

# 1,3-Cyclohexadiene, 1-bromo

<b>Inchi:</b>	InChI=1S/C6H7Br/c7-6-4-2-1-3-5-6/h1-2,4H,3,5H2
<b>InchiKey:</b>	GHRZOGONMXDLAP-UHFFFAOYSA-N
<b>Formula:</b>	C6H7Br
<b>SMILES:</b>	BrC1=CC=CCC1
<b>Mol. weight [g/mol]:</b>	159.02

## Physical Properties

Property code	Value	Unit	Source
gf	96.41	kJ/mol	Joback Method
hf	37.91	kJ/mol	Joback Method
hfus	9.40	kJ/mol	Joback Method
hvap	37.37	kJ/mol	Joback Method
log10ws	-2.86		Crippen Method
logp	2.615		Crippen Method
mvol	93.440	ml/mol	McGowan Method
pc	4890.21	kPa	Joback Method
rinpol	980.00		NIST Webbook
rinpol	980.00		NIST Webbook
tb	430.36	K	Joback Method
tc	663.66	K	Joback Method
tf	242.84	K	Joback Method
vc	0.340	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	149.28	J/molxK	430.36	Joback Method
cpg	197.06	J/molxK	624.78	Joback Method
cpg	188.88	J/molxK	585.90	Joback Method
cpg	180.06	J/molxK	547.01	Joback Method
cpg	170.54	J/molxK	508.13	Joback Method
cpg	160.29	J/molxK	469.24	Joback Method
cpg	204.62	J/molxK	663.66	Joback Method
dvisc	0.0003474	Paxs	430.36	Joback Method

dvisc	0.0004393	Paxs	399.11	Joback Method
dvisc	0.0005781	Paxs	367.85	Joback Method
dvisc	0.0008004	Paxs	336.60	Joback Method
dvisc	0.0011846	Paxs	305.35	Joback Method
dvisc	0.0019173	Paxs	274.09	Joback Method
dvisc	0.0035125	Paxs	242.84	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R25227&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R25227&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>m<sub>cvol</sub>:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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