

Glutaric acid, isohexyl 2,3,6-trichlorophenyl ester

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|----------------------|---|
| Inchi: | InChI=1S/C17H21Cl3O4/c1-11(2)5-4-10-23-14(21)6-3-7-15(22)24-17-13(19)9-8-12(18)16 |
| InchiKey: | WYMMYIJXBWGCSK-UHFFFAOYSA-N |
| Formula: | C17H21Cl3O4 |
| SMILES: | CC(C)CCCOC(=O)CCCC(=O)Oc1c(Cl)ccc(Cl)c1Cl |
| Mol. weight [g/mol]: | 395.70 |

Physical Properties

| Property code | Value | Unit | Source |
|---------------|---------|----------------------|----------------|
| gf | -330.29 | kJ/mol | Joback Method |
| hf | -734.19 | kJ/mol | Joback Method |
| hfus | 47.30 | kJ/mol | Joback Method |
| hvap | 88.78 | kJ/mol | Joback Method |
| log10ws | -6.23 | | Crippen Method |
| logp | 5.702 | | Crippen Method |
| mcvol | 278.230 | ml/mol | McGowan Method |
| pc | 1508.15 | kPa | Joback Method |
| rinpol | 2665.00 | | NIST Webbook |
| rinpol | 2665.00 | | NIST Webbook |
| tb | 894.41 | K | Joback Method |
| tc | 1111.41 | K | Joback Method |
| tf | 564.41 | K | Joback Method |
| vc | 1.069 | m ³ /kmol | Joback Method |

Temperature Dependent Properties

| Property code | Value | Unit | Temperature [K] | Source |
|---------------|-----------|---------|-----------------|---------------|
| cpg | 777.49 | J/molxK | 894.41 | Joback Method |
| cpg | 789.38 | J/molxK | 930.58 | Joback Method |
| cpg | 800.17 | J/molxK | 966.74 | Joback Method |
| cpg | 809.86 | J/molxK | 1002.91 | Joback Method |
| cpg | 818.46 | J/molxK | 1039.08 | Joback Method |
| cpg | 826.00 | J/molxK | 1075.25 | Joback Method |
| cpg | 832.48 | J/molxK | 1111.41 | Joback Method |
| dvisc | 0.0003779 | Paxs | 564.41 | Joback Method |

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|-------|-----------|------|--------|---------------|
| dvisc | 0.0002328 | Paxs | 619.41 | Joback Method |
| dvisc | 0.0001552 | Paxs | 674.41 | Joback Method |
| dvisc | 0.0001100 | Paxs | 729.41 | Joback Method |
| dvisc | 0.0000818 | Paxs | 784.41 | Joback Method |
| dvisc | 0.0000633 | Paxs | 839.41 | Joback Method |
| dvisc | 0.0000505 | Paxs | 894.41 | Joback Method |

Sources

| | |
|------------------------|---|
| Crippen Method: | https://www.chemeo.com/doc/models/crippen_log10ws |
| Joback Method: | https://en.wikipedia.org/wiki/Joback_method |
| McGowan Method: | http://link.springer.com/article/10.1007/BF02311772 |
| NIST Webbook: | http://webbook.nist.gov/cgi/cbook.cgi?ID=U359245&Units=SI |
| Crippen Method: | http://pubs.acs.org/doi/abs/10.1021/ci990307l |

Legend

| | |
|-----------------|---|
| cpg: | Ideal gas heat capacity |
| dvisc: | Dynamic viscosity |
| gf: | Standard Gibbs free energy of formation |
| hf: | Enthalpy of formation at standard conditions |
| hfus: | Enthalpy of fusion at standard conditions |
| hvap: | Enthalpy of vaporization at standard conditions |
| log10ws: | Log10 of Water solubility in mol/l |
| logp: | Octanol/Water partition coefficient |
| mcvol: | McGowan's characteristic volume |
| pc: | Critical Pressure |
| rinpol: | Non-polar retention indices |
| tb: | Normal Boiling Point Temperature |
| tc: | Critical Temperature |
| tf: | Normal melting (fusion) point |
| vc: | Critical Volume |

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<https://www.chemeo.com/cid/11-264-6/Glutaric-acid-isoheptyl-2-3-6-trichlorophenyl-ester.pdf>

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