12(19)10-11(13)2-4-15(14)16(18)6-7-17(18)20

**InchiKey:**

WKRLQDKEXYKHJB-UHFFFAOYSA-N

**Formula:**

C<sub>18</sub>H<sub>20</sub>O<sub>2</sub>

**SMILES:**

CC12CCC3C(=CCc4cc(O)ccc43)C1CCC2=O

**Mol. weight [g/mol]:**

268.35

**CAS:**

474-86-2

## Physical Properties

Property code	Value	Unit	Source
gf	99.14	kJ/mol	Joback Method
hf	-237.17	kJ/mol	Joback Method
hfus	26.06	kJ/mol	Joback Method
hvap	75.75	kJ/mol	Joback Method
log10ws	-5.28		Estimated Solubility Method
log10ws	-5.28		Aqueous Solubility Prediction Method
log10ws	-5.28		Aqueous and cosolvent solubility data for drug-like organic compounds
logp	3.737		Crippen Method
mvol	211.280	ml/mol	McGowan Method
pc	2657.03	kPa	Joback Method
rinpol	2766.10		NIST Webbook
rinpol	2765.40		NIST Webbook
rinpol	2745.30		NIST Webbook
tb	824.48	K	Joback Method
tc	1091.16	K	Joback Method
tf	595.46	K	Joback Method
vc	0.748	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	682.95	J/molxK	824.48	Joback Method
cpg	703.18	J/molxK	868.93	Joback Method
cpg	723.31	J/molxK	913.37	Joback Method
cpg	743.72	J/molxK	957.82	Joback Method
cpg	764.81	J/molxK	1002.27	Joback Method
cpg	786.96	J/molxK	1046.71	Joback Method
cpg	810.57	J/molxK	1091.16	Joback Method

## Sources

**Aqueous Solubility Prediction Method:** <http://onschallenge.wikispaces.com/file/view/AqueousDataset002.xlsx/351826032/AqueousDa>

**Estimated Solubility Method:** [http://pubs.acs.org/doi/suppl/10.1021/ci034243x/suppl\\_file/ci034243xsi20040112\\_053635.txt](http://pubs.acs.org/doi/suppl/10.1021/ci034243x/suppl_file/ci034243xsi20040112_053635.txt)

**Aqueous and cosolvent solubility data for drug-like organic compounds:** <https://www.ncbi.nlm.nih.gov/pmc/articles/PMC2751500/>

**McGowan Method:** <http://link.springer.com/article/10.1007/BF02311772>

**NIST Webbook:** <http://webbook.nist.gov/cgi/cbook.cgi?ID=C474862&Units=SI>

**Crippen Method:** <http://pubs.acs.org/doi/abs/10.1021/ci990307I>

**Joback Method:** [https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point

**vc:** Critical Volume

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