

# Cyclopropane, 1,1-diethynyl-

Inchi:	InChI=1S/C7H6/c1-3-7(4-2)5-6-7/h1-2H,5-6H2
InchiKey:	NYLVKOBKPNQZTK-UHFFFAOYSA-N
Formula:	C7H6
SMILES:	C#CC1(C#C)CC1
Mol. weight [g/mol]:	90.12
CAS:	72323-66-1

## Physical Properties

Property code	Value	Unit	Source
chl	-4113.00 ± 0.80	kJ/mol	NIST Webbook
gf	509.46	kJ/mol	Joback Method
hf	538.50 ± 1.20	kJ/mol	NIST Webbook
hfl	500.90 ± 1.20	kJ/mol	NIST Webbook
hfus	11.67	kJ/mol	Joback Method
hvap	37.60	kJ/mol	NIST Webbook
hvap	37.60	kJ/mol	NIST Webbook
ie	8.90	eV	NIST Webbook
ie	9.26	eV	NIST Webbook
log10ws	-1.99		Crippen Method
logp	1.033		Crippen Method
mvol	81.430	ml/mol	McGowan Method
pc	4966.33	kPa	Joback Method
tb	373.00 ± 0.30	K	NIST Webbook
tc	563.60	K	Joback Method
tf	304.43	K	Joback Method
vc	0.306	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	132.83	J/mol×K	346.78	Joback Method
cpg	144.48	J/mol×K	382.92	Joback Method
cpg	154.76	J/mol×K	419.05	Joback Method
cpg	163.83	J/mol×K	455.19	Joback Method

cpg	171.83	J/mol×K	491.33	Joback Method
cpg	178.90	J/mol×K	527.46	Joback Method
cpg	185.19	J/mol×K	563.60	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C72323661&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C72323661&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.cheméo.com/doc/models/crippen_log10ws">https://www.cheméo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>chl:</b>	Standard liquid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfl:</b>	Liquid phase enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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