

N-Ethylformamide

Other names:	Ethylformamide Formamide, N-ethyl- N-Aethylformamid N-Formylethylamine
Inchi:	InChI=1S/C3H7NO/c1-2-4-3-5/h3H,2H2,1H3,(H,4,5)
InchiKey:	KERBAAIBDHEFDD-UHFFFAOYSA-N
Formula:	C3H7NO
SMILES:	CCNC=O
Mol. weight [g/mol]:	73.09
CAS:	627-45-2

Physical Properties

Property code	Value	Unit	Source
gf	-35.75	kJ/mol	Joback Method
hf	-137.36	kJ/mol	Joback Method
hfus	10.91	kJ/mol	Joback Method
hvap	58.40	kJ/mol	NIST Webbook
hvap	58.44	kJ/mol	NIST Webbook
log10ws	-0.04		Crippen Method
logp	-0.248		Crippen Method
mcvol	64.680	ml/mol	McGowan Method
pc	4966.33	kPa	Joback Method
rinpol	794.00		NIST Webbook
tb	471.20	K	NIST Webbook
tc	546.91	K	Joback Method
tf	218.23	K	Joback Method
vc	0.256	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	147.28	J/molxK	546.91	Joback Method
cpg	142.03	J/molxK	516.90	Joback Method
cpg	136.55	J/molxK	486.90	Joback Method

cpg	130.83	J/mol×K	456.89	Joback Method
cpg	124.86	J/mol×K	426.88	Joback Method
cpg	118.65	J/mol×K	396.88	Joback Method
cpg	112.17	J/mol×K	366.87	Joback Method
rhoI	930.45	kg/m3	318.15	Volumetric Properties of Binary Mixtures of 1-Butyl-1-Methylpyrrolidinium Tris(pentafluoroethyl)trifluorophosphate with N-Methylformamide, N-Ethylformamide, N,N-Dimethylformamide, N,N-Dibutylformamide, and N,N-Dimethylacetamide from (293.15 to 323.15) K
rhoI	939.04	kg/m3	308.15	Volumetric Properties of Binary Mixtures of N-Ethylformamide with Tetrahydropyran, 2-Pentanone, and Propylacetate from (293.15 to 313.15) K
rhoI	934.78	kg/m3	313.15	Volumetric Properties of Binary Mixtures of N-Ethylformamide with Tetrahydropyran, 2-Pentanone, and Propylacetate from (293.15 to 313.15) K
rhoI	951.59	kg/m3	293.15	Volumetric Properties of Binary Mixtures of 1-Butyl-1-Methylpyrrolidinium Tris(pentafluoroethyl)trifluorophosphate with N-Methylformamide, N-Ethylformamide, N,N-Dimethylformamide, N,N-Dibutylformamide, and N,N-Dimethylacetamide from (293.15 to 323.15) K

rhoI	947.48	kg/m3	298.15	Volumetric Properties of Binary Mixtures of 1-Butyl-1-Methylpyrrolidinium Tris(pentafluoroethyl)trifluorophosphate with N-Methylformamide, N-Ethylformamide, N,N-Dimethylformamide, N,N-Dibutylformamide, and N,N-Dimethylacetamide from (293.15 to 323.15) K
rhoI	943.29	kg/m3	303.15	Volumetric Properties of Binary Mixtures of 1-Butyl-1-Methylpyrrolidinium Tris(pentafluoroethyl)trifluorophosphate with N-Methylformamide, N-Ethylformamide, N,N-Dimethylformamide, N,N-Dibutylformamide, and N,N-Dimethylacetamide from (293.15 to 323.15) K
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rhoI	926.06	kg/m3	323.15	Volumetric Properties of Binary Mixtures of 1-Butyl-1-Methylpyrrolidinium Tris(pentafluoroethyl)trifluorophosphate with N-Methylformamide, N-Ethylformamide, N,N-Dimethylformamide, N,N-Dibutylformamide, and N,N-Dimethylacetamide from (293.15 to 323.15) K
rhoI	951.59	kg/m3	293.15	Volumetric Properties of Binary Mixtures of 1-Butyl-3-Methylimidazolium Tris(pentafluoroethyl)trifluorophosphate with N-Methylformamide, N-Ethylformamide, N,N-Dimethylformamide, N,N-Dibutylformamide, and N,N-Dimethylacetamide from (293.15 to 323.15) K
rhoI	947.48	kg/m3	298.15	Volumetric Properties of Binary Mixtures of 1-Butyl-3-Methylimidazolium Tris(pentafluoroethyl)trifluorophosphate with N-Methylformamide, N-Ethylformamide, N,N-Dimethylformamide, N,N-Dibutylformamide, and N,N-Dimethylacetamide from (293.15 to 323.15) K

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Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.51551e+01
Coeff. B	-4.20439e+03
Coeff. C	-7.21300e+01
Temperature range (K), min.	354.92
Temperature range (K), max.	499.25

Sources

Crippen Method: <http://pubs.acs.org/doi/abs/10.1021/ci990307l>

Crippen Method: https://www.chemeo.com/doc/models/crippen_log10ws

Volumetric Properties of Binary Mixtures of N-Butylpyrrolidinium Tris(pentafluoroethyl)trifluorophosphate With N-Methylformamide <https://www.doi.org/10.1021/je400803f>

Volumetric Properties of Binary Mixtures of N-Ethylformamide with Tetrahydrofuran, 1,2-Dichloroethane, and N,N-Dimethylformamide <http://webbook.nist.gov/cgi/cbook.cgi?ID=C627452&Units=SI>

Volumetric Properties of Binary Mixtures of N,N-Dimethylformamide with N,N-Dimethylacetamide from (293.15 to 313.15) K <https://www.doi.org/10.1021/je300974g>

Volumetric Properties of Binary Mixtures of N,N-Dimethylformamide with N,N-Dimethylacetamide from (293.15 to 313.15) K <https://www.doi.org/10.1021/je5002945>

Volumetric Properties of Binary Mixtures of N,N-Dimethylformamide with N,N-Dimethylacetamide from (293.15 to 313.15) K <https://www.doi.org/10.1016/j.jct.2012.02.033>

Volumetric Properties of Binary Mixtures of N,N-Dimethylformamide with N,N-Dimethylacetamide from (293.15 to 313.15) K https://en.wikipedia.org/wiki/Joback_method

Volumetric Properties of Binary Mixtures of N,N-Dimethylformamide with N,N-Dimethylacetamide from (293.15 to 313.15) K <http://link.springer.com/article/10.1007/BF02311772>

McGowan Method:

N,N-Dimethylformamide, N,N-Dibutylformamide, and N,N-Dimethylacetamide from (293.15 to 323.15) K:

Legend

cpg: Ideal gas heat capacity

gf: Standard Gibbs free energy of formation

hf: Enthalpy of formation at standard conditions

hfus: Enthalpy of fusion at standard conditions

hvap: Enthalpy of vaporization at standard conditions

log10ws: Log10 of Water solubility in mol/l

logp: Octanol/Water partition coefficient

mcvol: McGowan's characteristic volume

pc: Critical Pressure

pvap: Vapor pressure

rho: Liquid Density

rinpol: Non-polar retention indices

tb: Normal Boiling Point Temperature

tc: Critical Temperature

tf: Normal melting (fusion) point

vc: Critical Volume

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