

# 1,3-Dioxane, 2,2-dimethyl-

<b>Other names:</b>	2,2-Dimethyl-1,3-dioxane
<b>Inchi:</b>	InChI=1S/C6H12O2/c1-6(2)7-4-3-5-8-6/h3-5H2,1-2H3
<b>InchiKey:</b>	RPLSBADGISFNSI-UHFFFAOYSA-N
<b>Formula:</b>	C6H12O2
<b>SMILES:</b>	CC1(C)OCCCO1
<b>Mol. weight [g/mol]:</b>	116.16
<b>CAS:</b>	695-30-7

## Physical Properties

Property code	Value	Unit	Source
chl	-3607.20 ± 2.00	kJ/mol	NIST Webbook
gf	-153.64	kJ/mol	Joback Method
hf	-361.61	kJ/mol	Joback Method
hfl	-468.90 ± 2.00	kJ/mol	NIST Webbook
hfus	12.79	kJ/mol	Joback Method
hvap	37.25	kJ/mol	Joback Method
ie	9.84	eV	NIST Webbook
log10ws	-1.01		Crippen Method
logp	1.159		Crippen Method
mcvol	96.280	ml/mol	McGowan Method
pc	4098.62	kPa	Joback Method
rinpol	820.00		NIST Webbook
rinpol	796.00		NIST Webbook
tb	397.65 ± 1.50	K	NIST Webbook
tc	625.18	K	Joback Method
tf	241.80	K	Joback Method
vc	0.344	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	194.00	J/mol×K	410.37	Joback Method
cpg	208.82	J/mol×K	446.17	Joback Method
cpg	222.56	J/mol×K	481.97	Joback Method

cpg	235.32	J/mol×K	517.77	Joback Method
cpg	247.20	J/mol×K	553.57	Joback Method
cpg	258.29	J/mol×K	589.38	Joback Method
cpg	268.70	J/mol×K	625.18	Joback Method
hfust	12.10	kJ/mol	229.60	NIST Webbook

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C695307&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C695307&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307I">http://pubs.acs.org/doi/abs/10.1021/ci990307I</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>chl:</b>	Standard liquid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfl:</b>	Liquid phase enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hfust:</b>	Enthalpy of fusion at a given temperature
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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