

# 4-Methyl-3-nitrobenzoyl chloride

<b>Other names:</b>	3-Nitro-p-toluyl chloride Benzoyl chloride, 4-methyl-3-nitro-
<b>Inchi:</b>	InChI=1S/C8H6ClNO3/c1-5-2-3-6(8(9)11)4-7(5)10(12)13/h2-4H,1H3
<b>InchiKey:</b>	DXMHBBURYDVYAI-UHFFFAOYSA-N
<b>Formula:</b>	C8H6ClNO3
<b>SMILES:</b>	<chem>Cc1ccc(C(=O)Cl)cc1[N+](=O)[O-]</chem>
<b>Mol. weight [g/mol]:</b>	199.59
<b>CAS:</b>	10397-30-5

## Physical Properties

Property code	Value	Unit	Source
gf	4.33	kJ/mol	Joback Method
hf	-133.94	kJ/mol	Joback Method
hfus	26.90	kJ/mol	Joback Method
hvap	64.72	kJ/mol	Joback Method
log10ws	-3.49		Crippen Method
logp	2.282		Crippen Method
mvol	131.050	ml/mol	McGowan Method
pc	3686.49	kPa	Joback Method
tb	662.22	K	Joback Method
tc	917.82	K	Joback Method
tf	454.84	K	Joback Method
vc	0.512	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	295.22	J/molxK	662.22	Joback Method
cpg	304.78	J/molxK	704.82	Joback Method
cpg	313.53	J/molxK	747.42	Joback Method
cpg	321.52	J/molxK	790.02	Joback Method
cpg	328.78	J/molxK	832.62	Joback Method
cpg	335.35	J/molxK	875.22	Joback Method
cpg	341.26	J/molxK	917.82	Joback Method

# Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	434.00 ± 1.00	K	2.00	NIST Webbook

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C10397305&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C10397305&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/11-331-1/4-Methyl-3-nitrobenzoyl-chloride.pdf>

Generated by Cheméo on 2024-04-27 23:24:30.420611826 +0000 UTC m=+16549519.341189142.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.